

Supplement of Atmos. Chem. Phys., 20, 6921–6951, 2020
<https://doi.org/10.5194/acp-20-6921-2020-supplement>
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Supplement of

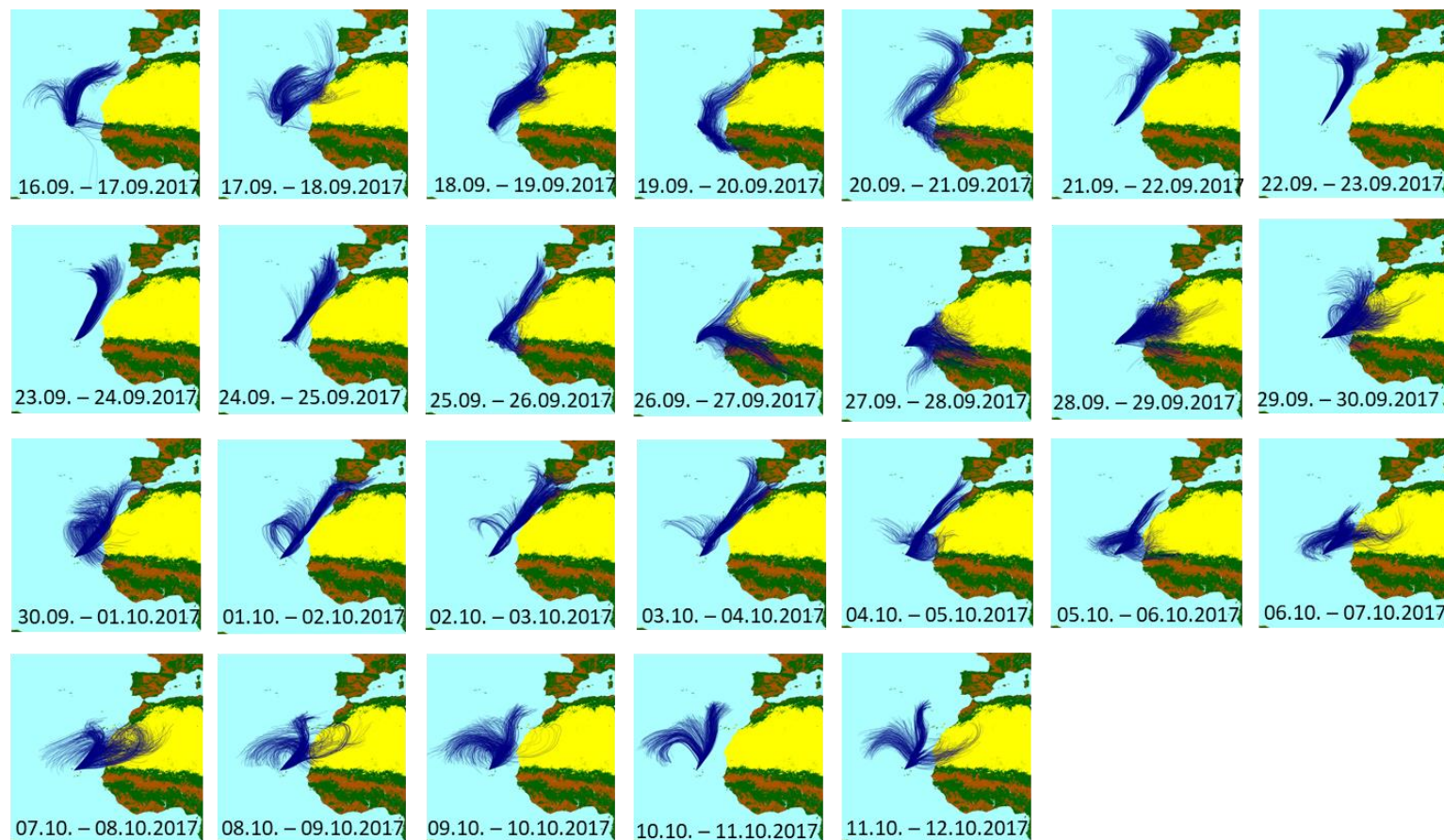
Marine organic matter in the remote environment of the Cape Verde islands – an introduction and overview to the MarParCloud campaign

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4 Fig. S1: 96 hour back trajectories calculated on an hourly basis within the intervals of the aerosol particle filter sampling at the CVAO, using the
 5 NOAA HYSPLIT model (HYbrid Single-Particle Lagrangian Integrated Trajectory, <http://www.arl.noaa.gov/ready/hysplit4.html>, 26.07.19) in the
 6 ensemble mode at an arrival height of $500 \text{ m} \pm 200 \text{ m}$ ((van Pinxteren, et al. 2010)). Starting time of the trajectories corresponded with the sampling
 7 time of the aerosol particles and was 21.00 UTC from 16th – 21st September and 16.00 UTC from 22nd September – 12th October.

