

## RESEARCH LETTER

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## Key Points:

- Iron cycle is simulated in a global biogeochemical ocean model
- Mesoscale iron fertilization experiments can be reproduced by the global model
- Iron-light colimitation enhances sensitivity of oceanic CO<sub>2</sub> uptake to dust

## Supporting Information:

- Figure S1
- Figure S2
- Table S1
- Text S1
- Text S2

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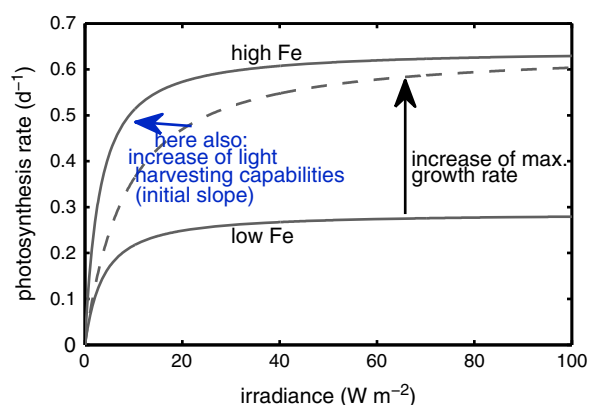
Enhanced sensitivity of oceanic CO<sub>2</sub> uptake to dust deposition by iron-light colimitationLevin Nickelsen<sup>1</sup> and Andreas Oschlies<sup>1</sup><sup>1</sup>GEOMAR, Helmholtz Centre for Ocean Research Kiel, Kiel, Germany

**Abstract** The iron hypothesis suggests that in large areas of the ocean phytoplankton growth and thus photosynthetic CO<sub>2</sub> uptake is limited by the micronutrient iron. Phytoplankton requires iron in particular for nitrate uptake, light harvesting, and electron transport in photosynthesis, suggesting a tight coupling of iron and light limitation. One important source of iron to the open ocean is dust deposition. Previous global biogeochemical modeling studies have suggested a low sensitivity of oceanic CO<sub>2</sub> uptake to changes in dust deposition. Here we show that this sensitivity is increased significantly when iron-light colimitation, i.e., the impact of iron bioavailability on light-harvesting capabilities, is explicitly considered. Accounting for iron-light colimitation increases the shift of export production from tropical and subtropical regions to the higher latitudes of subpolar regions at high dust deposition and amplifies iron limitation at low dust deposition. Our results reemphasize the role of iron as a key limiting nutrient for phytoplankton.

## 1. Introduction

Low concentrations of the micronutrient iron limit primary production in vast areas of the ocean and in particular in most parts of the Southern Ocean [Boyd and Ellwood, 2010]. One of the major sources of iron to the ocean is dust deposition that is suggested to be coupled intimately to climate [Martínez-García et al., 2011]. While today dust deposition to the Southern Ocean is very low, the iron hypothesis [Martin, 1990] states that enhanced dust deposition to the Southern Ocean during the Last Glacial Maximum triggered additional export of organically bound carbon and therefore decreased atmospheric CO<sub>2</sub> concentrations. Although a recent modeling study estimates only an increase of 2 ppmv in preindustrial atmospheric CO<sub>2</sub> when dust deposition is shut off completely [Tagliabue et al., 2014], other studies suggest that decreasing dust deposition in the future such as predicted by Mahowald et al. [2006] may possibly lead to more severe iron limitation and a larger reduction in oceanic CO<sub>2</sub> uptake by phytoplankton [Parekh et al., 2006; Tagliabue et al., 2008]. The role of iron in regulating the oceanic CO<sub>2</sub> uptake is thus important for understanding past and possibly future atmospheric CO<sub>2</sub> levels.

