



1 **Review Article: How does glacier discharge affect marine biogeochemistry and primary**  
2 **production in the Arctic?**

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23 **Abstract**

24 Freshwater discharge from glaciers is increasing across the Arctic in response to  
25 anthropogenic climate change, which raises questions about the potential downstream effects  
26 in the marine environment. Whilst a combination of long-term monitoring programmes and  
27 intensive Arctic field campaigns have improved our knowledge of glacier-ocean interactions  
28 in recent years, especially with respect to fjord/ocean circulation in the marine environment,  
29 there are extensive knowledge gaps concerning how glaciers affect marine biogeochemistry  
30 and productivity. Following two cross-cutting disciplinary International Arctic Science  
31 Committee (IASC) workshops addressing ‘The importance of glaciers for the marine  
32 ecosystem’, here we review the state of the art concerning how freshwater discharge affects  
33 the marine environment with a specific focus on marine biogeochemistry and biological  
34 productivity. Using a series of Arctic case studies (Nuup Kangerlua/Godthåbsfjord,  
35 Kongsfjorden, Bowdoin Fjord, Young Sound, and Sermilik Fjord), the interconnected effects  
36 of freshwater discharge on fjord-shelf exchange, nutrient availability, the carbonate system,  
37 and the microbial foodweb are investigated. Key findings are that whether the effect of  
38 glacier discharge on marine primary production is positive, or negative is highly dependent



39 on a combination of factors. These include glacier type (marine- or land-terminating) and the  
40 limiting resource for phytoplankton growth in a specific spatiotemporal region (light,  
41 macronutrients or micronutrients). Glacier fjords therefore often exhibit distinct discharge-  
42 productivity relationships and multiple case-studies must be considered in order to  
43 understand the net effects of glacier discharge on Arctic marine ecosystems.

#### 44 **1.0 Introduction**

45 Annual freshwater discharge volume from glaciers has increased globally in recent decades  
46 (Rignot et al., 2013; Bamber et al., 2018) and will continue to do so across most Arctic  
47 regions until at least the middle of this century under a Representative Concentration  
48 Pathway (RCP) 4.5 climate scenario (Bliss et al., 2014). This increase in discharge (surface  
49 runoff and subsurface discharge) raises questions about the downstream effects in marine  
50 ecosystems, particularly with respect to ecosystem services such as carbon sequestration and  
51 fisheries (Meire et al., 2015, 2017; Milner et al., 2017). In order to understand the effect of  
52 glaciers on the present-day marine environment, and under future climate scenarios,  
53 knowledge of the physical and chemical perturbations occurring in the water column as a  
54 result of glacier discharge and the structure, function, and resilience of ecosystems within  
55 these regions must be synthesized.

56 Quantifying the magnitude of environmental perturbations from glacial discharge is  
57 complicated by the multiple concurrent, and occasionally counter-acting, effects that glacial  
58 discharge has in the marine environment. For example, ice-rock abrasion means that glacially  
59 fed rivers can carry higher sediment loads than temperate rivers (Chu et al., 2009; Overeem et  
60 al., 2017). Extensive sediment plumes where glacier discharge first enters the ocean limit  
61 light penetration into the water column ( Murray et al., 2015; Halbach et al., 2019) and  
62 ingestion of glacial flour particles can be hazardous, or even fatal, to zooplankton, krill, and  
63 benthic fauna (White and Dagg, 1989; Wlodarska-Kowalczyk and Pearson, 2004; Arendt et  
64 al., 2011; Fuentes et al., 2016). However, these plumes also provide elevated concentrations  
65 of inorganic ions such as calcium carbonate, which affects seawater alkalinity (Yde et al.,  
66 2014; Fransson et al., 2015), and dissolved silicic acid (hereafter 'Si') (Brown et al., 2010;  
67 Meire et al., 2016a) and iron (Fe) (Statham et al., 2008; Lippiatt et al., 2010) which can  
68 potentially increase marine primary production (Gerringa et al., 2012; Meire et al., 2016a).



69

70 Figure 1. Locations of 5 key Arctic field-sites, where extensive work bridging the glacier and  
71 marine domains has been conducted, discussed herein in order to advance understanding of  
72 glacier-ocean interactions. Clockwise from top right: Kongsfjorden (Svalbard), Young Sound  
73 (E Greenland), Sermilik (SE Greenland), Nuup Kangerlua (SW Greenland), Bowdoin (NW  
74 Greenland).

75 The impacts of glacier discharge can also depend upon the spatial and temporal scales  
76 investigated (van de Poll et al., 2018). In semi-enclosed coastal regions and fjord systems,  
77 summertime discharge produces a strong, near-surface stratification. This results in a shallow,  
78 nutrient-poor layer which reduces primary production and drives phytoplankton biomass  
79 deeper in the water column (Rysgaard et al., 1999; Juul-Pedersen et al., 2015; Meire et al.,  
80 2017). On broader scales across continental shelves, freshening can similarly reduce vertical  
81 nutrient supply throughout summer (Coupel et al., 2015), but may also impede the breakdown  
82 of stratification in autumn extending the phytoplankton growing season (Oliver et al., 2018).  
83 Key research questions are how, and on what spatial and temporal timescales, these different  
84 effects interact to enhance, or reduce, marine primary production. Using a synthesis of field  
85 studies from glacier catchments with different characteristics (Fig. 1), we provide answers to  
86 three questions arising from two interdisciplinary workshops on ‘The importance of Arctic  
87 glaciers for the marine ecosystem’ under the umbrella of the International Arctic Science  
88 Committee (IASC).

89 (1) Where and when does glacial freshwater discharge promote or reduce marine primary  
90 production?

91 (2) How does spatiotemporal variability in glacial discharge affect marine primary  
92 production?

93 (3) How far reaching are the effects of glacial discharge on marine biogeochemistry?

94

95



96 **2.0 Fjords as critical zones for glacier-ocean interactions**

97 In the Arctic and sub-Antarctic, most glacial discharge enters the ocean through fjord systems  
98 (Iriarte et al., 2014; Straneo and Cenedese, 2015). The strong lateral gradients and seasonal  
99 changes in environmental conditions associated with glacial discharge in these coastal  
100 environments differentiate their ecosystems from offshore systems (Arendt et al., 2013;  
101 Lydersen et al., 2014; Krawczyk et al., 2018). Fjords can be efficient organic carbon (Smith  
102 et al., 2015) and CO<sub>2</sub> sinks (Rysgaard et al., 2012; Fransson et al., 2015), sustain locally-  
103 important fisheries (Meire et al., 2017), and are critical zones for deep mixing which dictate  
104 how glacially-modified waters are exchanged with the coastal ocean (Mortensen et al., 2014;  
105 Straneo and Cenedese, 2015; Beaird et al., 2018). Fjord-scale processes therefore comprise an  
106 integral part of all questions concerning how glacial discharge affects Arctic coastal primary  
107 production (Arimitsu et al., 2012; Renner et al., 2012; Meire et al., 2017).

108 Fjords act as highly-stratified estuaries, and provide a pathway for the exchange of heat, salt,  
109 and nutrients between near-glacier waters and adjacent coastal regions (Mortensen et al.,  
110 2014, 2018; Straneo and Cenedese, 2015). In deep fjords, such as those around much of  
111 Greenland, warm, saline water is typically found at depth (>200 m), overlaid by cold, fresher  
112 water and, during summer, a thin layer (~50 m or less) of relatively warm near-surface water  
113 (Straneo et al., 2012). The injection of freshwater into fjords from subglacial discharge (Xu et  
114 al., 2012; Carroll et al., 2015), and terminus (Slater et al., 2018) and iceberg melt (Moon et  
115 al., 2018) can drive substantial buoyancy-driven flows in the fjord (Carroll et al., 2015, 2017;  
116 Jackson et al., 2017), which amplify exchange with the shelf system as well as submarine  
117 melting and the calving rates of glacier termini. To date, such modifications to circulation and  
118 exchange between glacier fjords and shelf waters have primarily been studied in terms of  
119 their effects on ocean physics and melting at glacier termini, yet they also have profound  
120 impacts on marine productivity (Meire et al., 2016a; Kanna et al., 2018; Torsvik et al., 2019).



121 While renewal of fjord waters from buoyancy-driven processes is mainly thought to occur  
122 over seasonal to sub-annual timescales (Gladish et al., 2014; Mortensen et al., 2014; Carroll  
123 et al., 2017), energetic shelf forcing (i.e., from coastal/katabatic winds and coastally-trapped  
124 waves) can result in rapid exchange over synoptic timescales (Straneo et al., 2010; Jackson et  
125 al., 2014; Moffat, 2014) and similarly also affect productivity (Meire et al., 2016b).  
126 Additionally, topographic features such as sills and lateral constrictions can exert a strong  
127 control on fjord-shelf exchange (Gladish et al., 2014; Carroll et al., 2017, 2018). Ultimately,  
128 circulation can vary considerably depending on fjord geometry and the relative contributions  
129 from buoyancy, wind, and shelf forcing (Straneo and Cenedese, 2015). Some variability in

#### Nuup Kangerlua / Godthåbsfjord (SW Greenland) 64° N 051° W

Nuup Kangerlua (also known as Godthåbsfjord) is a large glacier-fjord system (~190 km long, 4–8 km wide and up to 625 m deep). The fjord hosts 6 different glaciers (3 land-terminating and 3 marine-terminating), including the marine-terminating glaciers Kangiata Nunaata Sermia and Narsap Sermia. The shallowest sill within the fjord is at ~170 m depth (Mortensen et al., 2011). Nuup Kangerlua is one of few well-studied Greenland fjord systems, due to extensive work conducted by the Greenland Institute of Natural Resources. A data portal is available containing monthly fjord data through the Greenland Ecosystem Monitoring Programme (GEM; <http://g-e-m.dk>).

130 the spatial patterns of primary production are therefore expected between glacier-fjord  
131 systems as differences in geometry and forcing affect exchange with the shelf and water  
132 column structure. These changes affect the availability of the resources (light and nutrients)  
133 which constrain local primary production (Meire et al., 2016b, 2017). This is verified by field  
134 observations around the Arctic which demonstrate that glacier fjords range considerably in  
135 productivity from very low (<40 mg C m<sup>-2</sup> day<sup>-1</sup>), to moderately productive systems (>500  
136 mg C m<sup>-2</sup> day<sup>-1</sup>) (Jensen et al., 1999; Rysgaard et al., 1999; Hop et al., 2002; Meire et al.,  
137 2017). For comparison, the pan-Arctic basin exhibits a mean production of 420 ± 26 mg C  
138 m<sup>-2</sup> day<sup>-1</sup> (mean March-September 1998-2006) (Pabi et al., 2008).

139 Fjord-shelf processes also contribute to the exchange of active cells and microbial species'  
140 resting stages, thus preconditioning primary production prior to the onset of the growth  
141 season (Krawczyk et al., 2015, 2018). Protists (unicellular eukaryotes) are the main marine  
142 primary producers in the Arctic. This highly-specialized and diverse group includes species  
143 that are ice-associated (sympagic) and/or pelagic. Many protists in fjords and coastal areas of  
144 the Arctic maintain diverse “seed banks” of resting stages, which promotes the resilience and  
145 adaptability of species on timescales from seasons to decades (Ellegaard and Ribeiro, 2018).  
146 Yet seawater inflow into fjords can still change the dominant species within a single season.  
147 In Godthåbsfjord, the spring phytoplankton bloom is typically dominated by *Fragilarisopsis*  
148 spp. diatoms and *Phaeocystis* spp. haptophytes. Yet unusually prolonged coastal seawater  
149 inflow in spring 2009 led to mass occurrence of chain forming *Thalassiosira* spp. diatoms  
150 and the complete absence of the normally abundant *Phaeocystis* spp. (Krawczyk et al., 2015)  
151 -a pattern which has been found elsewhere in the Arctic, including Kongsfjorden (Hegseth  
152 and Tverberg, 2013).

153

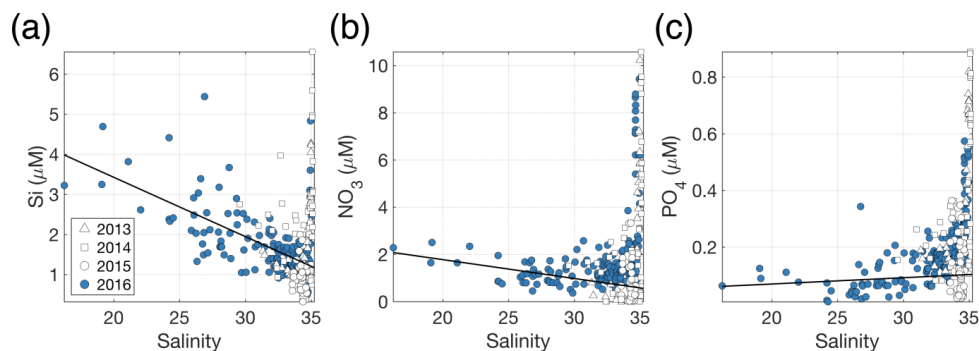


### 154 3.0 Effects of glacial discharge on marine inorganic nutrient distribution

155 One of the most direct mechanisms via which glacial discharge affects downstream marine  
156 primary production is by altering the availability of macronutrients (such as nitrate,  $\text{NO}_3$ ,  
157 phosphate,  $\text{PO}_4$ , and Si) and/or micronutrients (such as iron and manganese) in the ocean.  
158 The chemical composition of glacial discharge is now relatively well constrained, especially  
159 around Greenland (Yde et al., 2014; Meire et al., 2016a; Stevenson et al., 2017), Alaska  
160 (Hood and Berner, 2009; Schroth et al., 2011) and Svalbard (Hodson et al., 2004, 2016).  
161 Whilst Si is often associated with glacially-modified waters (Fig. 2a) around the Arctic  
162 (Azetsu-Scott and Syvitski, 1999; Brown et al., 2010; Meire et al., 2016a), the concentrations  
163 of all macronutrients in glacial discharge (Meire et al., 2016a) are relatively low compared to  
164 globally-averaged river water (Holmes et al., 2011). Indeed,  $\text{NO}_3$  and  $\text{PO}_4$  concentrations are  
165 often sufficiently low that a dilution of these inorganic nutrients with decreasing salinity can  
166 be observed in glacier fjords. In contrast, macronutrient concentrations in riverine estuaries  
167 typically increase with decreasing salinity.