8 **Helikite**

9 The measurements were done next to the CVAO using a Helikite (Allsopp Helikites Ltd,
10 Hampshire, UK). A Helikite is a unique combination of a tethered balloon and a kite. Helikites
11 are designed to be operated under extreme weather conditions. They fly stable even in strong
12 winds and due to the kite wing additional lift is generated in windy conditions. For the
13 measurements at the CVAO a 16 m³ Skyhook Helikite was used. The kite was attached to a 3
14 mm Dyneema rope (2000m long, ~ 4,6kg/1000m, Lyros D-Pro 3mm, breaking load 950 daN,
15 working elongation < 1%) and operated by a winch (TROPOS built). Under calm conditions
16 the Helikite has a net load capacity of ~ 8kg. At windy conditions the pull increases significantly
17 and reaches about 16 kg at 6 m s⁻¹. Depending on the prevailing conditions, measurements up
18 to an altitude of about 1000 m could be carried out. The payload with meteorological sensors
19 was attached to rope about 20 m below the helikite. The payload here was a measuring system
20 for standard meteorological parameters (p, T, rH, wind direction and velocity). The device is
21 based on a microcontroller-based data logger (TROPOS). The sensors were digital sensors,
22 tested and selected in the TROPOS wind tunnel (LACIS-T). Wind speed was measured using
23 a differential pressure sensor together with a pitot tube, wind direction was determined from an
24 orientation sensor (compass) of the wind vane. Data were recorded with a measuring frequency
25 of 2 Hz, stored on SD card and additionally transmitted to a ground station (via XBee). In total
26 19 flights measured at 10 days are available.

27

28 **Meteorological parameters and trace gases**

29 Temperature, relative humidity and wind measurements were measured using a meteorological
30 station fitted with various sensors (Campbell Scientific Ltd, UK).

31 Ozone was measured using a UV absorption instrument (Model 49i Thermo Scientific).

32 VOCs and OVOCs were measured using a dual channel gas chromatograph with flame
33 ionization detection (Agilent 7890-A). The instrument has two parallel columns, which
34 simultaneously resolve C₂- C₈ NMHCs and o-VOCS methanol and acetone using a more polar
35 LOWOX column. VOCs are pre-concentrated onto a multi-adsorbent bed at -30 °C and then
36 rapidly desorbed at 350 °C into helium flow using a desorption unit (Markes International
37 Unity2). NMHCs are calibrated monthly using a multicomponent hydrocarbon standard
38 (National Physics Laboratory, UK, typical concentrations ~ 5 ppbV) whilst for OVOCs a
39 permeation system is used for calibration at levels of 8 to 25 ppbv in conjunction with relative
40 detector response. Weighings of the OVOC permeation tubes are typically carried out every 6
41 weeks. The accuracy of the OVOC calibration is estimated as 10% for methanol and 5% for
42 acetone. DMS was measured using a gas chromatograph with mass spectrometric detection
43 (Agilent 7890-A, 5977 MSD). A Unity2 (Markes International) was used for the
44 preconcentration of DMS onto a Tenax trap at -30° C which was desorbed at 350 °C into a flow
45 of helium onto the GC. DMS was calibrated every ten hours during the campaign using a
46 working standard manufactured at the University of York and quantified using relative detector
47 response to benzene (10 ppm DMS standard, Korea Research Institute of Standards and
48 Science, Republic of Korea (KRISS)). All trace gas and meteorological measurements were
49 made from a height of 17.5 m asl.

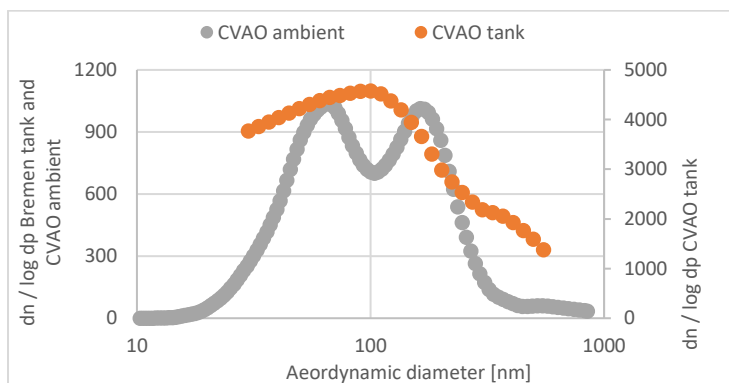
50

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52 **Plunging waterfall tank at the CVAO**