From a biological point of view iron limitation in coupled biogeochemical ocean circulation models has, until now, been treated in a very simplistic way and interactions with other limiting nutrients and factors are often neglected. Observations show that iron limitation of phytoplankton growth is created by the requirement of iron for nitrate uptake, for proteins in the electron transport chain, for synthesizing chlorophyll and photoreaction centers, and the functioning of light-harvesting antennae [Sunda and Huntsman, 1997; Behrenfeld and Milligan, 2013]. Although physiological adaptation of polar phytoplankton species to low-iron concentrations may compensate for some of the positive effect of iron on light-harvesting capabilities [Strzepek et al., 2011, 2012], incubation experiments show elevated light-harvesting capabilities of phytoplankton after adding iron [Feng et al., 2010]. However, in most biogeochemical models that have been used to investigate the sensitivity of ocean biogeochemistry and CO<sub>2</sub> uptake to dust deposition, iron limitation is included as a further Monod term in a minimum function [Bopp et al., 2003; Moore and Braucher, 2008; Parekh et al., 2008; Tagliabue et al., 2009a], while in explicit quota models such as in Tagliabue et al. [2009a, 2014] iron uptake is allowed to continue also under light-limiting conditions. Only the recent model of Galbraith et al. [2010] explicitly describes the impact of iron limitation on the chlorophyll-to-carbon ratio and the initial slope of how irradiance is processed into photosynthesis as observed in culture and field experiments [Greene et al., 1991; Davey and Geider, 2001; Hopkinson et al., 2007; Moore et al., 2007; Hopkinson and Barbeau, 2008]. The way the influence of iron limitation on light limitation is implemented in this model leads to parallel changes in the light-limited slope and light-saturated rate of photosynthesis with iron concentrations (Figure 1). The increase of the maximum growth rate only, as illustrated in Figure 1,



**Figure 1.** Impact of iron on the photosynthesis-irradiance (P-I) curve. The lower solid line represents the P-I curve for low-iron concentrations; the upper solid line represents the P-I curve for high-iron concentrations in the model we use here. The dashed line represents the P-I curve if a higher-iron availability only increases the maximum growth rate and not the light-harvesting capabilities.

is the response to additional iron as it is often treated in the other models. Here, in addition, also the initial slope increases. This response of both, maximum growth rate and initial slope, to the addition of iron is also observed in culture experiments [Behrenfeld *et al.*, 2004; Behrenfeld and Milligan, 2013].

The model used here has been shown to perform well in simulating the observed present-day global surface iron and phosphate concentrations, while the agreement to observations decreases if iron limitation of light-harvesting capabilities is not considered [Galbraith *et al.*, 2010]. However, how iron limitation of light-harvesting capabilities influences the response of oceanic CO<sub>2</sub> uptake to changes in dust deposition has yet to be answered.

## 2. Methods

The model we use is a coupled global ocean-biogeochemistry model with a detailed iron cycle [Galbraith *et al.*, 2010]. In brief, the biogeochemical model consists of four prognostic tracers, namely, phosphate (PO<sub>4</sub>), dissolved organic phosphorus, dissolved iron (Fe), and oxygen (O<sub>2</sub>). Phytoplankton biomass is modeled as a prognostic variable that is not transported. Export production, grazing, and community structure formulations are based on empirical formulations by Dunne *et al.* [2005]. External sources of iron to the ocean are dust deposition and sediment release. The complexation of iron with organic ligands is implicitly calculated as in Parekh *et al.* [2006]. A complete description of the biogeochemical model based on the model code made available by Galbraith *et al.* [2010] is included in the supporting information.

The difference of our approach in comparison to prior approaches is illustrated in the photosynthesis-irradiance (P-I) curve in Figure 1. If increased iron concentrations only increase the maximum photosynthesis, the effect is most pronounced at high-light levels. If the impact of iron on light-harvesting capabilities is considered as well, also the slope of the response of photosynthesis to irradiance increases. This has a particularly strong effect at low-light levels.

The physical ocean model configuration used in this study is described by Galbraith *et al.* [2010, 2011]. The model is the coupled ocean-sea ice model component of the climate model version 2 with the Modular Ocean Model version 4p1 at coarse resolution. It has a nominal resolution of 3° in longitudinal direction and 3° in latitudinal direction with a higher resolution up to 2/3° near the equator and at the latitudes of the Drake Passage and the equivalent latitudes on the Northern Hemisphere. The vertical resolution of the model consists of 28 levels with pressure as the vertical coordinate and a free sea surface. The vertical resolution varies from 10 m at the surface to 506 m in the lowest layer. The ocean surface is forced using a repeated climatological year from the Coordinated Ocean Reference Experiment [Griffies *et al.*, 2009]. Surface salinities are restored to observations with a time constant of 10 days over the top layer.