#### Kongsfjorden (W Svalbard) 79° N 012° E

Kongsfjorden is a small Arctic fjord on the west coast of Svalbard notable for pronounced sediment plumes originating from multiple pro-glacial streams and several shallow marine-terminating glaciers. There is no sill at the fjord entrance and thus warm Atlantic water can be found throughout the fjord in summer (Hop et al., 2002). The major marine-terminating glaciers at the fjord head (Kongsvegen and Kronebreen) have been retreating since before monitoring began (Liestøl, 1988; Svendsen et al., 2002) and are anticipated to transition to land-terminating systems in the coming decades (Torsvik et al., 2019). Research within the fjord is logged in the RIS (Research in Svalbard; <https://researchinsvalbard.no>) online system.



168

169 Figure 2. (a) Si, (b)  $\text{NO}_3$  and (c)  $\text{PO}_4$  distributions across the measured salinity gradient in  
170 Kongsfjorden in summer 2013 (Fransson et al., 2016), 2014 (Fransson et al., 2016), 2015  
171 (van de Poll et al., 2018) and 2016 (Cantoni et al., unpublished data). Full depth data is  
172 shown, with a linear regression (black line) for glacially modified waters ( $S < 34.2$ ) during  
173 summer 2016. The position of stations varies between the datasets, with the 2016 data



174 providing the broadest coverage of the inner-fjord. Linear regression details are shown in  
175 Supplementary Table 1.

176 The low concentration of macronutrients in glacier discharge relative to saline waters is  
177 evidenced by the estuarine mixing diagram in Kongsfjorden (Fig. 2) and confirmed by  
178 extensive measurements of freshwater nutrient concentrations (Hodson et al., 2004, 2005).  
179 For  $\text{PO}_4$  (Fig. 2c) there is a slight increase in concentration with salinity (i.e. discharge dilutes  
180 the nutrient concentration in the fjord). For  $\text{NO}_3$ , discharge slightly increases the  
181 concentration in the upper-mixed layer (Fig. 2b). For Si, a steady decline in Si with  
182 increasing salinity (Fig. 2a) is consistent with a discharge associated Si supply (Brown et al.,  
183 2010; Meire et al., 2016a). The spatial distribution of data for summer 2013–2016 is similar  
184 and representative of summertime conditions in the fjord (Hop et al., 2002).

185 Whilst dissolved macronutrient concentrations in glacial discharge are relatively low, a  
186 characteristic of glaciated catchments is extremely high particulate Fe concentrations. High  
187 Fe concentrations arise both directly from glacier discharge and also from resuspension of  
188 glacially-derived sediments throughout the year (Markussen et al., 2016; Crusius et al.,  
189 2017). Total dissolvable Fe (TdFe) concentrations within Godthåbsfjord are high in all  
190 available datasets (May 2014, August 2014, July 2015) and strongly correlated with turbidity  
191 (linear regression:  $R^2 = 0.88$ ,  $R^2 = 0.56$  and  $R^2 = 0.88$ , respectively, Hopwood et al., 2016,  
192 2018). A critical question in oceanography is to what extent this large pool of particulate Fe,  
193 or dissolved Fe derived from it, is transferred into open-ocean environments and thus  
194 potentially able to enhance marine primary production in Fe-limited offshore regions  
195 (Gerringa et al., 2012; Arrigo et al., 2017; Schlosser et al., 2018).

196 Fe profiles around the Arctic show strong spatial variability in TdFe concentrations, ranging  
197 from unusually high concentrations of up to  $20 \mu\text{M}$  found intermittently close to turbid  
198 glacial outflows (Zhang et al., 2015; Markussen et al., 2016; Hopwood et al., 2018) to  
199 generally low nanomolar concentrations at the interface between shelf and fjord waters  
200 (Zhang et al., 2015; Crusius et al., 2017; Cape et al., 2019). An interesting feature of some of  
201 these profiles around Greenland is the presence of peak Fe at  $\sim 50$  m depth, suggesting that  
202 much of the Fe-transport away from glaciers may occur in subsurface glacially modified  
203 waters (Hopwood et al., 2018; Cape et al., 2019). The spatial extent of Fe enrichment  
204 downstream of glaciers around the Arctic is still uncertain, but there is evidence of variability  
205 downstream of glaciers on the scale of 10–100 km (Gerringa et al., 2012; Annett et al., 2017;  
206 Crusius et al., 2017).

### 207 3.1 Non-conservative mixing processes for Fe and Si

208 A key reason for uncertainty in the fate of glacially-derived Fe is the non-conservative  
209 behaviour of dissolved Fe in saline waters. In the absence of biological processes (i.e.  
210 nutrient assimilation and remineralization),  $\text{NO}_3$  is expected to exhibit conservative  
211 behaviour across estuarine salinity gradients (i.e. the concentration at any salinity is a linear  
212 function of mixing between fresh and saline waters). For Fe, however, a classic non-  
213 conservative estuarine behaviour occurs due to the removal of dissolved Fe ( $\text{DFe}^1$ ) as it

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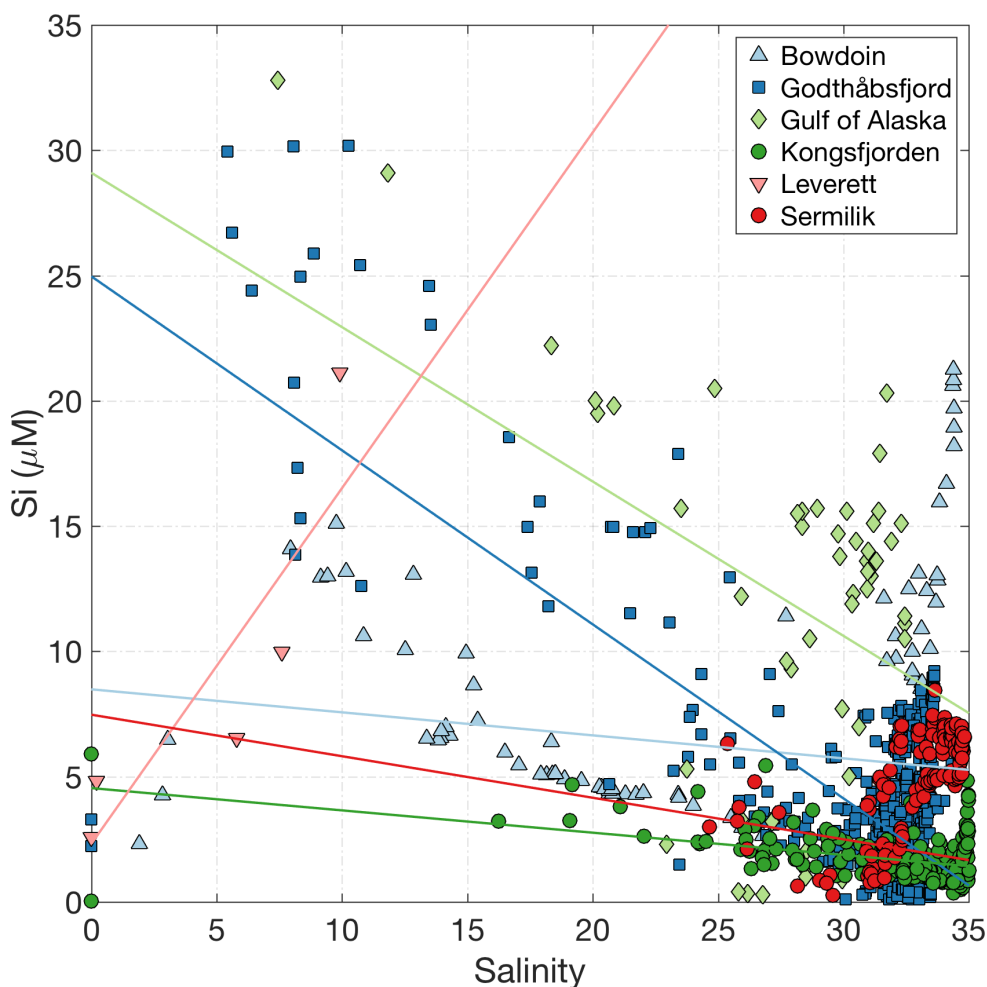
<sup>1</sup> For consistency, dissolved Fe is defined throughout operationally as  $<0.2 \mu\text{m}$  and is therefore inclusive of ionic, complexed, and colloidal species.



214 flocculates and is absorbed onto particle surfaces more readily at higher salinity and pH  
215 (Boyle et al., 1977). Dissolved Fe concentrations almost invariably exhibit strong (typically  
216 ~90%) non-conservative removal across estuarine salinity gradients (Boyle et al., 1977;  
217 Sholkovitz et al., 1978) and glaciated catchments appear to be no exception to this rule  
218 (Lippiatt et al., 2010). Dissolved Fe in Godthåbsfjord exhibits a removal of >80% DFe  
219 between salinities of 0–30 (Hopwood et al., 2016), and similar losses of approximately 98%  
220 for Kongsfjorden and 85% for the Copper river/estuary (Gulf of Alaska) system have been  
221 reported (Schroth et al., 2014; Zhang et al., 2015).

222 Conversely, Si is generally released from particulate phases during estuarine mixing,  
223 resulting in non-conservative addition to dissolved Si concentrations (Windom et al., 1991).  
224 This release of dissolved Si typically occurs at low salinities (Windom et al., 1991), with the  
225 behaviour of Si being more conservative at higher salinities (Brown et al., 2010). Estimating  
226 this release of Si from Kongsfjorden data (Fig. 2c) as the additional DSi present above the  
227 conservative mixing line for runoff mixing with unmodified saline water entering the fjord  
228 (via linear regression) suggests a Si enrichment of  $13 \pm 2\%$  (Fig. 2). This is broadly  
229 consistent with the 6–53% range reported for estuarine gradients in more temperate estuaries  
230 (Windom et al., 1991). Conversely, Hawkings et al. (2017) suggest a far greater dissolution  
231 downstream of Leverett glacier, equivalent to a 70–800% Si enrichment, and thus propose  
232 that the role of glaciers in the marine Si cycle has been underestimated. Given that such  
233 dissolution is substantially above the range observed in any other Arctic estuary, the apparent  
234 cause is worth further consideration.





235

236 Figure 3. Dissolved Si distribution vs. salinity for glaciated Arctic catchments. Data are from:  
237 Bowdoin Fjord (Kanna et al., 2018), Kongsfjorden (Fransson et al., 2016; van de Poll et al.,  
238 2018), Sermilik Fjord (Cape et al., 2019), Leverett Glacier (Hawkings et al., 2017),  
239 Godthåbsfjord (Hopwood et al., 2016; Meire et al., 2016b), and the Gulf of Alaska (Brown et  
240 al., 2010). Linear regressions are shown for surface (<20 m depth) data only. Linear  
241 regression details are shown in Supplementary Table 1.

242 The general distribution of Si in surface waters for Kongsfjorden (Fransson et al., 2016),  
243 Godthåbsfjord (Meire et al., 2016a), Bowdoin fjord (Kanna et al., 2018), Sermilik (Cape et  
244 al., 2019), and along the Gulf of Alaska (Brown et al., 2010) is similar; Si shows pseudo-  
245 conservative behaviour declining with increasing salinity in surface waters. The limited  
246 reported number of zero-salinity, or very low salinity, endmembers for Godthåbsfjord and  
247 Bowdoin are significantly below the linear regression derived from surface nutrient and  
248 salinity data (Fig. 3). In addition to some dissolution of particulate Si, another likely reason  
249 for this is the limitation of individual zero salinity data-points in dynamic fjord systems  
250 where different discharge outflows have different nutrient concentrations (Kanna et al.,



251 2018). As demonstrated by the two different zero salinity Si endmembers in Kongsfjorden  
252 (iceberg melt of  $\sim 0.03 \mu\text{M}$  and surface glacier discharge of  $\sim 5.9 \mu\text{M}$ ), pronounced deviations  
253 in nutrient content arise from mixing between various freshwater endmembers (surface  
254 discharge, ice melt, and subglacial discharge).

#### Bowdoin Fjord (NW Greenland) 78° N 069° W

Bowdoin Fjord is one of few glacier-fjord systems where biogeochemical and physical data are available in northern Greenland (Jouvet et al., 2018; Kanna et al., 2018). Bowdoin glacier, a small marine-terminating glacier at the fjord head, and four smaller land-terminating glaciers draining small ice caps isolated from the Greenland Ice Sheet drain into the fjord, which is typically subject to sea-ice cover until July. The fjord is  $\sim 20$  km long; the terminus of Bowdoin glacier is  $\sim 3$ -km wide.

255 Furthermore, macronutrient distributions in Bowdoin, Godthåbsfjord, and Sermilik  
256 unambiguously show that the main macronutrient supply associated with glacier discharge  
257 originates from mixing, rather than from sediment dissolution or freshwater addition (Meire  
258 et al., 2016a; Kanna et al., 2018; Cape et al., 2019). The apparently anomalous extent of Si  
259 dissolution downstream of Leverett Glacier (Hawkings et al., 2017) may therefore largely  
260 reflect underestimation of both the saline (assumed to be negligible) and freshwater  
261 endmembers, rather than unusually prolific particulate Si dissolution. In any case, Si  
262 concentrations downstream of Leverett fall within the range of other Arctic glacier estuaries  
263 (Fig. 3) making it challenging to support the hypothesis that glacial contributions to the Si  
264 cycle have been underestimated elsewhere (see also Tables 1 and 2).