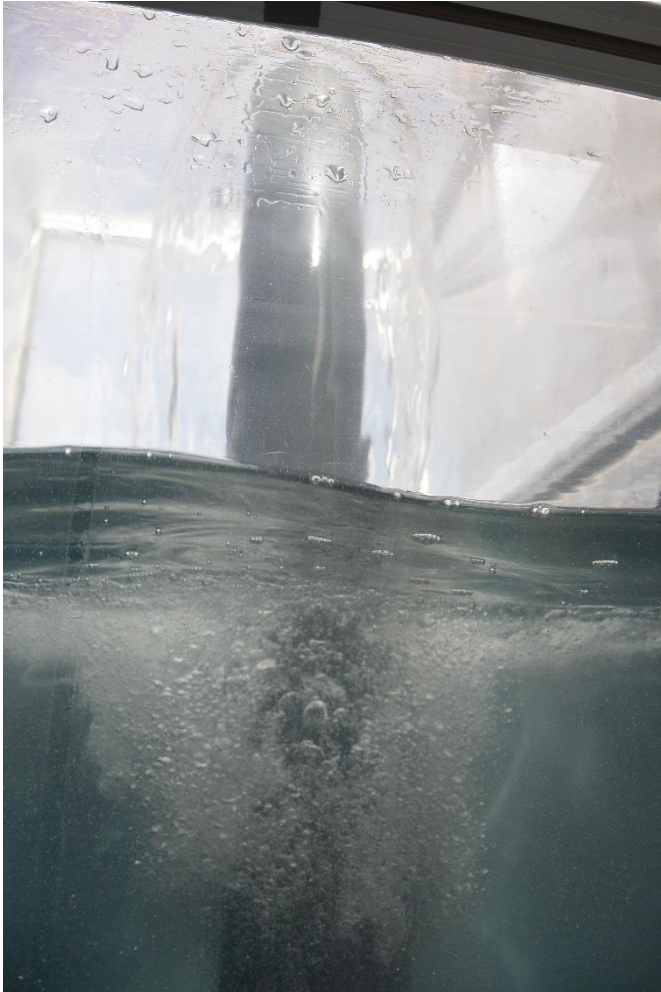
53 The tank was designed to study the bubble driven transfer of organic material from the bulk
54 water via the SML into the aerosol phase. It consisted of a 1400 L basin with a 500 L aerosol
55 chamber on top. The basin (120 cm length x 110 cm width x 100 cm height) was made of 10
56 mm polyvinyl chloride (PVC) plates, held together by an aluminum frame. One side was
57 transparent allowing for visual inspection. To minimize contamination from the tank walls, the
58 entire basin was lined with a Teflon FEP bag. Inlet and sampling ports were made from
59 Swagelok stainless steel fittings. PVC in contact with seawater was rinsed with artificial
60 seawater for at least two weeks prior to use inside the basin. The head air space was made of
61 Teflon FEP and had a total volume of 500 L (60 cm length, 110 cm width, 60 cm height).
62 Sampling ports for bulk water and the SML were located on top of the basin. Bulk water
63 samples were taken from 50 cm above the bottom and sampled into 1 L Duran glass bottles
64 through a Teflon PFA tube using the hydrostatic pressure. Prior to each sampling, the tubing
65 was first flushed with 100 mL of bulk water. SML samples were taken with a boron silicate
66 glass plate (4 mm, 15 x 60 cm) via a slit in the enclosure. The SML sampling volume was
67 limited to about 55 mL, corresponding to SML loss by about 50% when assuming SML layer
68 thickness to be 60 to 80 μm (Falkowska 1999). The glass plate was cleaned with ultrapure water
69 and ethanol prior to each sampling. Bubble driven transport of organic material was simulated
70 using a skimmer on a plunging waterfall. The plunging waterfall (Fig. SI3) was made from a
71 tubular pump (Osaga ORP 25000) that was placed at the bottom of the tank and a PVC tube of
72 100 cm length and 10 cm inner diameter. The tube towered 40 cm above the water surface.
73 Water was pumped at a flow rate of approximately 200 L min^{-1} through the tube. The large
74 diameter of the tubular pump and the tube ensured resulted in small pressure fluctuations and
75 hence prevent the pelagic phytoplankton and microbial community from damage. The falling
76 seawater simulated the process of wave breaking and was assumed to mimic the natural size
77 spectra of rising air bubbles. A stainless steel inlet was inserted in the headspace of the tank and
78 connected with three filter holders for offline aerosol particle sampling without size segregation
79 (TSP). In addition, the stainless steel inlet was connected to a SMPS (same type as used for
80 ambient aerosol particle characterisation) for online aerosol measurements. This method of
81 aerosol generation resulted in a very efficient generation of nascent sea salt aerosols with
82 aerosol particle size spectra centred around 100 nm as shown in Figure S2.