We ran the model in a coupled ice-ocean mode with fixed atmospheric forcing and prescribed atmospheric 278 μatm CO<sub>2</sub> for 2500 years as a spin-up run. To simulate aeolian deposition of iron, a repeated climatological seasonal cycle of dust deposition [Ginoux *et al.*, 2004] is used. Dust deposition is converted to iron deposition assuming a fraction of iron in dust of 3.5% in clay fractions and 1.2% in silt fractions and an iron solubility of 2% following Galbraith *et al.* [2010]. Burial of organic matter or CaCO<sub>3</sub> is not allowed in any of our simulations. After the spin-up, we applied a dynamic and homogenous atmospheric CO<sub>2</sub> reservoir with an initial value of 590 pg C [Sarmiento and Gruber, 2002] corresponding to 278 μatm. The atmospheric CO<sub>2</sub> reservoir is in exchange with the ocean but does not affect temperature. The experiments were started after 200 additional years of spin-up with a free atmospheric CO<sub>2</sub> concentration. During these 200 years the

change of average surface phosphate concentrations was  $-6.1 \times 10^{-4}$  mmol  $\text{PO}_4 \text{ m}^{-3}$ , and the atmospheric  $\text{CO}_2$  concentration decreased from 278  $\mu\text{atm}$  to 277.81 and to 277.63  $\mu\text{atm}$  during the following 1000 years of the control simulation. The decrease is likely due to the small but continuous accumulation of iron from the sediments [Galbraith *et al.*, 2010].

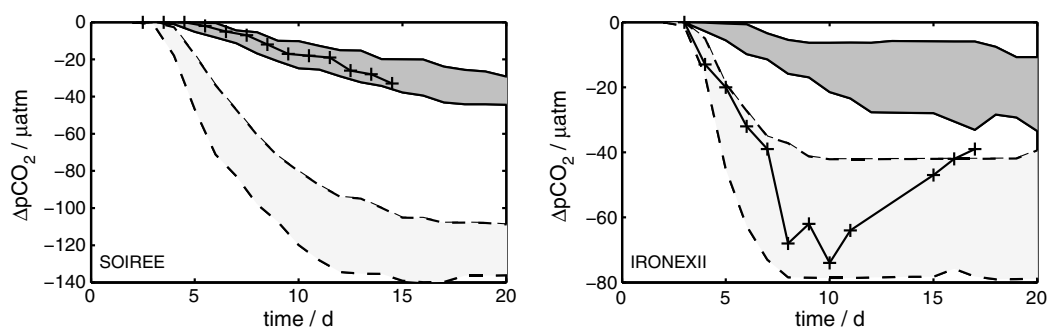
Before starting our model sensitivity experiments, the model was tuned to reproduce observed responses to the two iron fertilization experiments, Southern Ocean Iron RElease Experiment (SOIREE) [Boyd *et al.*, 2000] and the second iron enrichment experiment (IRONEXII) [Coale *et al.*, 1996], in the same way as Aumont and Bopp [2006]. The experiment SOIREE was conducted in the Southern Ocean while IRONEXII in the equatorial Pacific. To simulate the mesoscale iron fertilization experiments, the iron concentration was set to 2 nM in the whole mixed layer every  $5^\circ$  in latitudinal and  $9^\circ$  in longitudinal direction and held constant for 30 days. For SOIREE the ocean was fertilized only south of  $40^\circ\text{S}$  starting with February in our model and for IRONEXII between  $5^\circ\text{S}$  and between  $5^\circ\text{N}$  and  $140^\circ\text{E}$  and  $120^\circ\text{W}$  starting in May. To calculate  $\Delta p\text{CO}_2$ , the simulation was repeated without iron fertilization, and the difference in  $p\text{CO}_2$  was calculated from these two simulations. Following Aumont and Bopp [2006], the response to fertilization was determined from sites that were within  $\pm 10$  m difference in the mixed layer depth and  $\pm 2^\circ\text{C}$  to the respective fertilized location in SOIREE or IRONEXII. The responses in  $p\text{CO}_2$  of these fertilization sites give a range of responses that are then compared to the observed values. The model parameters were optimized to reduce the difference between observed and simulated  $\Delta p\text{CO}_2$ . In the resulting parameter set the stability constant of iron-ligand complexes increases from  $\text{KFeL}_{\text{min}} = 8 \times 10^9 \text{ M}^{-1}$ ,  $\text{KFeL}_{\text{max}} = 8 \times 10^{10} \text{ M}^{-1}$  to  $\text{KFeL}_{\text{min}} = 1 \times 10^{11} \text{ M}^{-1}$ ,  $\text{KFeL}_{\text{max}} = 5 \times 10^{11} \text{ M}^{-1}$ . These values are more in line with a recent compilation by Gledhill and Buck [2012] of  $\text{KFeL}$  being in the range of  $10^{11}$  to  $10^{12} \text{ M}^{-1}$ . The half-saturation constant of iron to phosphate uptake ( $k_{\text{Fe:p}}$ ) is reduced from 0.8 mmol Fe (mol  $\text{PO}_4$ ) $^{-1}$  to 0.4 mmol Fe (mol  $\text{PO}_4$ ) $^{-1}$ , and the mortality rate ( $\lambda_0$ ) is increased from 0.19  $\text{day}^{-1}$  to 0.38  $\text{day}^{-1}$  to better reproduce the observations.