### 265 3.2 Deriving glacier-ocean fluxes

266 The generally low concentrations of macronutrients and dissolved organic material (DOM) in  
267 glacier discharge, relative to coastal seawater (Table 1), has an important methodological  
268 implication because what constitutes a positive  $\text{NO}_3$ ,  $\text{PO}_4$ , or DOM flux into the Arctic Ocean  
269 in a glaciological context can actually reduce short-term nutrient availability in the marine  
270 environment. It is therefore necessary to consider both the glacier discharge and saline  
271 endmembers that mix in fjords, alongside fjord-scale circulation patterns, in order to  
272 constrain glacier-ocean fluxes correctly (Meire et al., 2016a; Hopwood et al., 2018; Kanna et  
273 al., 2018).



| Fjord                                      | Dataset  | Salinity  | NO <sub>3</sub> / μM                              | PO <sub>4</sub> / μM                                 | Si / μM   | TdFe / μM                   |
|--|--|---|---|--|---|-----------------------------|
| Kongsfjorden (Svalbard)                    | Summer 2016 (Cantoni et al., unpublished data)         | 0.0 (Ice melt)<br>0.0 (Surface discharge)<br>34.50 ± 0.17   | 0.87 ± 1.0<br>0.94 ± 1.0<br>1.25 ± 0.49           | 0.02 ± 0.03<br>0.057 ± 0.31<br>0.20 ± 0.06           | 0.03 ± 0.03<br>5.91 ± 4.1<br>1.00 ± 0.33              | 33.8 ± 100<br>74 ± 76<br>ND |
| Nuup Kangerlua / Godthåbsfjord (Greenland) | Summer 2014 (Hopwood et al., 2016; Meire et al., 2016) | 0.0 (Ice melt)<br>0.0 (Surface discharge)   | 1.96 ± 1.68<br>1.60 ± 0.44<br>11.5 ± 1.5          | 0.04 ± 0.04<br>0.02 ± 0.01<br>0.79 ± 0.04            | ND<br>12.2 ± 16.3<br>8.0 ± 1.0                        | 0.31 ± 0.49<br>13.8<br>ND   |
| Sermilik (Greenland)                       | Summer 2015 (Cape et al., 2019)                        | 0.0 (Subglacial discharge)<br>0.0 (Ice melt)<br>34.9 ± 0.1  | 1.8 ± 0.5<br>0.97 ± 1.5<br>12.8 ± 1               | ND<br>ND<br>ND                                       | 10 ± 8<br>4 ± 4<br>6.15 ± 1                           | ND<br>ND<br>ND              |
| Bowdoin (Greenland)                        | Summer 2016 (Kanna et al., 2018)                       | 0.0 (Surface discharge)<br>34.3 ± 0.1   | 0.22 ± 0.15<br>14.7 ± 0.9                         | 0.30 ± 0.20<br>1.1 ± 0.1                             | BD<br>19.5 ± 1.5                                      | ND<br>ND                    |
| Young Sound (Greenland)                    | Summer 2014 (Paulsen et al., 2017)                     | 0.0 (Runoff July-August)<br>0.0 (Runoff September-October)<br>33.6 ± 0.1 (July-August)<br>33.5 ± 0.04 (September-October) | 1.2 ± 0.74<br>1.0 ± 0.7<br>6.4 ± 1.1<br>5.6 ± 0.2 | 0.29 ± 0.2<br>0.35 ± 0.2<br>1.18 ± 0.5<br>0.62 ± 0.2 | 9.52 ± 3.8<br>29.57 ± 10.9<br>6.66 ± 0.4<br>6.5 ± 0.1 | ND<br>ND<br>ND<br>ND        |

274 Table 1. Measured/computed discharge and saline endmembers for well-studied Arctic fjords (ND, not determined/not reported; BD,  
 275 below detection).

276



| Nutrient        | Freshwater endmember concentration / $\mu\text{M}$ | Flux                          | Estuarine modification                      | Data                       |
|-----------------|--|-------------------------------|---|----------------------------|
| Fe              | 0.13   | >26 Mmol yr <sup>-1</sup>     | Inclusive, >80% loss                        | Hopwood et al., 2016       |
|                 | 1.64   | 39 Mmol yr <sup>-1</sup>      | Assumed 90% loss                            | Stevenson et al., 2017     |
|                 | 0.053  | 53 Mmol yr <sup>-1</sup>      | Discussed, not applied                      | Statham et al., 2008       |
|                 | 3.70   | 180 Mmol yr <sup>-1</sup>     | Assumed 90% loss                            | Bhatia et al., 2013a       |
|                 | 0.71   | 290 Mmol yr <sup>-1</sup>     | Discussed, not applied                      | Hawkings et al., 2014      |
| DOC             | 16-100   | 6.7 Gmol yr <sup>-1</sup>     | Not discussed                               | Bhatia et al., 2010, 2013b |
|                 | 12-41  | 11-14 Gmol yr <sup>-1</sup>   | Not discussed                               | Lawson et al., 2014b       |
|                 | 15-100   | 18 Gmol yr <sup>-1</sup>      | Not discussed                               | Hood et al., 2015          |
|                 | 2-290  | 24-38 Gmol yr <sup>-1</sup>   | Not discussed                               | Csank et al., 2019         |
|                 | 27-47  | 40 Gmol yr <sup>-1</sup>      | Not discussed                               | Paulsen et al., 2017       |
|                 |  |                               |   | Not discussed              |
| DON             | 2.3  | 2.3 Gmol yr <sup>-1</sup>     | Not discussed                               | Paulsen et al., 2017       |
|                 | 4.7 – 5.4  | 5 Gmol yr <sup>-1</sup>       | Not discussed                               |                            |
| Si              | 13 (ice) 28 (meltwater)                            | 22 Gmol yr <sup>-1</sup>      | Inclusive                                   | Meire et al., 2016a        |
|                 | 9.6  | 4 Gmol yr <sup>-1</sup>       | Discussed (+190 Gmol yr <sup>-1</sup> ASi)  | Hawkings et al., 2017      |
| PO <sub>4</sub> | 0.23   | 0.10 Gmol yr <sup>-1</sup>    | Discussed (+0.23 Gmol yr <sup>-1</sup> LPP) | Hawkings et al., 2016      |
|                 | 0.26   | 0.26 Gmol yr <sup>-1</sup>    | Not discussed                               | Meire et al., 2016a        |
| NO <sub>3</sub> | 1.4 (ice) 1.5 (meltwater)                          | 0.42 Gmol yr <sup>-1</sup>    | Not discussed                               | Wadham et al., 2016        |
|                 | 0.5-1.7  | 0.5-1.7 Gmol yr <sup>-1</sup> | Not discussed                               | Paulsen et al., 2017       |
|                 | 1.79   | 1.79 Gmol yr <sup>-1</sup>    | Not discussed                               | Meire et al., 2016a        |

277 Table 2. Flux calculations for dissolved nutrients (Fe, DOC, DON, NO<sub>3</sub>, PO<sub>4</sub>, Si) from Greenland Ice Sheet discharge. Where a flux  
 278 was not calculated in the original work, an assumed discharge volume of 1000 km<sup>3</sup> yr<sup>-1</sup> is used to derive a flux for comparative  
 279 purposes (ASi, amorphous silica; LPP, labile particulate phosphorous). For DOM, PO<sub>4</sub>, and NO<sub>3</sub>, non-conservative estuarine  
 280 behaviour is expected to be minor or negligible.



281 Despite the relatively well-constrained nutrient signature of glacial discharge globally,  
282 estimated fluxes of nutrients from glaciers to the ocean appear to be subject to greater  
283 variability, especially for nutrients subject to non-conservative mixing (Table 2). Estimates of  
284 the Fe flux from the Greenland Ice Sheet, for example, have an 11-fold difference between  
285 the lowest ( $>26 \text{ Mmol yr}^{-1}$ ) and highest ( $290 \text{ Mmol yr}^{-1}$ ) values (Hawkings et al., 2014;  
286 Stevenson et al., 2017). A scaled-up calculation using freshwater concentrations (C) and  
287 discharge volumes (Q) is the simplest way of determining the flux from a glaciated catchment  
288 to the ocean. However, discharge nutrient concentrations vary seasonally, often resulting in  
289 variable C-Q relationships due to changes in mixing ratios between different discharge flow  
290 paths, post-mixing reactions, and seasonal changes in microbial behaviour in the snowpack,  
291 on glacier surfaces, and in proglacial forefields (Brown et al., 1994; Hodson et al., 2005).  
292 Therefore, full seasonal data sets from a range of representative glaciers are required to  
293 accurately describe C-Q relationships. Furthermore, as the indirect effects of discharge on  
294 nutrient availability to phytoplankton via estuarine circulation and stratification are expected  
295 to be a greater influence than the direct nutrient outflow associated with discharge (Rysgaard  
296 et al., 2003; Juul-Pedersen et al., 2015; Meire et al., 2016a), freshwater data must be coupled  
297 to physical and chemical time series in the coastal environment if the net effect of discharge  
298 on nutrient availability in the marine environment is to be understood. Indeed, the recently  
299 emphasized hypothesis that macronutrient fluxes from glaciers into the ocean have been  
300 significantly underestimated (Hawkings et al., 2016, 2014, 2017; Wadham et al., 2016) is  
301 difficult to reconcile with a synthesis of available nutrient distributions in glaciated Arctic  
302 catchments, especially for Si (Fig. 3) and Fe (Table 2).

#### Young Sound-Tyrolerfjord (NE Greenland) $74^\circ \text{ N } 021^\circ \text{ W}$

Young Sound-Tyrolerfjord is a catchment fed by rivers from three land-terminating glaciers. Tyrolerfjord is the narrow innermost part of the fjord system in the west, and Young Sound is the wider outer part in the east towards the Atlantic. The fjord system has a surface area of  $390 \text{ km}^2$ , a length of 90 km, and a maximum depth of 360 m. A shallow  $\sim 45 \text{ m}$  deep sill restricts exchange with the Greenland shelf and summertime productivity in the fjord is among the lowest measured in the Arctic (as low as  $<40 \text{ mg C m}^{-2} \text{ day}^{-1}$ ). In recent years, fjord waters have freshened (Sejr et al., 2017), and freshening of coastal waters has prevented renewal of fjord bottom waters (Boone et al., 2018). A data portal is available reporting work done in the catchment through the Greenland Ecosystem Monitoring Programme (GEM; <http://g-e-m.dk>).

303 Differences in Fe flux calculations from glaciers to the ocean largely arise because of the  
304 estuarine removal factor applied. Given that the difference between an estimated removal  
305 factor of 90% and 99% is a factor of 10 difference in the calculated DFe flux, there is overlap  
306 in all of the calculated fluxes for Greenland Ice Sheet discharge into the ocean (Table 2)  
307 (Statham et al., 2008; Bhatia et al., 2013a; Hawkings et al., 2014; Stevenson et al., 2017).  
308 Conversely, estimates of DOM export (quantified as DOC) are confined to a slightly  
309 narrower range of  $7\text{--}40 \text{ Gmol yr}^{-1}$ , with differences arising from changes in measured DOM  
310 concentrations (Bhatia et al., 2013b; Lawson et al., 2014b; Hood et al., 2015). The  
311 characterization of glacial DOM, with respect to its lability, C:N ratio, and implications for



312 bacterial productivity in the marine environment (Hood et al., 2015; Paulsen et al., 2017) is  
313 however not readily apparent from a simple flux calculation.

314 A particularly interesting case study concerning the link between marine primary production,  
315 circulation and discharge-derived nutrient fluxes is Young Sound. It was initially stipulated  
316 that increasing discharge into the fjord in response to climate change would increase  
317 estuarine circulation and therefore macronutrient supply. Combined with a longer sea-ice free  
318 growing season as Arctic temperatures increase, this would be expected to increase primary  
319 production within the fjord (Rysgaard et al., 1999; Rysgaard and Glud, 2007). Yet freshwater  
320 input also stratifies the fjord throughout summer and ensures low macronutrient availability  
321 in surface waters (Bendtsen et al., 2014; Meire et al., 2016a), which results in low  
322 summertime productivity in the inner- and central fjord ( $<40 \text{ mg C m}^{-2} \text{ day}^{-1}$ ) (Rysgaard et  
323 al., 1999, 2003; Rysgaard and Glud, 2007). Whilst annual discharge volumes into the fjord  
324 have indeed increased over the past two decades, resulting in a mean annual  $0.12 \pm 0.05$   
325 (practical salinity units) freshening of fjord waters (Sejr et al., 2017), shelf waters have also  
326 freshened. This has impeded the dense inflow of saline waters into the fjord (Boone et al.,  
327 2018), and therefore counteracted the expected increase in productivity.

### 328 **3.3 How do variations in the behaviour and location of higher trophic levels affect** 329 **nutrient availability to marine micro-organisms?**

330 With the exception of some zooplankton and fish species that struggle to adapt to the strong  
331 salinity gradients and/or suspended particle loads in inner-fjord environments (Węsławski W  
332 and Legezyńska, 1998; Lydersen et al., 2014), higher trophic level organisms (including  
333 mammals and birds) are not directly affected by the physical/chemical gradients caused by  
334 glacier discharge. However, their food sources, such as zooplankton and some fish species,  
335 are directly affected and therefore there are many examples of higher level organisms  
336 adapting their feeding strategies within glacier fjord environments (Arimitsu et al., 2012;  
337 Renner et al., 2012; Laidre et al., 2016).

338 It is debatable to what extent shifts in these feeding patterns could have broad-scale  
339 biogeochemical effects. Whilst some species are widely described as “ecosystem engineers”,  
340 such as *Alle alle* (the Little Auk) in the Greenland North Water Polynya (González-  
341 Bergonzoni et al., 2017), for changes in higher-trophic level organisms’ feeding habits to  
342 have significant direct biogeochemical effects on the scale of a glacier-fjord system would  
343 require relatively large concentrations of such animals. Never-the-less, in some specific  
344 ‘hotspot’ regions this effect is significant enough to be measurable. There is ample evidence  
345 that birds intentionally target upwelling plumes in front of glaciers as feeding grounds,  
346 possibly due to the stunning effect that these turbid, upwelling plumes have upon prey such  
347 as zooplankton (Hop et al., 2002; Lydersen et al., 2014). This feeding activity therefore  
348 concentrates the effect of avian nutrient-recycling within a smaller area than would otherwise  
349 be the case, potentially leading to modest nutrient enrichment of these proglacial  
350 environments. Yet, with the exception of large, concentrated bird colonies, the effects of such  
351 activity are likely modest. In Kongsfjorden, bird populations are well studied, and in several  
352 species are associated with feeding in pro-glacial plumes, yet still collectively consume only  
353 between 0.1 and 5.3% of the carbon produced by phytoplankton in the fjord (Hop et al.,  
354 2002). The estimated corresponding nutrient flux into the fjord from birds is  $2 \text{ mmol m}^{-2} \text{ yr}^{-1}$   
355 nitrogen and  $0.3 \text{ mmol m}^{-2} \text{ yr}^{-1}$  phosphorous.