83



84

85 Fig. S2: Size distribution of the aerosols generated in the tank experiments in comparison to
86 the size distribution of ambient aerosols at the CVAO.



87

88 Fig. S3: *View of the plunging waterfall.*

89

90 **The MarParCat**

91 The MarParCat is a remotely controllable catamaran designed as a platform for water and SML
92 sampling. It has a size of 245 x 180 x 140(l, w, h) and a total weight of 125 kg. The payload is
93 approximately 160 kg. The catamaran is made from two Technus PE-floats (Type 90/150, 245
94 x 31 x 38 cm) hold together by an aluminum frame. The catamaran is powered with a 12V
95 outboard motor (Minn Kota Traxxis 55) and uses two 12V 60 Kwh lead batteries for power
96 supply.

97 The SML is sampled by means of rotating glass plates (Duran, DWK Life Sciences, Germany).
98 The glass plates have a diameter of 60cm and are placed 10 cm above the water surface in the
99 gravity center of the catamaran. The plates rotated at a speed of 8 rpm. Depending of the
100 thickness of the SML the sampling volume typically varied between 80 and 120 ml min⁻¹
101 allowing to sample about 10 L SML within 2 h. The water collected in the wiper was transferred
102 into 2x 5 L Duran glass bottles with a peristaltic pump (Verder M25, 12V. Prior sampling the
103 SML sampling device was flushed for at least 15 minutes at the sampling site. Bulk water
104 samples were collected at a rate of 120 ml min⁻¹ from a depth of 70 cm using a peristaltic pump
105 (Verder M25, 12V) and collected in two 5 L Duran glass bottles. The material of the tubing is
106 Teflon PFA.

107 During the campaign, the catamaran was towed to the sampling site with a fishing boat and
108 typically operated 20 m apart from the fishing boat where manual SML sampling took place.
109

110 **Aerosol particle sampling and chemical analysis of inorganic ions, OC/EC and WSOC**

111 Aerosol particles (PM₁ and PM₁₀) were collected on preheated 150 mm quartz fiber filters
112 (Minktel, MK 360) at a flow rate of 700 L min⁻¹. Size-resolved aerosol particles were sampled
113 on pre-combusted aluminium rings with a Berner impactor. To avoid condensation of
114 atmospheric water on the surface of the aluminium foils, a conditioning unit was mounted
115 between the impactor inlet and the sampling unit consisting of a 3 m tube. By heating the
116 sampled air, high relative humidity of the ambient air was reduced to 75-80% before the
117 collection of the aerosol particles. The temperature difference between the ambient air at the
118 impactor inlet and the sampled air after the conditioning unit was below 9 K. Sampling time
119 was typically 24 h. After sampling, filters and aluminium foils were stored in aluminum boxes
120 at -20 °C and transported in dry ice to the TROPOS laboratories in Leipzig, Germany. The
121 chemical analysis of inorganic ions, the water soluble organic carbon (in the aerosol particles
122 as well as in the ocean water and cloud water) and elemental carbon are described in detail in
123 (van Pinxteren, et al. 2017; van Pinxteren, et al. 2015) and in Triesch, et al. (2019).
124

125 **Ice nucleating particles: sampling and analysis**

126 The quartz fiber filters mentioned in the previous paragraph (PM₁₀, PM₁ at CVAO and Mt.
127 Verde) were used for INP measurements. INP concentrations were also analysed for bulk
128 seawater and SML seawater collected at the seawater station Bahia das Gatas and in cloud water
129 collected during cloud events on the mountain top (Mt. Verde). All of the filter and water
130 samples were stored at -20 °C at Cape Verde and cooled below -20 °C during transportation to
131 TROPOS, where all samples were again stored at -20 °C until they were prepared for the
132 measurements. Two droplet freezing devices called LINA (Leipzig Ice Nucleation Array) and
133 INDA (Ice Nucleation Droplet Array) (Chen, et al. 2018; Hartmann, et al. 2019) were deployed
134 to characterize INP number concentrations (N_{INP}) from filter samples and in bulk and SML
135 seawater, and cloud water, yielding results in the temperature range from roughly -5°C to -
136 25°C.
137