We perform four sensitivity experiments to test the importance of iron-light colimitation at different iron concentrations: (i) Abrupt increase of dust deposition to a deposition as estimated for the Last Glacial Maximum [Mahowald *et al.*, 2006] hereafter abbreviated as LGM-ILL. (ii) Equal to (i) but without the impact of iron on light-harvesting capabilities (LGM-NOILL). (iii) Abrupt decrease of dust deposition to a deposition as estimated for a climate with double  $\text{CO}_2$  concentrations relative to today [Mahowald *et al.*, 2006] hereafter abbreviated as  $2x\text{CO}_2$ -ILL. (iv) As in (iii) but without the impact of iron on light-harvesting capabilities ( $2x\text{CO}_2$ -NOILL). In addition to the sensitivity experiments, the spin-up run is continued with a prognostic atmospheric  $\text{CO}_2$  reservoir as a control simulation (CTL) to compare the experiments. All dust deposition fields are shown in the supporting information Figure S1.

The dust deposition used in the control run and the preindustrial estimate by Mahowald *et al.* [2006] differ. To make the experiments independent of the control dust deposition, the dust deposition estimates in the experiments are created by multiplying the dust deposition in the control run with the ratio of the LGM or  $2x\text{CO}_2$  dust deposition estimates by Mahowald *et al.* [2006] to the preindustrial estimate by Mahowald *et al.* [2006]. Additional experiments were performed without scaling the change in dust deposition to the preindustrial estimate, thus using the absolute dust deposition fields as simulated by Mahowald *et al.* [2006] (supporting information Figure S2). In these additional runs more  $\text{CO}_2$  is taken up using the LGM dust and less using the  $2x\text{CO}_2$  dust. The impact of iron limitation of light-harvesting capabilities, however, is as strong as in the simulations shown here.

Note that we are not trying to realistically simulate past conditions of the Last Glacial Maximum or predictions into the future. Atmospheric forcing, temperature and circulation remain at preindustrial conditions in all our sensitivity experiments, and we concentrate our analysis of a more mechanistic parameterization of iron limitation on the isolated impact of changes in aeolian iron supply. Also, fraction and solubility of iron in dust in all experiments are kept as in the spin-up run for reasons of comparability.