#### 356 4.0 Critical differences between surface and subglacial discharge release

##### Sermilik Fjord (SE Greenland) 66° N 038° W

Sermilik fjord is home to Helheim glacier, Greenland's fifth largest in terms of annual discharge volume. The fjord is ~100 km long and ~600–900 m deep, with no sill to restrict fjord-shelf exchange. The circulation of watermasses within the fjord, fjord-shelf exchange (Straneo et al., 2011; Beaird et al., 2018), and iceberg dynamics along the fjord have all been characterised. Whilst a large fraction (40–60%) of freshwater from Greenland enters the ocean as solid ice, rather than as meltwater discharge, surprisingly little is known about the fate and effects of this component in the marine environment (Sutherland et al., 2014; Enderlin et al., 2018; Moon et al., 2018).

357 Critical differences arise between land-terminating and marine-terminating glaciers with  
358 respect to their effects on water column structure. Glacier fjord surveys have shown that  
359 fjords with large marine-terminating glaciers around the Arctic are normally more productive  
360 than their land-terminating glacier-fjord counterparts (Meire et al., 2017; Kanna et al., 2018).  
361 A particularly critical insight is that fjord-scale summertime productivity along the west  
362 Greenland coastline scales approximately with discharge downstream of marine-terminating  
363 glaciers, but not land-terminating glaciers (Meire et al., 2017). The primary explanation for  
364 this phenomenon is the vertical nutrient flux associated with mixing driven by subglacial  
365 discharge plumes, which has been quantified in field studies at Bowdoin glacier (Kanna et al.,  
366 2018), Sermilik fjord (Cape et al., 2019), Kongsfjorden (Halbach et al., 2019) and in  
367 Godthåbsfjord (Meire et al., 2016a). As discharge is released at the glacial grounding line  
368 depth, its buoyancy and momentum result in an upwelling plume that entrains and mixes with  
369 ambient seawater (Carroll et al., 2015, 2016; Cowton et al., 2015). In Bowdoin, Sermilik, and  
370 Godthåbsfjord, this 'nutrient pump' provides 99%, 97%, and 87%, respectively, of the NO<sub>3</sub>  
371 associated with glacier inputs to each fjord system (Meire et al., 2016a; Kanna et al.,  
372 2018; Cape et al., 2019).

373 Whilst the pan-Arctic magnitude of this 'nutrient pump' is challenging to quantify because of  
374 the uniqueness of glacier-fjord systems in terms of their geometry, circulation, residence  
375 time, and glacier grounding line depths (Straneo and Cenedese, 2015; Morlighem et al.,  
376 2017), it can be approximated in generic terms because plume theory (Morton et al., 1956)  
377 has been used extensively to describe subglacial discharge plumes in the marine environment  
378 (Jenkins, 2011). Estimates of subglacial discharge for the 12 Greenland glacier-fjord systems  
379 where sufficient data are available to simulate plume entrainment (Carroll et al., 2016)  
380 suggest that the entrainment effect is at least two orders of magnitude more important for  
381 macronutrient availability than direct freshwater runoff (Hopwood et al., 2018). This is  
382 consistent with limited available field observations (Meire et al., 2016a; Kanna et al., 2018;  
383 Cape et al., 2019). As macronutrient fluxes have been estimated independently using  
384 different datasets and plume entrainment models in two of these glacier-fjord systems  
385 (Sermilik and Illulissat), an assessment of the robustness of these fluxes can also be made  
386 (Table 3) (Hopwood et al., 2018; Cape et al., 2019). Despite different definitions of the  
387 macronutrient flux (Table 3; **A** refers to the out-of-fjord transport at a defined fjord cross-



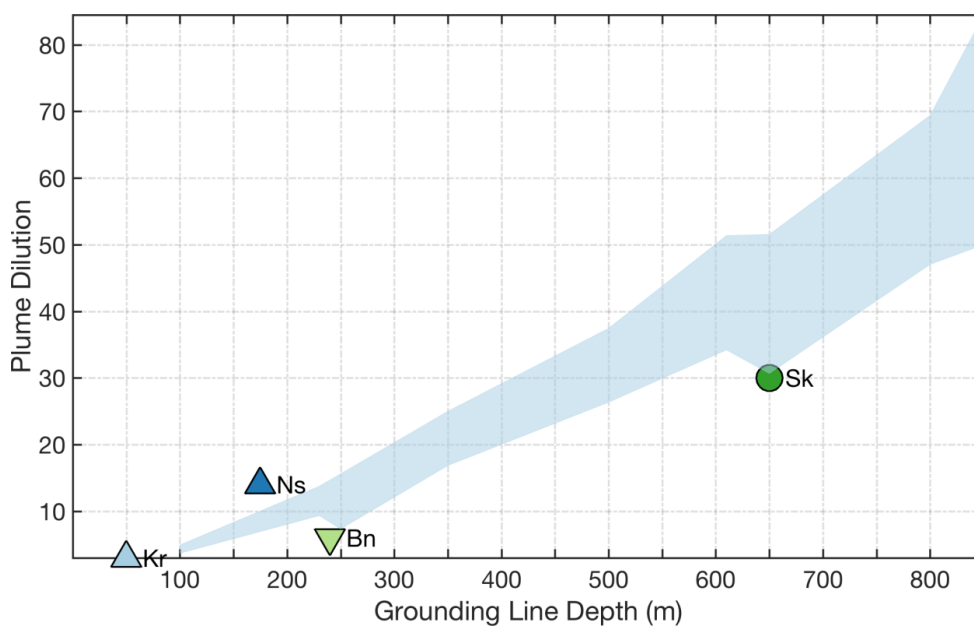
388 section window, whereas **B** refers to the vertical transport within the immediate vicinity of  
 389 the glacier), the fluxes are reasonably comparable and in both cases unambiguously dominate  
 390 macronutrient glacier associated input into these fjord systems (Hopwood et al., 2018; Cape  
 391 et al., 2019).

| Location                                  | Field campaign(s)<br>for <b>A</b> | <b>A</b> Calculated out-fjord<br>NO <sub>3</sub> export Gmol yr <sup>-1</sup> | <b>B</b> Idealized NO <sub>3</sub><br>upwelling Gmol yr <sup>-1</sup> |
|---|-----------------------------------|---|---|
| Ilulissat Icefjord<br>(Jakobshavn Isbrae) | 2000-2016                         | 2.9 ± 0.9   | 4.2   |
| Sermilik (Helheim<br>glacier)             | 2015                              | 0.88  | 2.0   |
| Sermilik (Helheim<br>glacier)             | 2000-2016                         | 1.2 ± 0.3   |   |

392 Table 3. A comparison of upwelled NO<sub>3</sub> fluxes calculated from fjord-specific observed  
 393 nutrient distributions (**A**) (Cape et al., 2019) and using regional nutrient profiles with  
 394 idealized plume theory (**B**) (Hopwood et al., 2018). **A** refers to the out-of-fjord transport of  
 395 nutrients, whereas **B** refers to the vertical transport close to the glacier terminus.

396 Whilst large compared to changes in macronutrient availability from discharge without  
 397 entrainment (Table 2), it should be noted that these nutrient fluxes (Table 3) are still only  
 398 intermediate contributions to fjord-scale macronutrient supply compared to total annual  
 399 consumption in these environments. For example, in Godthåbsfjord mean annual primary  
 400 production is 103.7 g C m<sup>-2</sup> yr<sup>-1</sup>, equivalent to biological consumption of 1.1 mol N m<sup>-2</sup> yr<sup>-1</sup>.  
 401 Entrainment from the three marine-terminating glaciers within the fjord is conservatively  
 402 estimated to supply 0.01-0.12 mol N m<sup>-2</sup> yr<sup>-1</sup> (Meire et al., 2017) i.e. 1-11% of the total N  
 403 supply required for primary production if production were supported exclusively by new NO<sub>3</sub>  
 404 (rather than recycling) and equally distributed across the entire fjord surface. Whilst this is  
 405 consistent with observations suggesting relative stability in mean annual primary production  
 406 in Godthåbsfjord from 2005-2012 (103.7 ± 17.8 g C m<sup>-2</sup> yr<sup>-1</sup>; Juul-Pedersen et al., 2015),  
 407 despite pronounced increases in total discharge into the fjord, this does not preclude a much  
 408 stronger influence of entrainment on primary production in the inner-fjord environment. The  
 409 time series is constructed at the fjord mouth, over 120 km from the nearest glacier, and the  
 410 estimates of subglacial discharge and entrainment used by Meire et al., (2017) are both  
 411 unrealistically low. If the same conservative estimate of entrainment is assumed to only affect  
 412 productivity in the main fjord branch (where the 3 marine-terminating glaciers are located),  
 413 for example, the lower bound for the contribution of entrainment becomes 3-33% of total N  
 414 supply. Similarly, in Kongsfjorden- the surface area of which is considerably smaller  
 415 compared to Godthåbsfjord (~230 km<sup>2</sup> compared to 650 km<sup>2</sup>)- even the relatively weak  
 416 entrainment from shallow marine-terminating glaciers (Fig. 4) accounts for approximately  
 417 19-32% of N supply. An additional mechanism of N supply evident there, which partially  
 418 offsets the inefficiency of macronutrient entrainment at shallow grounding line depths, is the  
 419 entrainment of ammonium from shallow benthic sources (Halbach et al., 2019). Changes in  
 420 subglacial discharge, or in the entrainment factor (e.g. from a shift in glacier grounding line  
 421 depth, Carroll et al., 2016) can therefore potentially change fjord-scale productivity.





422

423 Figure 4. The plume dilution (entrainment) factor relationship with glacier grounding line  
424 depth as modelled by Carroll et al., (2016) for subglacial freshwater discharge rates of 250–  
425  $500 \text{ m}^3 \text{ s}^{-1}$  and grounding lines of >100 m (shaded area). Also shown are the entrainment  
426 factors determined from field observations for Kronebreen (Kongsfjorden, Kr, Halbach et al.,  
427 2019), Bowdoin (Bn, Kanna et al., 2018) and Narsap Sermia (Ns, Meire et al., 2016a)  
428 (derived from CTD profiles and/or nutrient budgets) and Sermilik (Sk, Beaird et al., 2018)  
429 (derived from CTD profiles and noble gases).

430 A specific deficiency in the literature to date is the absence of measured subglacial discharge  
431 rates from marine-terminating glaciers. Variability in such rates on diurnal and seasonal  
432 timescales is expected (Schild et al., 2016) and intermittent periods of extremely high  
433 discharge are known to occur, for example from ice-dammed lake drainage in Godthåbsfjord  
434 (Kjeldsen et al., 2014). Yet determining the extent to which these events affect fjord-scale  
435 mixing and biogeochemistry will certainly require further field observations. Paradoxically,  
436 one of the major knowledge gaps concerning low-frequency, high-discharge events is their  
437 biological effects; yet these events first became characterised in Godthåbsfjord after  
438 observations by a fisherman of a sudden *Sebastes marinus* (Redfish) mortality event in the  
439 vicinity of a marine-terminating glacier terminus. These unfortunate fish were propelled  
440 rapidly to the surface by ascending freshwater during a high discharge event (Kjeldsen et al.,  
441 2014).

## 442 5.0 Contrasting Fe and $\text{NO}_3$ limited regions of the ocean

443 Whether or not nutrients transported to the ocean surface have an immediate positive effect  
444 on marine primary production depends on the identity of the resource(s) that limit marine  
445 primary production. Light attenuation is the ultimate limiting control on marine primary  
446 production and this is exacerbated close to turbid glacial outflows (Hop et al., 2002; Arimitsu  
447 et al., 2012; Murray et al., 2015). However the spatial extent of sediment plumes and/or ice



448 mélange, which limit light penetration into the water column, is typically restricted to within  
449 a few kilometres of the glacier terminus (Arimitsu et al., 2012; Hudson et al., 2014; Lydersen  
450 et al., 2014). Beyond the turbid, light-limited vicinity of glacial outflows, the proximal  
451 limiting resource for summertime marine primary production will likely be a nutrient, the  
452 identity of which varies with location globally (Moore et al., 2013). Increasing the supply of  
453 the proximal limiting nutrient would be expected to have a positive influence on marine  
454 primary production, whereas increasing the supply of other nutrients alone would not; a  
455 premise of ‘the law of the minimum’ (Debaar, 1994).

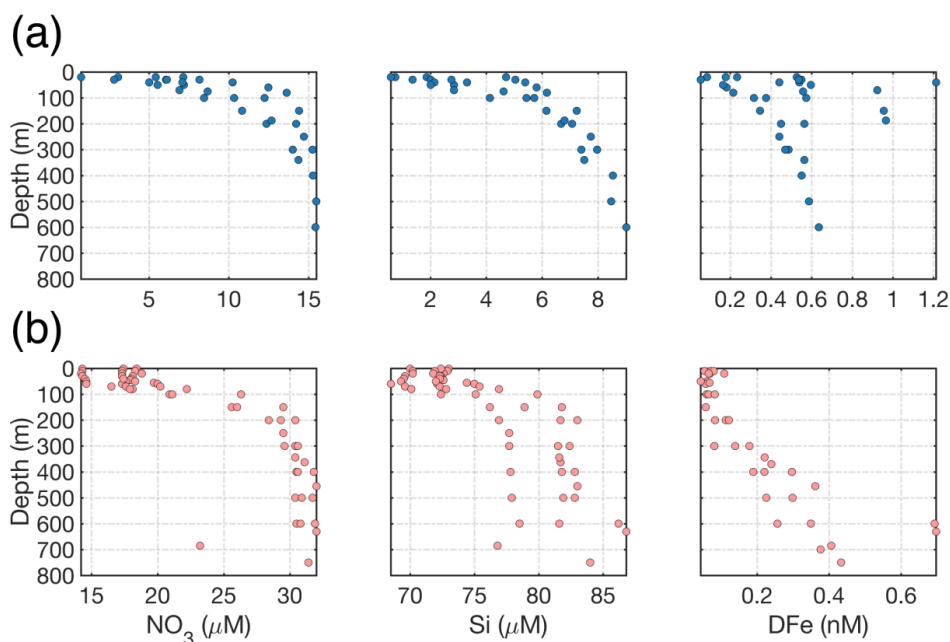
456 The continental shelf is a major source of Fe into the ocean (Lam and Bishop, 2008; Charette  
457 et al., 2016), and this results in clear differences in proximal limiting nutrients between Arctic  
458 and Antarctic marine environments. The isolated Southern Ocean is the world’s largest High-  
459 Nitrate, Low-Chlorophyll (HNLC) zone, where Fe extensively limits primary production and  
460 macronutrients are generally present at high concentrations in surface waters (Martin et al.,  
461 1990a, 1990b). Conversely, the Arctic Ocean is exposed to extensive broad shelf areas and  
462 thus generally has a greater availability of Fe relative to macronutrient supply (Klunder et al.,  
463 2012). Fe-limited summertime conditions have been reported in parts of the Arctic  
464 (Nielsdottir et al., 2009; Ryan-Keogh et al., 2013; Rijkenberg et al., 2018;), but are spatially  
465 and temporally limited compared to the geographically extensive HNLC conditions in the  
466 Southern Ocean.