138 **SAS analysis**

139 Phase sensitive alternating current (AC) voltammetry was applied for SAS analysis, being
140 already used as a successful tool for the determination of SAS in a plethora of environmental
141 aquatic samples (Frka, et al. 2009; Frka, et al. 2012; Kroflić, et al. 2018). Measurements (out-
142 of-phase mode, frequency 77 Hz, and amplitude 10 mV) were performed with µAutolab-type
143 II (Eco Chemie B. V., The Netherlands), GPES 4.6 software (Eco Chemie B. V., The
144 Netherlands), followed the method of Ćosović and Vojvodić (1998). A standard polarographic
145 Metrohm cell of 50 cm³ with a three-electrode system was used: working electrode - hanging
146 mercury drop electrode (HMDE; Metrohm, Switzerland; A = 0.01245 cm²), reference electrode
147 - Ag/AgCl/3 mol L⁻¹ KCl, auxiliary electrode - platinum coil. Before each measurement,
148 previously purified (450 °C for 5 h; charcoal organic residue removal) saturated NaCl (Kemika,
149 Croatia) solution was added to the sample to adjust the electrolyte concentration to 0.55 mol
150 L⁻¹.

151

152 **Lipid analysis: seawater**

153 Total lipid and lipid class quantitation was performed by Iatroscan thin layer chromatography–
154 flame ionization detection (TLC–FID) (Iatroscan MK-VI, Iatron, Japan). Lipids were separated
155 on Chromarods-SIII and quantified by an external calibration with a standard lipid mixture.
156 Quantified lipid classes include hydrocarbons (HC), lipid degradation indices (DI) (fatty acid
157 methyl esters (ME), free fatty acids (FFA), alcohols (ALC), 1,3-diacylglycerols (1,3DG), 1,2-
158 diacylglycerols (1,2DG) and monoacylglycerols (MG)), wax esters (WE), phytoplankton
159 energy reserves (triacylglycerols (TG)), membrane lipids including three phospholipids (PL)
160 (phosphatidylglycerols (PG), phosphatidylethanolamines (PE) and phosphatidylcholines (PC)),
161 glycolipids (GL) (sulfoquinovosyldiacylglycerols (SQDG), monogalactosyldiacylglycerols
162 (MGDG) and digalactosyldiacylglycerols (DGDG)), sterols (ST) and pigments (PIG). Lipids
163 indicating OM degradation (DegLip) comprise the sum of ME, FFA, ALC, DG, MG. The
164 standard deviation accounted for 3 to 11% of the signal magnitude of lipid classes (Gašparović,
165 et al. 2015; Gašparović, et al. 2017).

166

167 **Lipid biomarker and isotope analysis: aerosol particles**

168 Typically one half of the filter was used for the lipid analysis. The lipids were extracted with a
169 3:1 DCM mixture (4 times 40 ml) and the combined extracts were evaporated to a volume of 1
170 ml. The concentrated extract was dried with NaSO₄ and subsequently saponified with
171 KOH/MeOH (50 g L⁻¹, 1h, 80°C) and then evaporated to dryness. Afterwards the extract was
172 fractionated over an aminopropyl-column into four fractions of different polarity following
173 standard geochemical procedures (Hinrichs, et al. 2000). The fatty acids were converted to their
174 respective methylester using a BF₃/MeOH reagent and the alcohols were silylated with BSTFA
175 in the presence of pyridine. Each fraction was dried under nitrogen and picked up in a small
176 amount of hexane for the following analysis. Quantification was carried out with GC-FID and
177 GC-MS was used for identification. Compound specific carbon isotope analysis was done by
178 GC-C-IRMS.

179

180 **Trace metal analysis**

181 Size resolved aerosol particles were collected on polycarbonate foils (Wicom, Heppenheim,
182 Germany) using a low-pressure Berner impactor with a PM10 isokinetic inlet at a flow rate of
183 75 l min⁻¹. The collected particles were impacted on the foils creating spots of compressed
184 particles. The impacted spots were analyzed for their trace metal content using a Total
185 Reflection X-Ray Fluorescence (TXRF) S2 PICOFOX (Bruker AXS, Berlin, Germany)
186 spectrometer equipped with a Molybdenum X-ray source. Trace metals including Fe, Mn, Ca,
187 Cu, Zn, Se, V, Cr, Pb, Ni, Ti, Rb, Sr, Ba, La, and Ce were analyzed with a detection limit of a
188 few picograms. The sample preparation procedure and instrument specifications are described
189 in detailed in Fomba, et al. (2013). Trace metals were used to identify days of mineral dust
190 influence and its estimation.