To turn off the effect of iron on light-harvesting capabilities in experiments (ii) and (iv), the variables describing the light-harvesting capabilities, the initial slope in the P-I curve ( $\alpha^{\text{chl}}$ ), and the chlorophyll-to-carbon ratio ( $\theta_{\text{max}}^{\text{Fe}}$ ) (also see the model description in the supporting information), are kept at the annual mean values they have at the end of the spin-up run at each point in space. There is thus no seasonal cycle of these variables in the experiments with no iron limitation of light-harvesting capabilities, whereas in the control experiment  $\alpha^{\text{chl}}$  and  $\theta_{\text{max}}^{\text{Fe}}$  vary seasonally. In a comparison between the control experiment to an additional control experiment (not shown here) in which we fixed  $\alpha^{\text{chl}}$  and  $\theta_{\text{max}}^{\text{Fe}}$  to the annual mean of the last year



**Figure 2.** Comparison between observed and simulated response in  $p\text{CO}_2$  to mesoscale iron fertilization as in the experiments (left) SOIREE [Boyd *et al.*, 2000] and (right) IRONEXII [Coale *et al.*, 1996]. The crosses are observed differences between  $p\text{CO}_2$  inside and outside the fertilization area as read by eye from [Aumont and Bopp, 2006]. The light shaded area indicates the simulated response with parameters as in Galbraith *et al.* [2010] and the dark shaded area with the new parameter values. Note the different scales.

of the spin-up, the differences are very small and, in terms of the atmospheric carbon reservoir, amount to  $0.6 \mu\text{atm}$ .

### 3. Results and Discussion

#### 3.1. Tuning the Model

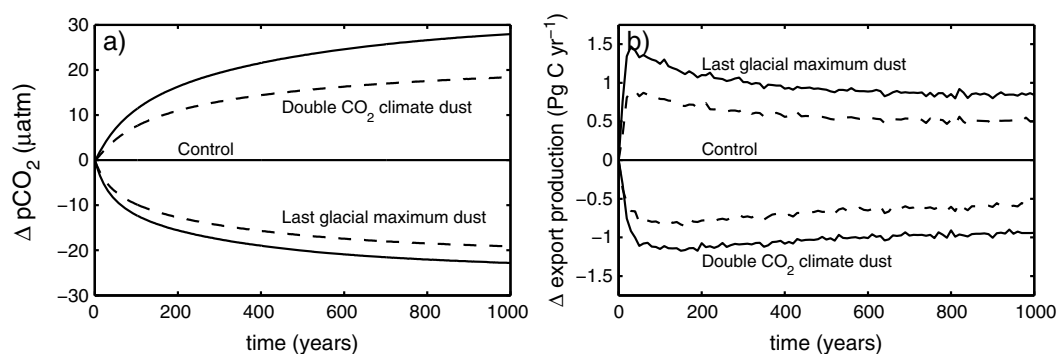
In order to validate the response of the model to changes in iron concentrations, we tune the model to be able to reproduce observed responses to mesoscale iron fertilization experiments in the same way as Aumont and Bopp [2006] (Figure 2a). With the original parameter set of Galbraith *et al.* [2010]  $p\text{CO}_2$  is much more reduced than observed in SOIREE, while with the tuned parameter set the observed values are perfectly within the simulated range of  $\Delta p\text{CO}_2$ . On the other hand, the difference between new and old parameters is not that pronounced in the fertilization experiment IRONEXII. With the new parameter set  $\Delta p\text{CO}_2$  is underestimated although at the end of the experiment the observed value lies in between of the simulated ranges of original and new parameter set.

The root-mean-square errors (RMSE) for the simulated phosphate, oxygen, and iron concentrations compared to observations from the World Ocean Atlas and Tagliabue *et al.* [2012] are  $0.30 \text{ mmol PO}_4 \text{ m}^{-3}$ ,  $34.79 \text{ mmol O}_2 \text{ m}^{-3}$ , and  $1.0 \text{ nM dFe}$ , respectively, using the original parameter set by Galbraith *et al.* [2010]. With the tuned parameter set the RMSE for phosphate does not change, the RMSE for oxygen concentrations increases slightly to  $36.73 \text{ mmol O}_2 \text{ m}^{-3}$  and the RMSE for iron concentrations reduces strongly to  $0.89 \text{ nM dFe}$  for the full ocean and from  $0.27 \text{ nM dFe}$  to  $0.26 \text{ nM dFe}$  at the surface. It is encouraging that the model we use is able to reproduce the observed response to iron fertilization during SOIREE in the Southern Ocean and that the agreement to observed iron concentrations is improved with the tuned parameter set.