467 However, few experimental studies have directly assessed the nutrient limitation status of  
468 regions within the vicinity of glaciated Arctic catchments. With extremely high Fe input into  
469 these catchments, NO<sub>3</sub> limitation might be expected year-round. However, PO<sub>4</sub> limitation is  
470 also plausible close to glaciers in strongly-stratified fjords, due to the low availability of PO<sub>4</sub>  
471 in freshwater (Prado-Fiedler, 2009). Conversely, in the Southern Ocean, it is possible that Fe-  
472 limited conditions occur extremely close to glaciers and ice shelves (Fig. 5). High-NO<sub>3</sub>, low-  
473 Fe water can be found in the immediate vicinity of Antarctica’s coastline (Gerringa et al.,  
474 2012; Marsay et al., 2017), and even in inshore bays (Annett et al., 2015; Höfer et al., 2019).  
475 Macronutrient data from Maxwell Bay (King George Island, South Shetland Islands), for  
476 example, suggests that Fe from local glaciers mixes with high-NO<sub>3</sub>, high-Si ocean waters,  
477 providing ideal conditions for phytoplankton blooms in terms of nutrient availability. The  
478 lowest surface macronutrient concentrations measured in Maxwell Bay in a summer  
479 campaign were 17 μM NO<sub>3</sub>, 1.4 μM PO<sub>4</sub>, and 47 μM Si (Höfer et al., 2019). Similarly, in  
480 Ryder Bay (Antarctic Peninsula), the lowest measured annual macronutrient concentrations-  
481 occurring after strong drawdown during a pronounced phytoplankton bloom (22 mg m<sup>-3</sup>  
482 chlorophyll a)- were 2.5 μM NO<sub>3</sub> and 0.4 μM PO<sub>4</sub> (Annett et al., 2015). This contrasts starkly  
483 with the summertime surface macronutrient distribution in glaciated fjords in the Arctic,  
484 including Kongsfjorden (Fig. 2), where surface macronutrient concentrations are typically  
485 depleted throughout summer.

486 For a hypothetical nutrient-flux from a glacier, the same flux could be envisaged in two  
487 endmember scenarios; one several kilometres inside an Arctic fjord (e.g. Godthåbsfjord or  
488 Kongsfjorden) and one at the coastline of an isolated Southern Ocean island such as the  
489 Kerguelen (Bucciarelli et al., 2001) or South Shetland Islands (Höfer et al., 2019). In the  
490 Arctic fjord, a pronounced Fe flux from a summertime discharge would likely have no  
491 immediate positive effect upon fjord-scale marine primary production because Fe may  
492 already be replete (Hopwood et al., 2016; Crusius et al., 2017). This is consistent with the



493 observation that Fe-rich discharge from land-terminating glaciers around west Greenland  
494 does not have a positive fjord-scale fertilization effect (Meire et al., 2017). Conversely, the  
495 same Fe input into coastal waters around the Kerguelen Islands would be expected to have a  
496 pronounced positive effect upon marine primary production, because the islands occur with  
497 the world's largest HNLC zone. Wherever Fe is advected offshore in the wake of the islands,  
498 a positive effect on primary production is expected (Blain et al., 2001; Bucciarelli et al.,  
499 2001) even though there are marked changes in the phytoplankton community composition  
500 between the Fe-enriched bloom region (dominated by microphytoplankton) and the offshore  
501 HNLC area (dominated by small diatoms and nanoflagellates) (Uitz et al., 2009).



502

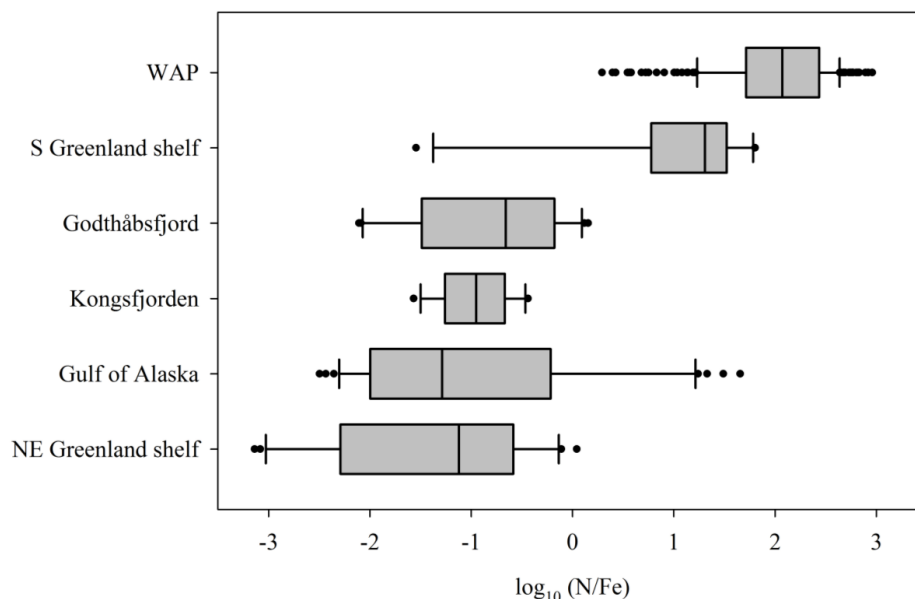
503 Figure 5. Contrasting nutrient properties of water on the (a) southeast Greenland shelf, data  
504 from Achterberg et al., (2018), with (b) the Ross Sea shelf, data from Marsay et al., (2017).  
505 Note the different scales used on the x-axes.

### 506 5.1 The subglacial discharge ‘pump’; from macronutrients to iron

507 The effect of the subglacial discharge ‘nutrient pump’ may similarly vary with location.  
508 Contrasting the  $\text{NO}_3$  and DFe concentrations of marine environments observed adjacent to  
509 different glacier systems suggests substantial variations in the proximal limiting nutrient of  
510 these waters on a global scale (Fig. 6). In Antarctic shelf regions, such as the Western  
511 Antarctic Peninsula, a high ratio of summertime  $\text{NO}_3$ :DFe (median value 1.5) is indicative of  
512 Fe-limitation. Across the Arctic there is a broader range of ratios (median values -2.2 to 1.3)  
513 indicating spatial variability in the balance between Fe and  $\text{NO}_3$ -limitation (Fig. 6). Variation  
514 is evident even within specific regions. The range of  $\text{NO}_3$ :DFe ratios for both the Gulf of  
515 Alaska ( $\log_{10}$  -2.5 to 1.7) and the south Greenland shelf ( $\log_{10}$  -1.5 to 1.8) include values that  
516 are indicative of the full spectrum of responses from  $\text{NO}_3$ -limitation, to Fe/ $\text{NO}_3$  co-limitation,  
517 to Fe-limitation (Browning et al., 2017). This suggests a relatively rapid spatial transition



518 from excess to deficient DFe conditions.



519  
520 Figure 6. Variations in the ratio of dissolved  $\text{NO}_3$  and Fe in surface waters (< 20 m) adjacent  
521 to glaciated regions: whiskers show 10<sup>th</sup> and 90<sup>th</sup> percentiles; bars median, 25<sup>th</sup> and 75<sup>th</sup>  
522 percentiles; dots all outliers. Data from: Western Antarctic Peninsula (WAP, Annett et al.,  
523 2017; Ducklow et al., 2017), the South Greenland shelf (Achterberg et al., 2018; Tonnard et  
524 al., 2018), Godthåbsfjord (Hopwood et al., 2016), Kongsfjorden (Hopwood et al., 2017), the  
525 Gulf of Alaska (Lippiatt et al., 2010) and the NE Greenland shelf (Hopwood et al., 2018). For  
526 Kongsfjorden,  $\text{NO}_3$  and Fe data were interpolated using the  $\text{NO}_3$ /salinity relationship.

527 How would the marine-terminating glacier upwelling effect operate in a Fe-limited system?  
528 The physical mechanism of a ‘nutrient pump’ would be identical for glaciers with the same  
529 discharge and grounding line; one in a high-Fe, low- $\text{NO}_3$  Arctic system and one in a low-Fe,  
530 high- $\text{NO}_3$  Antarctic system. However, the biogeochemical consequences with respect to  
531 marine primary production would be different (Table 4). In the case of subglacial discharge,  
532 for simplicity, we consider a mid-depth glacier (grounding line of 100–250 m below sea-  
533 level) with a constant discharge rate of  $250 \text{ m}^3 \text{ s}^{-1}$ . An entrainment factor of 6–10 would then  
534 be predicted by plume theory (Fig. 4) (Carroll et al., 2016). In a Greenland fjord with no sill  
535 to constrain circulation and residence time short enough that inflowing nutrient  
536 concentrations were not changed significantly prior to entrainment, an average  $\text{NO}_3$   
537 concentration of 5–12  $\mu\text{M}$  is predicted in the entrained water compared to  $\sim 2 \mu\text{M}$  in glacier  
538 discharge (Hopwood et al., 2018). Over a 2-month discharge period, this would produce a  
539  $\text{NO}_3$  flux of 40–160  $\text{Mmol NO}_3$ , with 2–6% of the  $\text{NO}_3$  flux arising from meltwater discharge  
540 and 94–98% from plume entrainment. Complete utilization of this  $\text{NO}_3$  by phytoplankton  
541 according to the Redfield ratio (106 C:16 N) (Redfield, 1934), would correspond to a  
542 biological sink of 0.27–1.0  $\text{Gmol C}$ .

543 In an analogous HNLC environment, surface  $\text{NO}_3$  requirements would already vastly exceed  
544 phytoplankton requirements (Fig. 6) due to extensive Fe-limitation of primary production.



545 Thus, whilst the upwelled  $\text{NO}_3$  flux would be larger in a Fe-limited system, due to higher  
 546 concentrations of  $\text{NO}_3$  in the water column (see Fig. 5), the short-term biological effect of  
 547 upwelling  $\text{NO}_3$  alone would be negligible. More important would be the upwelling of the  
 548 proximal limiting nutrient Fe. If we assume that dissolved Fe in the marine water column is  
 549 in a stable, bioavailable form, and that additional dissolved Fe from freshwater is delivered to  
 550 the marine environment with a 90–99% loss during estuarine mixing (Table 2), the upwelled  
 551 Fe flux can be estimated. Upwelled water from a 100–250 m grounding line would be  
 552 expected to contain 0.06–0.12 nM Fe (Marsay et al., 2017), compared to a mean of 33–680  
 553 nM in freshwater (Annett et al., 2017; Hodson et al., 2017). Upwelling via the same  $250 \text{ m}^3 \text{ s}^{-1}$   
 554 discharge as per the Arctic scenario, would generate a combined upwelled and discharge  
 555 flux (after estuarine removal processes) of 0.89–89 kmol Fe with 2–52% of the Fe arising  
 556 from upwelling and 48–98% from glacier discharge. Using an intermediate Fe:C value of 5  
 557  $\text{mmol Fe mol}^{-1} \text{ C}$ , which is broadly applicable to the coastal environment (Twining and  
 558 Baines, 2013), this would correspond to a biological pool of 0.019–1.9 Gmol C.

559 For a surface discharge of  $250 \text{ m}^3 \text{ s}^{-1}$ , nutrient entrainment is assumed to be negligible. In the  
 560 case of Fe outflow into a low-Fe, high- $\text{NO}_3$  system, we assume that the glacier outflow is the  
 561 dominant local Fe source over the fertilized area during the discharge period (i.e. changes to  
 562 other sources of Fe such as the diffusive flux from shelf sediments are negligible). For the  
 563 case of surface discharge into a low- $\text{NO}_3$ , high-Fe system, this is not likely to be the case for  
 564  $\text{NO}_3$ . Stratification induced by discharge decreases the vertical flux of  $\text{NO}_3$  from below, thus  
 565 negatively affecting  $\text{NO}_3$  supply, although there are to our knowledge no studies quantifying  
 566 this change in glacially-modified waters.

|   | Surface discharge                                   | Subglacial discharge                                   |
|---|---|--|
| <b>high-Fe, low-<math>\text{NO}_3</math> environment</b><br>(Predominant Arctic condition)    | <i>e.g. Young Sound</i><br><0–0.017 Gmol C          | <i>e.g. Bowdoin fjord, Sermilik</i><br>0.27–1.0 Gmol C |
| <b>low-Fe, high-<math>\text{NO}_3</math> environment</b><br>(Predominant Antarctic condition) | <i>e.g. Antarctic Peninsula</i><br>0.009–1.9 Gmol C | <i>e.g. Antarctic Peninsula</i><br>0.019–1.9 Gmol C    |

567 Table 4. Suppositional effect of different discharge scenarios calculated from the Redfield  
 568 ratio 106 C:16 N:1 P:0.005 Fe (Redfield, 1934; Twining and Baines, 2013). A steady  
 569 freshwater discharge of  $250 \text{ m}^3 \text{ s}^{-1}$  is either released from a land-terminating glacier or from a  
 570 marine-terminating glacier at 100–250 m depth, in both cases for two months into Fe-replete,  
 571  $\text{NO}_3$ -deficient; or Fe-deficient,  $\text{NO}_3$ -replete marine environments. Freshwater endmembers  
 572 are defined as  $2 \mu\text{M NO}_3$  and 33–675 nM dissolved Fe (Annett et al., 2017; Hodson et al.,  
 573 2017; Hopwood et al., 2018). Ambient water column conditions are defined as Greenland  
 574 (Achterberg et al., 2018) (i.e., high-Fe, low  $\text{NO}_3$ ) and Ross Sea (Marsay et al., 2017) (i.e.,  
 575 low-Fe, high- $\text{NO}_3$ ) shelf profiles.