191

192 **Pigments and Chlorophyll-*a***

193 For pigment analysis, several liters of bulk water were collected in a water depth of
194 approximately 30 cm by filling 5 L polypropylene bottles. The bulk water was filtered over GF-
195 F filters (Whatman, Germany) and the filters were immediately frozen at -20 °C until analysis.
196 For analysis, the GF-F filters were extracted in 5 mL ethanol, and 20 µL of the extract were

197 injected into the HPLC (Dionex, Sunnyvale, CA, USA) under gradient elution using
198 methanol/acetonitrile/water systems as eluents. Chlorophyll a, chlorophyll b, phaeophorbide a,
199 phaeophorbide b as well as chlorophyllide a were detected with fluorescence detection (FLD)
200 as described in van Pinxteren, et al. (2017). All other pigments were analyzed with a diode array
201 detector (DAD). Standard components were used for peak assignment and external calibration.

202

203 **DOM classes**

204 SML, bulk and cloud samples and aqueous aerosol extracts for DOM classes were filtered
205 through 0.45 μm Polyethersulfone (PES) syringe filters and stored chilled (0 to 4 $^{\circ}\text{C}$) in pre-
206 combusted glass “TOC” vials until analysis, which occurred within 2 days of sample
207 preparation. LC-OCD-OND (Liquid chromatography with organic carbon detection and
208 organic nitrogen detection), allows $\sim 1\text{ml}$ of whole water to be injected onto a size exclusion
209 column (SEC; 2 ml min^{-1} ; HW50S, Tosoh, Japan) with a phosphate buffer (potassium
210 dihydrogen phosphate 1.2 g L^{-1} plus 2 g L^{-1} di-sodium hydrogen phosphate x 2 H_2O , pH 6.58)
211 and separated into five “compound-group specific” DOM fractions. The resulting fractions are
212 identified using unique detectors for organic carbon, UV-amenable carbon and organic nitrogen
213 Huber, et al. (2011). All peaks were identified and quantified with bespoke software (Labview,
214 2013) normalized to International Humic Substances Society humic and fulvic acid standards,
215 potassium hydrogenphthalate and potassium nitrate. Cloud water and aerosol samples were
216 blank corrected based on sample blanks that were extracted in the same way. No sample blanks
217 were available for SML and bulk water samples.

218

219 **TEP**

220 Water samples (SML, cloud and bulk water) are then filtered onto 0.2 μm polycarbonate filters
221 at low pressure and stained with alcian blue. We used the spectrophotometric method to
222 analyse the stain (Passow and Alldredge 1995) which gives TEP in μg of xanthum gum
223 equivalent (μgXeqL^{-1}).

224

225 **Microbial cell counts**

226 Prokaryotic cell numbers were counted via flow cytometry after water samples were fixed,
227 flash-frozen in liquid nitrogen, and stored at -20°C . Prior measurements, all samples were
228 stained with SYBR Green solution. Counting was performed after addition of latex beads
229 serving as an internal standard. Further details can be found in Robinson, et al. (2019).

230 Small autotrophic cells were counted after addition of red fluorescent latex beads (Polysciences,
231 Eppelheim, Germany) and were detected by their signature in a plot of red (FL3) vs. orange
232 (FL2) fluorescence, and red fluorescence vs. side scatter (SSC). This approach allows
233 discrimination between different groups of prokaryotic and eukaryotic autotrophs (Marie, et al.
234 2010) which in our case were size classes defined as *Synechococcus*-like cells and
235 Nanoeukaryotes.

236

237 **Nitrous acid (HONO) using a LOnG Path Absorption Photometer (LOPAP)**

238 Nitrous acid (HONO) was continuously measured using a commercialized LOPAP HONO
239 analyzer (LOnG Path Absorption Photometer, QUMA Elektronik & Analytik GmbH), which
240 has been described in more details elsewhere. The instrument was placed in a ventilated
241 aluminum box. The temperature of the stripping coil was held at 25°C using a thermostat.

242 HONO was collected into a stripping coil and immediately converted in azodye. HONO
243 concentration is indirectly measured through the azodye absorption from 550 to 610 nm into
244 long path optical cell (2 meters). Temporal resolution of HONO was fixed to 30 seconds.
245 Sampling flow of the gas was 1 L min^{-1} . The uncertainty of HONO concentration was 10%
246 (2σ) with a detection limit of few ppt under our measurement condition. HONO concentration
247 was calibrated using a standard solution of NO_2^- (Titritisol Nitrite standard, $1000 \text{ mg L}^{-1} \text{ NO}_2^-$
248 in water). The instrument was frequently calibrated over the measurement period. To account
249 for zero drift, automatic zero air measurements were operated for 1 hour every 6.5 hours.