#### 3.2. Oceanic $\text{CO}_2$ Uptake

The LGM dust deposition leads to a total decrease of atmospheric  $\text{CO}_2$  by  $22.8 \mu\text{atm}$  (Figure 3a) in our model simulations. This decrease is about 19% larger (or  $3.7 \mu\text{atm}$ ) than that of simulation LGM-NOILL, which does not account for iron-light colimitation. Recent estimates of the  $\text{CO}_2$  uptake of the ocean by increasing the dust to LGM conditions have all been smaller than in our idealized model results. The oceanic drawdown of  $\text{CO}_2$  in simulations with dust of the Last Glacial Maximum from the literature are  $11 \mu\text{atm}$  [Bopp *et al.*, 2003],  $10 \mu\text{atm}$  [Parekh *et al.*, 2008],  $16 \mu\text{atm}$  [Tagliabue *et al.*, 2009b], and  $25 \mu\text{atm}$  [Oka *et al.*, 2011] and are thus a small part of the full glacial decrease in atmospheric  $\text{CO}_2$  of  $\sim 50 \mu\text{atm}$  prior to carbonate compensation, i.e., the burial of carbon as  $\text{CaCO}_3$  in ocean sediments [Brovkin *et al.*, 2007; Tagliabue *et al.*, 2009b]. Particularly, the interactive limitation of the phytoplankton in the Southern Ocean by iron and light could produce a strong impact of dust deposition in our experiment LGM-ILL. Our new simulations suggest that dust deposition can have a larger impact on the biological carbon pump than suggested by recent studies and thus could be a major factor contributing to the reduction of atmospheric  $\text{CO}_2$  concentration during glacial times.

The difference of our simulated sensitivity of atmospheric  $\text{CO}_2$  already in the NOILL simulations in comparison to other studies stems from differences in the residence time of dissolved iron at the surface.