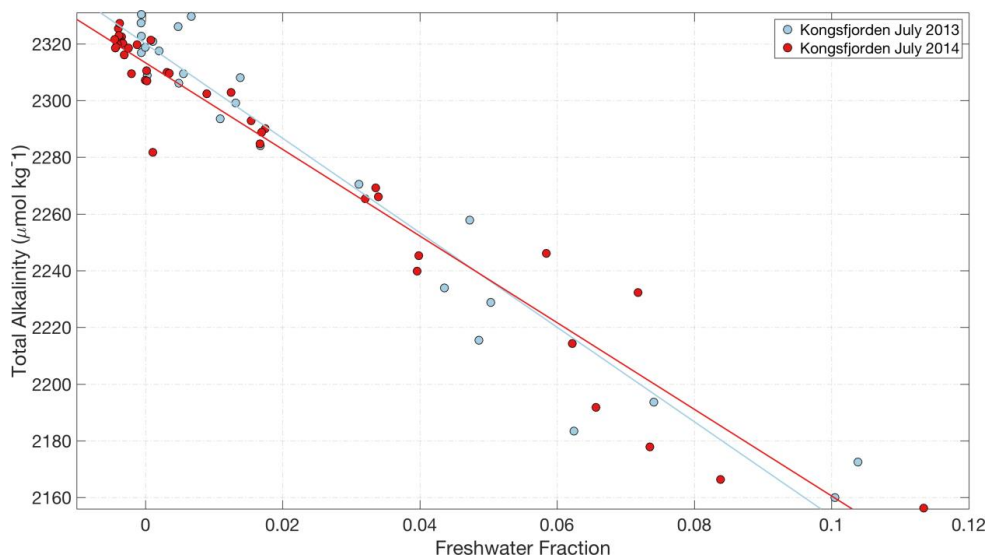
576 It is clear from these simplified discharge scenarios (Table 4) that both the depth at which  
 577 glacier discharge is released into the water column and the relative availabilities of  $\text{NO}_3$  and



578 Fe in downstream waters are critical for determining the response of primary producers. The  
579 response of primary producers in low-Fe regimes is notably subject to much larger  
580 uncertainty, mainly because of uncertainty in the extent of Fe removal during estuarine  
581 mixing (Schroth et al., 2014; Zhang et al., 2015). Whilst the effects of the marine-terminating  
582 glacier ‘nutrient pump’ on macronutrient fluxes have been defined in numerous systems, its  
583 effect on Fe availability is poorly constrained (Gerringa et al., 2012; St-Laurent et al., 2017,  
584 2019). Furthermore, Fe bioavailability is conceptually more complicated than discussed  
585 herein, as marine organisms at multiple trophic levels affect the speciation and bioavailability  
586 of Fe, and the transfer between less-labile and more-labile Fe pools in the marine  
587 environment (Poorvin et al., 2004; Vraspir and Butler, 2009; Gledhill and Buck, 2012). Many  
588 microbial species release organic ligands into solution which stabilizes dissolved Fe as  
589 organic complexes and these feedbacks are challenging to model (Strzepek et al., 2005), but  
590 may exert a cap on the lateral transfer of Fe away from glacier inputs (Lippiatt et al., 2010;  
591 Thuroczy et al., 2012). To date, Fe fluxes from glaciers into the ocean have primarily been  
592 constructed from an inorganic, freshwater perspective (Raiswell et al., 2006; Raiswell and  
593 Canfield, 2012; Hawkings et al., 2014). Yet to understand the net change in Fe availability to  
594 marine biota, a greater understanding of how ligands and estuarine mixing processes  
595 moderate the glacier-to-ocean Fe transfer will evidently be required (Lippiatt et al., 2010;  
596 Schroth et al., 2014; Zhang et al., 2015).

## 597 **6.0 Effects on the carbonate system**

598 Beyond its impact on inorganic nutrient dynamics, glacial discharge also affects the  
599 carbonate system in seawater. Discharge reduces the buffering capacity (total alkalinity) of  
600 glacially-modified waters (Fig. 7). This consequently impacts the saturation state of  
601 biologically-important calcium carbonate minerals (Doney et al., 2009; Fransson et al., 2015).  
602 Alkalinity measurements of glacial discharge across the Arctic reveal a range from 20–550  
603  $\mu\text{mol kg}^{-1}$  (Yde et al., 2005; Sejr et al., 2011; Rysgaard et al., 2012; Evans et al., 2014;  
604 Fransson et al., 2015, 2016; Meire et al., 2015; Turk et al., 2016). Similar to Si  
605 concentrations, the broad range is likely explained by different degrees of interaction between  
606 meltwater and bedrock, with higher alkalinity corresponding to greater discharge-bedrock  
607 interaction (Wadham et al., 2010; Ryu and Jacobson, 2012). The extent of alkalinity  
608 enrichment from bedrock also depends on the local geology. For drainage basins containing  
609 carbonate and silicate-rich bedrock, a higher alkalinity is expected (Yde et al., 2005; Fransson  
610 et al., 2015). However, in absolute terms even the upper end of the alkalinity range reported  
611 in glacial discharge is very low compared to the volume-weighted average of Arctic rivers,  
612  $1048 \mu\text{mol kg}^{-1}$  (Cooper et al., 2008).



613

614 Figure 7. Total alkalinity in Kongsfjorden during the meltwater season (data from Fransson et  
615 al., 2016). A decline in alkalinity is evident with increasing freshwater fraction in response to  
616 the low alkalinity concentrations in glacier discharge. Freshwater fraction was calculated  
617 using an average marine salinity endmember of 34.96, hence some slightly negative values  
618 are calculated in the outer fjord associated with the higher salinity of unmodified Atlantic  
619 water. Linear regression details are shown in Supplementary Table 1.

620 By decreasing the alkalinity of glacially-modified waters, glacier discharge reduces the  
621 aragonite and calcite ( $\Omega_{Ar}$  and  $\Omega_{Ca}$ , respectively) saturation states thereby amplifying the  
622 effect of ocean acidification (Fransson et al., 2015, 2016; Ericson et al., 2019). High primary  
623 production can mitigate this impact as photosynthetic  $CO_2$  uptake reduces dissolved  
624 inorganic carbon and  $pCO_2$  (e.g. Fig. 8) in surface waters and increases the calcium carbonate  
625 saturation state (Chierici and Fransson, 2009; Rysgaard et al., 2012; Meire et al., 2015). In  
626 relatively productive fjords, the negative effect of alkalinity dilution may therefore be  
627 counter-balanced. However, in systems where discharge-driven stratification is responsible  
628 for low productivity, the increased input of discharge may create a positive feedback on  
629 ocean acidification state in the coastal zone resulting in a lower saturation state of calcium  
630 carbonate (Chierici and Fransson, 2009; Ericson et al., 2019). Low-calcium carbonate  
631 saturation states ( $\Omega < 1$ ; i.e. corrosive conditions) have indeed been observed in the inner part  
632 of Glacier Bay (Alaska), demonstrating that glaciers can amplify seasonal differences in the  
633 carbonate system and negatively affect the viability of shell-forming marine organisms  
634 (Evans et al., 2014). Low  $\Omega_{Ar}$  has also been observed in the inner parts of Kongsfjorden,  
635 coinciding with high glacial discharge (Fransson et al., 2016). Such critically low  $\Omega_{Ar}$  ( $< 1.4$ )  
636 conditions have negative effects on aragonite-shell forming calcifiers such as the pteropod  
637 *Limacina helicina* (Comeau et al., 2009, 2010; Lischka et al., 2011; Lischka and Riebesell,  
638 2012; Bednaršek et al., 2014). Under future climate scenarios, in addition to the effect of  
639 increased glacier drainage in glacier fjords, synergistic effects with a combination of  
640 increased ocean  $CO_2$  uptake and warming will further amplify changes to the ocean  
641 acidification state (Fransson et al., 2016; Ericson et al., 2019), resulting in increasingly



642 pronounced negative effects on calcium carbonate shell formation (Lischka and Riebesell,  
643 2012).

#### 644 **7.0 Organic matter in glacial discharge**

645 In addition to inorganic ions, glacial discharge also contains many organic compounds  
646 derived from biological activity on glacier surfaces and overridden sediments (Barker et al.,  
647 2006; Lawson et al., 2014b). Organic carbon stimulates bacterial activity, and  
648 remineralization of organic matter is a pathway to resupply labile nitrogen and phosphorous  
649 to microbial communities. Similar to macronutrient concentrations, DOM concentrations in  
650 glacial discharge are generally low (Table 2) compared to runoff from large Arctic rivers,  
651 which have DOM concentrations 1–2 orders of magnitude higher (Dittmar and Kattner, 2003;  
652 Le Fouest et al., 2013). This is evidenced in Young Sound where DOC concentrations  
653 increase with salinity in surface waters, demonstrating that glaciers are a relatively minor  
654 source of DOM to the fjord (Paulsen et al., 2017).

655 While DOM concentrations are low in glacial discharge, the bioavailability of this DOM is  
656 much higher than its marine counterpart (Hood et al., 2009; Lawson et al., 2014b; Paulsen et  
657 al., 2017). This is likely due to the low C:N ratio of glacial DOM, as N-rich DOM of  
658 microbial origin is generally highly labile (Lawson et al., 2014a). It has been suggested that  
659 as glaciers retreat and the surrounding catchments become more vegetated, DOC  
660 concentrations in these catchments will increase (Hood and Berner, 2009; Csank et al., 2019).  
661 However, DOM from non-glacial terrestrial sources has a higher composition of aromatic  
662 compounds and thus is less labile (Hood and Berner, 2009; Csank et al., 2019). Furthermore,  
663 glacier coverage in watersheds is negatively correlated with DOC:DON ratios, so a reduction  
664 in the lability of DOM with less glacial coverage is also expected (Hood and Berner, 2009;  
665 Hood and Scott, 2008).

666 While DOC is sufficient to drive bacterial metabolism, bacteria also depend on nitrogen and  
667 phosphorus for growth. In this respect, bacteria are in direct competition with phytoplankton  
668 for macronutrients and increasing additions of labile DOM downstream of glaciers could give  
669 bacteria a competitive edge. This would have important ecological consequences for the  
670 function of the microbial food web and the biological carbon sink (Larsen et al., 2015).  
671 Experiments with Arctic fjord communities, including Kongsfjorden, have shown that when  
672 bacteria are supplied with additional subsidies of labile carbon under nitrate-limitation, they  
673 out-compete phytoplankton for nitrate (Thingstad et al., 2008; Larsen et al., 2015). This is  
674 even the case when there is an addition of excess Si, which might be hypothesized to give  
675 diatoms a competitive advantage. The implications of such competition for the carbon cycle  
676 are however complicated by mixotrophy (Ward and Follows, 2016; Stoecker et al., 2017). An  
677 increasing number of primary producers have been shown to be able to simultaneously  
678 exploit inorganic resources and living prey, combining autotrophy and phagotrophy in a  
679 single cell. Mixotrophy allows protists to sustain photosynthesis in waters that are severely  
680 nutrient-limited and provides an additional source of carbon as a supplement to  
681 photosynthesis. This double benefit decreases the dependence of primary producers on short-  
682 term inorganic nutrient availability. Moreover, mixotrophy promotes a shortened, and  
683 potentially more efficient, chain from nutrient regeneration to primary production (Mitra et  
684 al., 2014). Whilst mixotrophy is sparsely studied in Arctic fjords, both increasing  
685 temperatures and stratification are expected to favour mixotrophic species (Stoecker and





686 Lavrentyev, 2018), and thus an understanding of microbial foodweb dynamics is vital to  
687 predict the implications of increasing discharge on the carbon cycle in glacier-fjord systems.

688 Regardless of the high bioavailability of DOM from glacial discharge, once glacial DOM  
689 enters a fjord and is diluted by ocean waters, evidence of its uptake forming a significant  
690 component of the microbial food web in the Arctic has yet to be observed. Work from several  
691 outlet glacier fjords around Svalbard shows that the stable isotopic C ratio of bacteria does  
692 not match that of DOC originating from local glaciers, suggesting that the main source of  
693 carbon to the microbial loop even within inner-glaciated fjord environments is autochthonous  
694 and that glacially supplied DOC is a minor component of bacterial consumption (Holding et  
695 al., 2017; Paulsen et al., 2018). Whilst concentrations of DOM are low in glacier discharge,  
696 DOM sourced nitrogen and phosphorous could still be relatively important in stratified outlet  
697 glacier fjords simply because inorganic nutrient concentrations are also low. Refractory DON  
698 in rivers that is not directly degraded by bacteria is subsequently broken down by  
699 photoammonification processes releasing ammonium (Xie et al., 2012). In large Arctic rivers,  
700 this nitrogen supply is greater than that supplied from inorganic sources (Le Fouest et al.,  
701 2013). For glacier discharge, processing of refractory DOM could potentially produce a  
702 comparable nitrogen flux to inorganic sources (Table 2). Similarly, in environments where  
703 inorganic PO<sub>4</sub> concentrations are low, DOP is an important source of phosphorous for both  
704 bacteria and phytoplankton. Many freshwater and marine phytoplankton species are able to  
705 synthesize the enzyme alkaline phosphatase in order to efficiently utilize DOP (Hoppe, 2003;  
706 Štrojsová et al., 2005). In the context of stratified, low salinity inner-fjord environments,  
707 where inorganic PO<sub>4</sub> concentrations are potentially low enough to limit primary production  
708 (Prado-Fiedler, 2009), this process may be particularly important- yet is also understudied in  
709 glaciated catchments (Stibal et al., 2009).

710 Finally, whilst DOC concentrations in glacier discharge are low, particulate organic carbon  
711 (POC) concentrations, which may also impact microbial productivity in the marine  
712 environment, are less well characterized. Downstream of Leverett Glacier, mean runoff POC  
713 concentrations are reported to be 43–346 μM; 5 times higher than DOC (Lawson et al.,  
714 2014b). However, the opposite is reported for Young Sound, where DOC concentrations in  
715 three glacier-fed streams were found to be 7–13 times higher than POC concentrations  
716 (Paulsen et al., 2017). Similarly, low POC concentrations of only 5 μM were found in  
717 supraglacial discharge at Bowdoin glacier (Kanna et al., 2018). In summary, relatively little is  
718 presently known about the distribution, fate, and bioavailability of POC in glaciated  
719 catchments.

## 720 **8.0 A link between retreating glaciers and harmful algal blooms?**

721 Shifts between different microbial groups in the ocean can have profound implications for  
722 ecosystem services. For example, addition of DOM can induce shifts in the microbial loop to  
723 favour bacteria in their competition with phytoplankton for macronutrient resources which  
724 directly affects the magnitude of CO<sub>2</sub> uptake by primary producers (Thingstad et al., 2008;  
725 Larsen et al., 2015). Similarly, changing the availability of Si relative to other macronutrients  
726 affects the viability of diatom growth and thus, due to the efficiency with which diatom  
727 frustules sink, potentially the efficiency of the biological carbon pump (Honjo and  
728 Manganini, 1993; Dugdale et al., 1995). A particularly concerning hypothesis, recently



729 proposed from extensive work in Patagonian glacier-fjord systems, is that an increase in  
730 harmful algal bloom (HAB) intensity is related to glacier retreat (León-Muñoz et al., 2018).

731 In southern Patagonia, particularly around the Straits of Magellan, most glaciers have  
732 experienced varying degrees of retreat in recent decades (Rivera et al., 2012). The combined  
733 reasons for this regional pattern, including atmospheric warming and reduced precipitation,  
734 are projected to continue. The seasonal cycle of phytoplankton in these waters is well  
735 characterized: the main phytoplankton blooms occur in austral spring-summer-fall along the  
736 coastal periphery, where high rates of primary production are supported by high near-surface  
737 light levels, and high nutrient availability. Across the Patagonian region, the spring bloom is  
738 typically diatom dominated, with diatoms constituting ~80–90% of phytoplankton (by carbon  
739 biomass). The initial dominance of diatoms is followed by a seasonal maximum of thecate  
740 dinoflagellates, and sporadically high biomass of phytoflagellates during summer (Iriarte et  
741 al., 2007; González et al., 2010). Such a seasonal progression in Patagonia is similar to that  
742 reported in some Arctic systems. In Kongsfjorden, for example, the spring bloom is similarly  
743 dominated by diatoms followed by dinoflagellate and flagellate dominance in the inner-fjord  
744 during summer (Hop et al., 2002).

745 Similar to Arctic systems, glacier discharge in Patagonia is invariably associated with fjord-  
746 scale water column stability; this confines phytoplankton to a favourable light regime in  
747 spring, but then proceeds to negatively affect primary production through summer by limiting  
748 the input of new nutrients across the pycnocline and increasing turbidity from runoff-derived  
749 particles (Iriarte et al., 2014). However, unlike most Arctic catchments, in recent decades  
750 glacier discharge in Patagonia has been declining (Bliss et al., 2014), largely due to reduced  
751 precipitation. Therefore summertime stratification is weakening in many regions, and also  
752 increasingly driven via surface heating rather than discharge (Lara et al., 2008; Rebolledo et  
753 al., 2011; León-Muñoz et al., 2013).

754 HAB events in the region spanning 50–56°S correspond hydrographically to outlet glacier  
755 fjord systems and thus are hypothesized to arise as a result of decreasing discharge, which  
756 facilitates weaker summertime stratification (Iriarte et al., 2014). Dinoflagellate species like  
757 *Alexandrium catenella*, associated with historically recurrent toxic outbreaks in the inner seas  
758 of Patagonian, have progressively expanded their spatial distribution in the last four decades  
759 (León-Muñoz et al., 2018). This is a major concern for fisheries in a region where  
760 aquaculture is of growing economic importance (Apablaza et al., 2017; Méndez and Carreto,  
761 2018). Recently, other diatom and dinoflagellate species of concern have also been detected  
762 in this region such as *Pseudo-nitzschia delicatissima* and *Alexandrium ostenfeldii* (León-  
763 Muñoz et al., 2018). Given the ongoing intensification of climate change and the interacting  
764 effects of different environmental drivers of primary production in glacier-fjord systems (e.g.  
765 surface warming, carbonate chemistry, light availability, stratification, nutrient availability,  
766 zooplankton distribution, etc.), it is challenging to predict the impact of future changes on  
767 HAB event intensity. Furthermore, different HAB associated groups (e.g. toxin-producing  
768 diatom and flagellate species) may show opposite responses to the same environmental  
769 perturbation (Wells et al., 2015). Moreover, many known toxin-producing species in the  
770 Arctic are mixotrophic, further complicating their interactions with other microbial groups  
771 (Stoecker and Lavrentyev, 2018). Whilst HAB associated species are known to be present in  
772 Arctic waters (Lefebvre et al., 2016; Richlen et al., 2016) and HAB events appear to be  
773 associated with glacier retreat in Patagonia (León-Muñoz et al., 2018), there are fundamental



774 knowledge gaps concerning the mechanisms of HAB development. Given the socio-  
775 economic importance of glacier-fjord scale subsistence fisheries, especially around  
776 Greenland, a clear priority for future research in the Arctic, is therefore to establish to what  
777 extent HAB associated species are likely to benefit from future climate scenarios (Richlen et  
778 al., 2016).

### 779 **9.0 Insights into the long-term effects of glacier-retreat**

780 Much of the present interest in Arctic ice-ocean interactions arises because of the accelerating  
781 increase in discharge from the Greenland Ice Sheet, captured by multi-annual to multi-  
782 decadal time-series. This trend is attributed to atmospheric and oceanic warming due to  
783 anthropogenic forcing, at times enhanced by persistent shifts in atmospheric circulation (Box,  
784 2002; Ahlström et al. 2017). From existing observations, it is clear that strong climate  
785 variability patterns are at play, such as the North Atlantic Oscillation/Arctic Oscillation, and  
786 that in order to place recent change in context, time series exceeding the satellite era are  
787 required. Insight can be potentially gained from research into past sedimentary records of  
788 productivity from high-latitude marine and fjord environments. Records of productivity and  
789 the dominance of different taxa as inferred by microfossils, biogeochemical proxies, and  
790 genetic records from those species that preserve well in sediment cores can help establish  
791 long-term spatial and temporal patterns around the present-day ice sheet periphery (Ribeiro et  
792 al., 2012). Around Greenland, sediment cores largely corroborate recent fjord-scale surveys  
793 suggesting that inner-fjord water column environments are generally low productivity  
794 systems with protist taxonomic diversity and overall productivity normally higher in shelf  
795 waters than in inner-fjord environments (Ribeiro et al., 2017).

796 Several paleoclimate archives and numerical simulations suggest that the Arctic was warmer  
797 than today during the early to mid-Holocene thermal maximum, which was registered by ~1  
798 km thinning of the Greenland Ice Sheet (Lecavalier et al., 2017). Multiproxy analyses  
799 performed on high-resolution and well-dated Holocene marine sediment records from  
800 contrasting fjord systems are therefore one approach to understand the nature of such past  
801 events, as these sediments simultaneously record climate and some long-term biotic changes  
802 representing a unique “window” into the past. However, while glacial-interglacial changes  
803 can provide insights into large scale ice-ocean interactions and the long-term impact of  
804 glaciers on primary production, these time-scales are of limited use to understanding more  
805 recent variability at the ice-ocean interface of fjord systems such as those mentioned in this  
806 review. The five well-characterised Arctic fjords used as case studies here (Bowdoin,  
807 Kongsfjorden, Sermilik, Godthåbsfjord and Young Sound), for example, did not exist during  
808 the Last Glacial Maximum ~19000 years ago (Knutz et al., 2011).

809 On long timescales, glacier-ocean interactions are subject to marked temporal changes  
810 associated with glacial/interglacial cycles. In the short-term, the position of glacier termini  
811 shifts inland during ice sheet retreat, or outwards during ice sheet expansion, and in the long-  
812 term proglacial regions respond to isostatic uplift and delta progradation. The uplift of fine-  
813 grained glaciomarine and deltaic sediments is a notable feature of landscape development in  
814 fjord environments following the retreat of continental-scale ice sheets (Cable et al., 2018;  
815 Gilbert et al., 2018). This results in the gradual exposure and subsequent erosion of these  
816 sediment infills and their upstream floodplains, releasing labile organic matter to coastal  
817 ecosystems. Whilst the direct biogeochemical significance of such chemical fluxes may be



818 limited in the marine environment (Table 2), potentially more important is the Fe fertilisation  
819 following wind erosion and dust emittance from glacial floodplains. Ice core records from  
820 Greenland and Antarctica, spanning several climatic cycles, suggest that aeolian deposition  
821 rates at high latitudes were as much as 20 times greater during glacial than interglacial  
822 periods (Kohfeld and Harrison, 2001). Elevated input of terrigenous Fe during windy glacial  
823 episodes, and associated continental drying, has therefore been hypothesized to stimulate  
824 oceanic productivity through time and thus modify the oceanic and atmospheric CO<sub>2</sub> balance  
825 (Martin, 1990). While there seems to be a pervasive dust-climate feedback on a glacial-  
826 interglacial planetary scale (Shaffer and Lambert, 2018), glacier retreat also exposes new  
827 areas of unconsolidated glacial sediments leading to an increase in both dust storm events and  
828 sediment yields from glacial basins locally (Crusius et al., 2011; Prospero et al., 2012;  
829 Bullard, 2013). The spatial scale over which glacially derived dust can be transported (100-  
830 500 km) far exceeds that of discharge-carried nutrients.

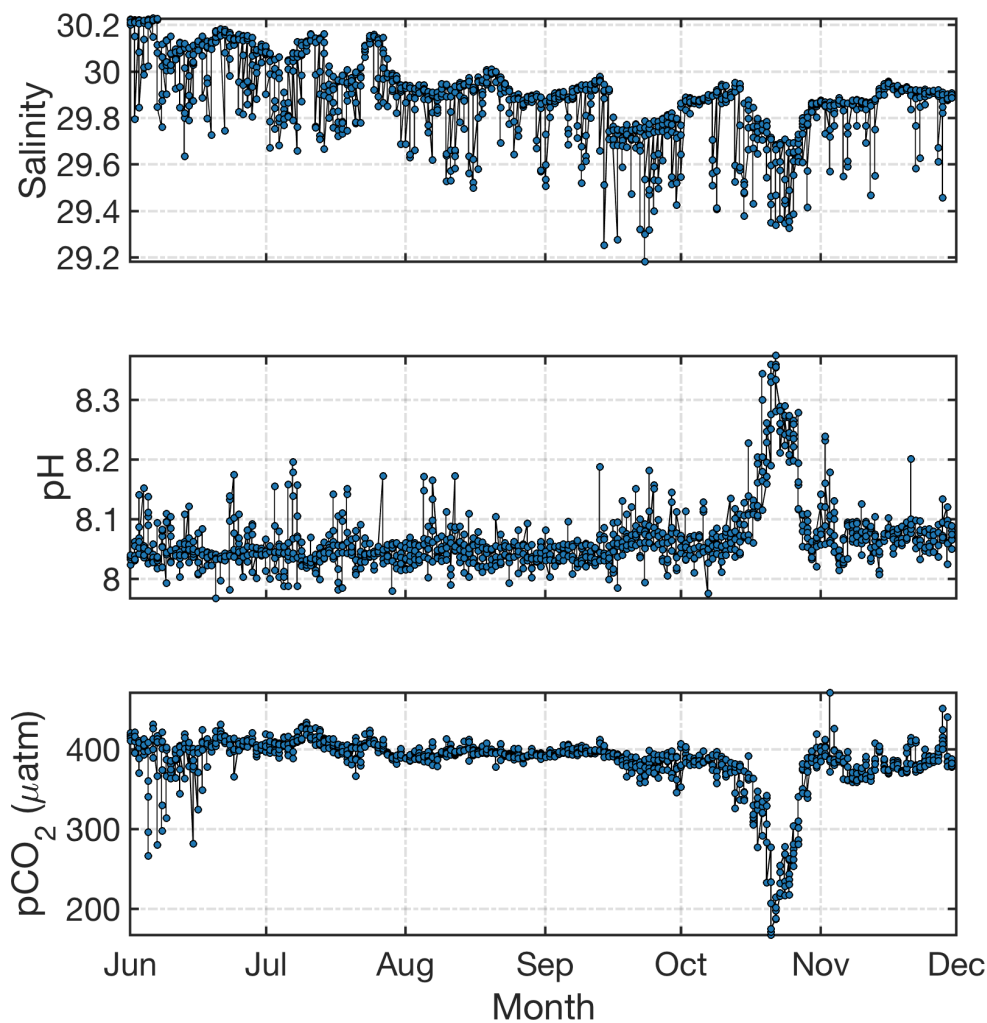
#### 831 **10.0 A need for new approaches?**

832 The pronounced temporal and spatial variations evident in the properties of glacier discharge  
833 and glacially-modified waters emphasize the need for high-resolution data on both short  
834 (hourly-daily) and long (seasonal to interannual) timescales in order to understand glacial  
835 processes and their downstream effects. In Godthåbsfjord, Juul Pedersen et al., (2015) provide  
836 a detailed study of seasonal primary production dynamics. This monthly monitoring  
837 programme captures seasonal, annual and interannual trends in the magnitude of primary  
838 production. Whilst such a timeseries does clearly highlight a strong interannual stability in  
839 both seasonal and annual primary production ( $103.7 \pm 17.8 \text{ g C m}^{-2} \text{ yr}^{-1}$ ; Juul-Pedersen et al.,  
840 2015), it is unable to fully characterise shorter (i.e. days-weeks) timescale events such as the  
841 spring bloom period. Yet higher data resolution cannot feasibly be sustained by shipboard  
842 campaigns.