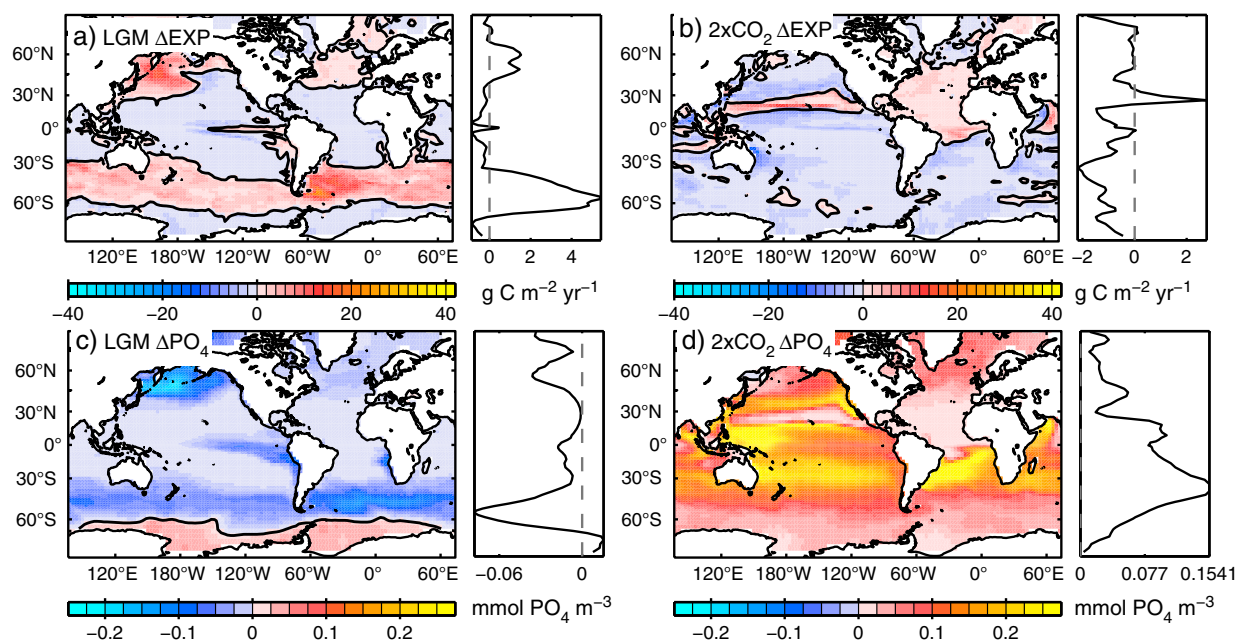


**Figure 3.** Change of (a) atmospheric CO<sub>2</sub> concentration for the 2xCO<sub>2</sub> dust deposition (upper lines) and for the LGM dust deposition (lower lines) and (b) globally integrated export production (at 100 dbar) relative to the control simulation for the LGM dust deposition (upper lines) and for the 2xCO<sub>2</sub> dust deposition (lower lines). Dashed lines are runs without the dependence of light-harvesting capabilities on iron, solid lines, are runs with considering the effect of iron on light-harvesting capabilities.

In the model we use, the equilibrium constant between free iron, ligands, and their complexation ( $K_{FeL} = 1 \times 10^{11}$  to  $5 \times 10^{11} \text{ M}^{-1}$ ) is lower than in other models such as in [Tagliabue *et al.*, 2009b] ( $K_{FeL} = 10^{12} \text{ M}^{-1}$ ). In addition, in the model we use, photodissociation of iron-ligand complexes reduces the equilibrium constant to the lower end of  $K_{FeL} = 1 \times 10^{11}$  to  $5 \times 10^{11} \text{ M}^{-1}$  at the surface. The low-equilibrium constant at the surface leads to fast iron scavenging and a short residence time of dissolved iron. The dissolved iron concentrations rely much more on external sources because of the low-background concentrations. A further factor reducing the background concentration is the neglect of a hydrothermal source of dissolved iron in our model configuration—although the link of this iron source to biological productivity in the surface ocean has been argued to be negligible [Tagliabue *et al.*, 2014]. The response of the biological pump to changes in iron supply is hence much stronger than with a long residence time of dissolved iron at the surface. For a better estimate of how the oceanic CO<sub>2</sub> uptake changes with a varying degrees of iron limitation of phytoplankton, the residence time of iron in surface water needs to be better constrained in observational studies.

For predictions of future atmospheric CO<sub>2</sub> concentrations an estimation of the susceptibility of the ocean biogeochemistry to possible decreases in dust deposition in a warmer and wetter climate is necessary [Mahowald *et al.*, 2006, 2010]. Accounting for the iron limitation of light-harvesting capabilities at low dust deposition leads to an extra increase of atmospheric CO<sub>2</sub> by 9.6 μatm in experiment 2xCO<sub>2</sub>-ILL compared to experiment 2xCO<sub>2</sub>-NOILL. This makes up for 32% of the total response of 28.0 μatm and is around twice the CO<sub>2</sub> increase estimated by previous studies. In the modeling study by Tagliabue *et al.* [2014] shutting the dust deposition off completely leads to a slight increase of the atmospheric CO<sub>2</sub> concentration by 2 ppmv. Another modeling study with a different model simulated an increase of 14 μatm by reducing current dust deposition by half [Parekh *et al.*, 2006]. Based on observations of interactions between iron and light limitation in incubation experiments and culture studies [Greene *et al.*, 1991; Davey and Geider, 2001; Hopkinson *et al.*, 2007; Moore *et al.*, 2007; Hopkinson and Barbeau, 2008] our global model results show that a decrease in dust deposition could lead to a larger decrease in future oceanic CO<sub>2</sub> uptake than estimated previously.

The globally integrated export production shows a strong response to the changes in dust deposition particularly during the first 100 years of the experiments (Figure 3b). The fluctuations on shorter timescales stem from fluctuations in sea ice coverage and are mediated to export production by affecting the irradiance reaching the ocean surface. In the case of the LGM dust, excess macronutrients are taken up, and in the case of the 2xCO<sub>2</sub> dust, excess iron is taken up during the first 100 years until in the end global export production equilibrates at a higher (+0.86 pg C yr<sup>-1</sup> at 100 dbar) or lower level (−0.94 pg C yr<sup>-1</sup> at 100 dbar) relative to the control simulation, respectively. At the end of the simulations the difference in the response of export production between applying and not applying the iron limitation of light-harvesting capabilities of phytoplankton is 0.36 pg C yr<sup>-1</sup> at 100 dbar in the case of LGM dust and 0.35 pg C yr<sup>-1</sup> at 100 dbar in the case of 2xCO<sub>2</sub> dust and thus very pronounced on the globally integrated scale (compare also supporting information Table S1). The iron limitation of light-harvesting capabilities has thus a strong control on the



**Figure 4.** Difference between the simulations with and without considering the effect of iron on (a and b) light-harvesting in export production at 100 dbar ( $\text{g C m}^{-2} \text{ yr}^{-1}$ ) and (c and d) in surface phosphate concentrations ( $\text{mmol m}^{-3}$ ). Figures 4a and 4c show the results using the LGM dust and Figures 4b and 4d the results using the 2xCO<sub>2</sub> dust. The zonal mean is displayed right to each map.

sensitivity of simulated global export production and atmospheric CO<sub>2</sub> concentrations to the supply of iron to the surface ocean.

The regional difference between the experiments with and without consideration of the impact of iron on light-harvesting capabilities (LGM-ILL minus LGM-NOILL) reveals that in comparison to the LGM-NOILL experiment, export production is particularly increased in the North Pacific, the North Atlantic, and the Southern Ocean (Figure 4). Accordingly, surface phosphate concentrations are reduced in these regions. The reason for that is that due to the consideration of iron-light colimitation, growth rates are increased the most at low-light (not saturated) levels which leads to the strongest response to iron addition in areas with light limitation (Figure 1). In contrast, in the 2xCO<sub>2</sub>-ILL experiment the effect of iron limitation is enhanced so that carbon export is generally reduced, particularly in the northern subtropical Pacific for which a large decline in dust deposition is predicted under global warming (Figure 4). With export production being reduced under 2xCO<sub>2</sub>, more macronutrients are left unutilized in these regions and can be transported into the more oligotrophic subtropical gyres, where export production can thus increase in the 2xCO<sub>2</sub> scenario.

#### 4. Conclusions

Iron-light colimitation is, in contrast to colimitation of, for example, nitrogen and phosphorus, biochemically dependent in that iron is needed for light-harvesting antennae and enzymes in the electron transport [Saito *et al.*, 2008]. We show that our model has a higher sensitivity to changes in dust deposition than earlier models and that the direct effect of iron concentrations on light-harvesting capabilities of phytoplankton further enhances the model sensitivity to changes in dust deposition. Decreasing dust deposition could decrease oceanic CO<sub>2</sub> uptake, by a larger amount than suggested previously. Furthermore, we show that the CO<sub>2</sub> uptake triggered by LGM dust is up to twice as large in our simulations than estimated before. We suggest that the consideration of the effect of iron on light harvesting has a strong impact on the response of the ocean biogeochemistry to dust deposition. The influence of iron on light harvesting increases the response of atmospheric CO<sub>2</sub> to dust deposition by 19% of the total response for the LGM dust deposition and 32% for the 2xCO<sub>2</sub> dust deposition. Due to the importance of this mechanism, more observational and experimental constraints on iron limitation and colimitation with other nutrients and factors are needed for accurate reconstructions of the past climate and prediction of the future. Small details of nutrient limitation of phytoplankton could have large effects of the oceanic response to changes in dust deposition.

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