843 Low-frequency, high-discharge events are known to occur in Godthåbsfjord, and other  
844 glacier fjords (Kjeldsen et al., 2014), but are challenging to observe from monthly-resolution  
845 data and thus there is sparse data available to quantify their occurrence and effects.  
846 Consequently, modelled subglacial discharge rates and glacier discharge derived from  
847 regional models (e.g. RACMO, Noël et al., 2015), which underpin our best-available  
848 estimates of the subglacial ‘nutrient pump’ (e.g. Carroll et al., 2016), do not yet consider such  
849 events. Time lapse imagery shows that the lifetimes and spatial extents of subglacial  
850 discharge plumes can vary considerably (Schild et al., 2016). While buoyant plume theory  
851 has offered important insights into the role of subglacial plumes in the ‘nutrient pump’,  
852 buoyant plume theory does not characterise the lateral expansion of plume waters.  
853 Furthermore, determining the influence of discharge, beyond the immediate vicinity of  
854 glacial outflows, is a Lagrangian exercise, yet the majority of existing observational and  
855 modelling studies have been conducted primarily in the Eulerian reference frame (e.g., ship-  
856 based profiles and moored observations that describe the water column at a fixed location).  
857 Moving towards an observational Lagrangian framework will require the deployment of new  
858 technology such as the recent development of low-cost GPS trackers which, especially when  
859 combined with *in situ* sensors, may improve our understanding of the transport and mixing of  
860 heat, freshwater, sediments, and nutrients downstream of glaciers (Carlson et al., 2017;  
861 Carlson and Rysgaard, 2018). For example, GPS trackers deployed on ‘berg bits’ have



862 revealed evidence of small-scale, retentive eddies in Godthåbsfjord (Carlson et al., 2017) and  
863 established the surface flow variability in Sermilik Fjord (Sutherland et al., 2014).  
864 Unmanned aerial vehicles and autonomous surface/underwater vehicles can also be used to  
865 observe the spatiotemporal variability of subglacial plumes at high resolution (Mankoff et al.,  
866 2016; Jouvét et al., 2018). Complementing these approaches are developments in the rapidly-  
867 maturing field of miniaturized chemical sensors suitable for use in cryosphere environments  
868 (Beaton et al., 2012). Such technology will ultimately reduce much of the uncertainty  
869 associated with glacier-ocean interactions by facilitating more comprehensive, more  
870 sustainable field campaigns (Straneo et al., 2019), with reduced costs and environmental  
871 footprints (Nightingale et al., 2015; Grand et al., 2017, 2019). This is evidenced by a  
872 successful prolonged mooring deployment in the Santa Inés glacier-fjord system (Fig. 8).



873

874 Figure 8. Winter-spring dynamics of salinity, pH and  $p\text{CO}_2$  at the Santa Inés Glacier-fjord,  
875 Ballena (Patagonia). High-resolution  $p\text{CO}_2$  and pH measurements (every three hours) were

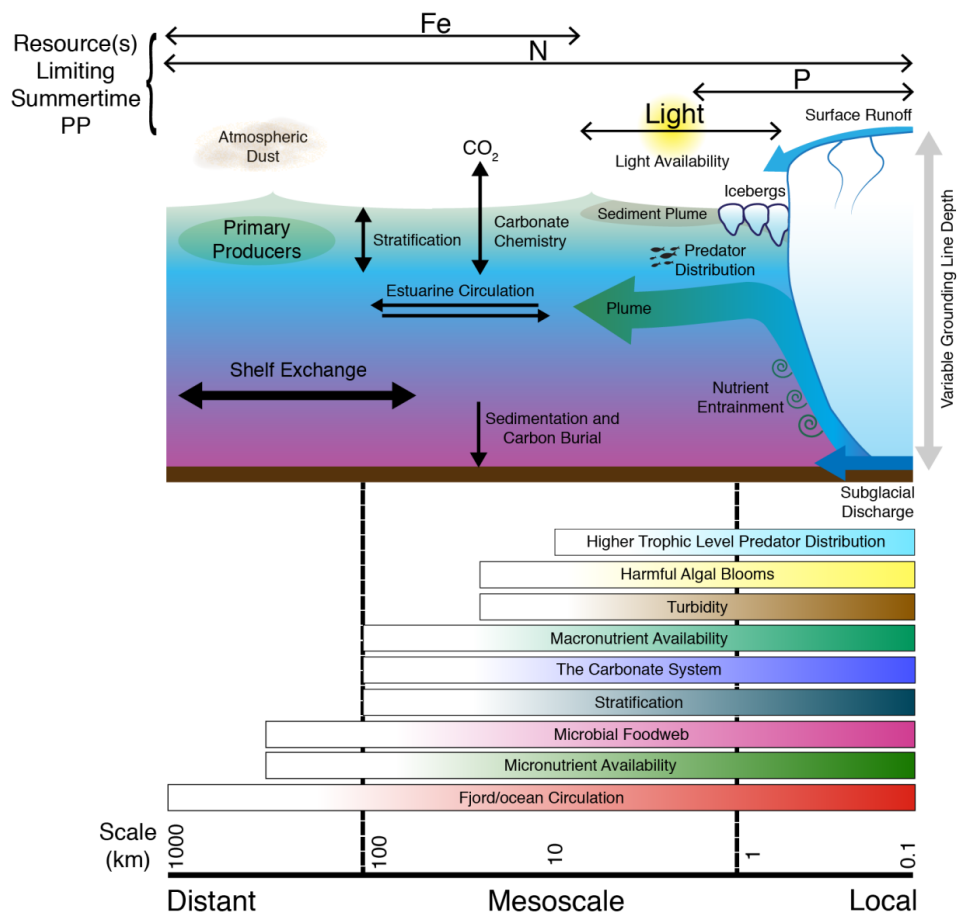


876 taken *in situ* using autonomous SAMI-CO<sub>2</sub> and SAMI-pH sensors (Sunburst Sensors, LLC)  
877 starting in the austral autumn (March 2018). All sensors were moored at 10 m depth.

878 The Santa Inés Glacier-fjord sits adjacent to the open water of the Straits of Magellan in  
879 southwest Patagonia. Moored high resolution measurements are now collected *in situ* using  
880 sensor technology and a mooring within the fjord. Measurements include the carbonate  
881 system parameters *p*CO<sub>2</sub> and pH. The 2018 winter to spring timeseries (Fig. 8) demonstrates  
882 a sharp decline in *p*CO<sub>2</sub>, and corresponding increase in pH, associated with the onset of the  
883 spring bloom in early October. Such a pronounced event, occurring over ~2 weeks would be  
884 impossible to characterise fully with monthly sampling of the fjord. Over winter, pH and *p*CO<sub>2</sub>  
885 were more stable, but sensor salinity data still reveals short-term dynamics within the fjords'  
886 surface waters (Fig. 8). A general decline in salinity is evident moving from winter into  
887 spring. Short-term changes on diurnal timescales-presumably linked to tidal forcing-and also  
888 on day-weekly timescales- possibly linked to weather patterns are also evident (Fig. 8). Much  
889 work remains to be done to deduce the role of these short-term drivers on primary production.



890 **11.0 Understanding the role of glaciers alongside other manifestations of climate change**



891

892 Figure 9. The approximate spatial scale over which glaciers directly affect different drivers of  
 893 marine primary production (PP) compared to the likely limiting resources constraining  
 894 primary production.

895 In order to comprehensively address the questions posed in this review, it is evident that a  
 896 broader perspective than a narrow focus on freshwater discharge alone is required.  
 897 Freshwater discharge is not the sole biogeochemical connection between the glaciers and the  
 898 ocean. Dust plumes from pro-glacial terrain supply glacial flour to the ocean on scales of  
 899 >100 km and thus act as an important source of Fe to the ocean at high latitudes, where other  
 900 atmospheric dust sources are scarce (Prospero et al., 2012; Bullard, 2013). Similarly, icebergs  
 901 have long been speculated to act as an important source of Fe to the offshore ocean (Hart,  
 902 1934; Raiswell et al., 2008; Lin et al., 2011) and induce mixing of the surface ocean (Helly et  
 903 al., 2011; Carlson et al., 2017). Whilst freshwater discharge is a driver of biogeochemical  
 904 changes in nearshore and fjord environments downstream of glaciers, the distant (>100 km  
 905 scale) biogeochemical effects of glaciers on the marine environment, are likely dominated by  
 906 these alternative mechanisms (Fig. 9).



907 Discharge derived effects must also be interpreted in the context of other controls on primary  
908 production in the high latitude marine environment. Sea-ice properties, and particularly the  
909 timing of its breakup and the duration of the ice-free season, are a key constraint on the  
910 seasonal trend in primary production in the Arctic (Rysgaard et al., 1999; Rysgaard and Glud,  
911 2007). Similarly, whilst discharge affects multiple aspects of the three-dimensional water  
912 column including fjord-scale circulation and mixing (Kjeldsen et al., 2014; Carroll et al.,  
913 2017), stratification (Meire et al., 2016b; Oliver et al., 2018) and boundary current properties  
914 (Sutherland et al., 2009); other changes in the Earth system from wind patterns (Spall et al.,  
915 2017; Sundfjord et al., 2017; Le Bras et al., 2018), sea-ice dynamics and regional temperature  
916 changes (Cook et al., 2016) are driving changes in these parameters on similar spatial and  
917 temporal scales (Stocker et al., 2013; Hop et al., 2019).

918 Several key uncertainties remain in constraining the role of glaciers in the marine  
919 biogeochemical system. Outlet glacier fjords are challenging environments in which to gather  
920 data and there is a persistent deficiency of both physical and biogeochemical data within  
921 kilometres of large marine-terminating glacier systems, where glacier discharge first mixes  
922 with ocean properties. Subglacial discharge plume modelling and available data from further  
923 downstream can to some extent evade this deficiency for conservative physical (e.g. salinity,  
924 temperature) and chemical (e.g. noble gases, NO<sub>3</sub>, PO<sub>4</sub>) parameters in order to understand  
925 mixing processes (Mortensen et al., 2014; Carroll et al., 2017; Beird et al., 2018). However,  
926 the mixing behaviour of non-conservative chemical parameters (pH, Si, Fe) is more  
927 challenging to deduce from idealized models. Furthermore, the biogeochemical effects of  
928 low-frequency, high-discharge events and small-scale mixing, such as that induced around  
929 icebergs, remain largely unknown. There is a critical need to address this deficiency by the  
930 deployment of new technology to study marine-terminating glacier mixing zones and  
931 downstream environments.

932 The uniqueness of glacier-fjord systems, due to highly variable fjord circulation and  
933 geometry, is itself a formidable challenge in ‘scaling-up’ results from Arctic field studies to  
934 produce a process-based understanding of glacier-ocean interactions. A proposed solution,  
935 which works equally well for physical, chemical and biological perspectives, is to focus  
936 intensively on a select number of key field sites at the land-ocean interface rather than mainly  
937 on large numbers of broad-scale, summertime-only surveys (Straneo et al., 2019). In addition  
938 to facilitating long-term time series, focusing in detail on fewer systems facilitates greater  
939 seasonal coverage to understand the changes in circulation and productivity that occur before,  
940 during, and after the melt season. However, the driving rationale for the selection of ‘key’  
941 glacier fieldsites to date was in many cases mainly their contribution to sea-level rise. Thus,  
942 well-studied sites account for a large fraction of total Arctic glacier discharge into the ocean,  
943 but only represent a small fraction of the glaciated coastline. For example, around the  
944 Greenland coastline, the properties of over 200 marine-terminating glaciers are characterized  
945 (Morlighem et al., 2017). Yet just 5 glaciers (including Helheim in Sermilik Fjord) account  
946 for 30% of annual combined meltwater and ice discharge from Greenland, and 15 account for  
947 >50% (year 2000 data, Enderlin et al., 2014). Whilst small glaciated catchments, such as  
948 Kongsfjorden and Young Sound, are far less important for sea-level rise, similar ‘small’  
949 glaciers occupy a far larger fraction of the high latitude coastline and are thus more  
950 representative of glaciated coastline habitat.

951





952 **12.0 Conclusions**

953 **Where and when does glacial freshwater discharge promote or reduce marine primary**  
954 **production?**

- 955 • In the Arctic, marine-terminating glaciers are associated with the enhanced vertical  
956 fluxes of macronutrients, which can drive summertime phytoplankton blooms  
957 throughout the meltwater season.
- 958 • In the Arctic, land-terminating glaciers are generally associated with the local  
959 suppression of primary production, due to light limitation and stratification impeding  
960 vertical nutrient supply from mixing.
- 961 • In contrast to the Arctic, input of Fe from glaciers around the Southern Ocean is  
962 anticipated to have a positive effect on marine primary production, due to the  
963 extensive limitation of primary production by Fe.
- 964 • In some brackish, inshore waters, DOM from glaciated catchments could enhance  
965 bacterial activity at the expense of primary production, but a widespread effect is  
966 unlikely due to the low DOM concentration in freshwater.
- 967 • Glacier discharge reduces the buffering capacity of glacially modified waters and  
968 amplifies the negative effects of ocean acidification, especially in low productivity  
969 systems, which negatively effects calcifying organisms.

970 **How does spatiotemporal variability in glacial discharge affect marine primary**  
971 **production?**

- 972 • Glacier retreat associated with a transition from marine- to land- terminating systems  
973 is expected to negatively affect downstream productivity in the Arctic, with long-term  
974 inland retreat also changing the biogeochemical composition of freshwater.
- 975 • Low-frequency, high-discharge events are speculated to be important drivers of  
976 physical and biogeochemical processes in the marine environment, but their  
977 occurrence and effects are poorly constrained.
- 978 • Declines in discharge volumes in Patagonia have been associated with an expansion  
979 of Harmful Algal Blooms in glacier fjord systems. Whether such a pattern will occur  
980 in future Arctic glacier fjords, where peak regional discharge is yet to occur, is  
981 unknown.
- 982 • A time series in Godthåbsfjord suggests that on inter-annual timescales, fjord-scale  
983 primary production is relatively stable despite sustained increases in glacier discharge.

984 **How far reaching are the effects of glacial discharge on marine biogeochemistry?**

- 985 • Local effects of glaciers (km/fjord scale) include light suppression, impediment of  
986 filter-feeding organisms, and influencing the foraging habits of higher organisms.
- 987 • Mesoscale effects of glaciers (10–100s km) include nutrient upwelling, Fe enrichment  
988 of seawater, modification of the carbonate system (both by physical and biological  
989 drivers), and enhanced stratification.



- 990       • Remote effects are less certain, beyond the 10–100 km scale over which discharge  
991       plumes can be evident, other mechanisms of material transfer between glaciers and  
992       the ocean, such as atmospheric deposition of glacial flour and icebergs are likely more  
993       important (Fig. 9).

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