250

251 **GO: PAM**

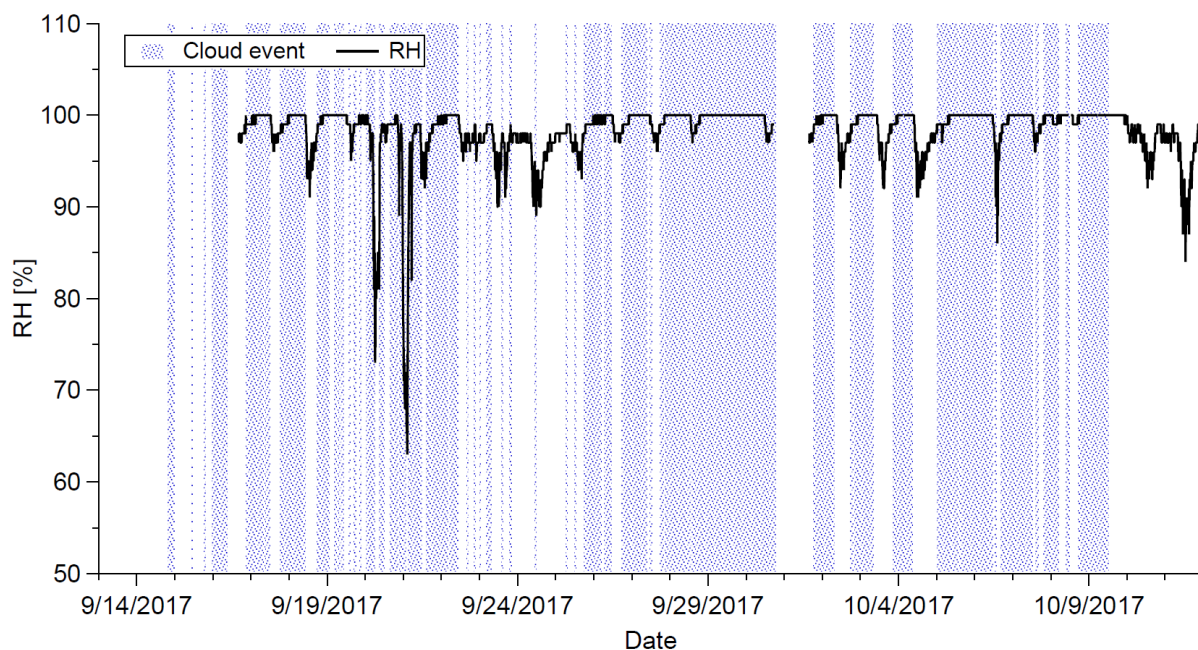
252 The photochemical setup consisted of a Quartz cell (2 cm diameter and 5 cm length) half filled
253 with SML samples collected the previous day and irradiated by means of a Xenon lamp. This
254 reactor was flushed by a flow of air containing a large concentration of ozone (ppm levels),
255 triggering O_3/OH reactions at the surface, but also in the bulk of the SML samples. The gaseous
256 products of these reactions were then injected into the Go:PAM where aerosol production could
257 take place.

258

259 **Characterization of cloud events at the Mt. Verde**

260 Cloud events were frequently observed at MT. Verde and characterized by relative humidity
261 (RH) values of 100 %. In addition, cloud events were verified from the PNSD as described in
262 Gong, et al. (2020).

263



264

265 Fig. S4: Time series of RH at Mt. Verde illustrated by the black line. Cloud event times are
266 indicated by the blue shadows. The method and the figure are closely adopted from Gong, et
267 al. 2020.

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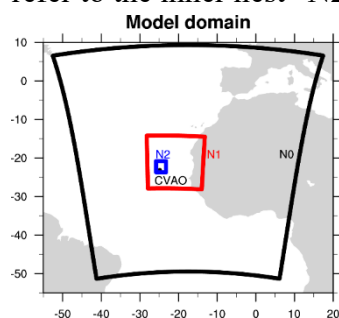
269 **Simulations I**

270 The meteorological model data by COSMO are used to define a vertical meteorological data
271 field as radio sounding data provided from the meteorological station Sal (Station number:

272 8594) are missing during the campaign. These data are used for meteorological 2D-simulations
273 with COSMO-MUSCAT (Wolke, et al. 2012) that is done to calculate if a cloud will also occur
274 over a flat surface area as from the 3D-simulations the effect of topography cannot be
275 completely ruled out. Therefore, the meteorological model data by the Consortium for Small-
276 scale Modeling-Multiscale Chemistry Aerosol Transport Model (COSMO), (Baldauf, et al.
277 2011) were used to define a vertical meteorological data field. For analyzing important
278 multiphase chemical pathways, and the impact of the horizontal and vertical transport on the
279 aerosol and cloud droplet composition within the MBL, the first model approach will be a box
280 modeling study with the air parcel model SPACCIM (Spectral Aerosol Cloud Chemistry
281 Model, Wolke, et al. 2005). SPACCIM is designed to investigate complex atmospheric
282 multiphase chemistry processes. It has proven its excellent capability to investigate important
283 multiphase chemical pathways (Hoffmann, et al. 2016; Hoffmann, et al. 2019; Tilgner and
284 Herrmann 2010) Therefore, the simulations enable the determination of the most relevant
285 multiphase chemical pathways for the chemical processing of marine aerosols during the
286 campaign. The second model approach will apply 2D-simulations of the marine multiphase
287 chemistry with the chemical transport model COSMO-MUSCAT (Baldauf, et al. 2011), which
288 numerical scheme is able to treat cloud droplet chemistry (Schrödner, et al. 2014). Hence, the
289 COSMO-MUSCAT simulations are able to investigate (i) the impact of horizontal and vertical
290 transport on the chemical aerosol as well as cloud droplet composition and (ii) the direct and
291 indirect impact of clouds on multiphase chemistry within, above and below the cloud. At the
292 end of the simulations the model results will be compared with the measured aerosol
293 concentration and composition and the in-cloud measurements at top of the Monte Verde
294 mountain. For this purpose, a novel reduced marine multiphase chemistry module will be
295 applied.

296 **Simulations II**

297 The simulations for the marine boundary layer height were carried out on three domains centred
298 on Sao Vicente, Cape Verde, applying one-way offline nesting. The outer domain “N0” has a
299 horizontal resolution of around 14 km and the two inner nests “N1” and “N2” have a grid
300 spacing of 3.5 km and 0.875 km, respectively (see Fig. S5). The results shown in this work only
301 refer to the inner nest “N2”.



302
303
304 *Fig. S5: Model domains of COSMO-MUSCAT simulations performed within the MarParCloud*
305 *project and geographical location of the Cape Verde Atmospheric Observatory (CVAO) at Sao*
306 *Vicente, Cape Verde. Nested 14 km (black line), 3.5 km (red line), and 0.875 km (blue line).*

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308

309 Table S1: Instruments employed at the CVAO during the campaign.

310

Measured parameter	Instrument	Performance	No. of measurements /samples (incl. blanks)	Running period	Responsible Institution
Chemical and biological composition of aerosol particles (PM ₁ , PM ₁₀ , TSP _{h.v.} , TSP _{l.v.}), INP measurements	Digitel sampler. 5-stage Berner impactor. High volume (h.v.) sampler. Low Volume (l.v.) samplers	8h to 48 h sampling	PM ₁ : 30 PM ₁₀ : 30 TSP _{h.v.} : 12 TSP _{l.v.} : 23 Impactor: 18x5 stages	15 th Sept – 11 th Oct	TROPOS. Leipzig, Germany
Physical characterization of aerosol size distribution	MPSS, APS, PNSD, CCNC	15 min sampling		13 th Sep – 13 th October	TROPOS. Leipzig
Meteorology	Automatic weather station	Minute samples		Continuous	NCAS. University of York. UK
VOCs, OVOCs,	Dual channel, gas chromatograph with flame ionization detection	Hourly samples		15 th Sep – 8 th October	NCAS. University of York. UK
Ozone	UV absorption	Minute samples		Continuous	NCAS. University of York. UK
Vertical profiles of meteorological parameters	Helikite	10 days of measurements	19 profiles	13 th Sep – 13 th October	TROPOS. Leipzig
SML and aerosol particles	Plunging waterfall tank	9 days of measurements	5 x SML and bulk water, 2 x 7 TSP samples	2 nd Oct – 10 th Oct	ZMT, Bremen, Germany
HONO	LoPAP	Continuously		15 th Sept – 11 th Oct	ICARE, Orleans, France
SOA forming potential	Go:PAM	Continuously for ambient air sampling	3 SML samples	15 th Sept – 11 th Oct	IRCELYON, Lyon, France.

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318 Table S2: Instruments employed at the Mt. Verde during the campaign.

Measured parameter	Instrument	Performance	No. of measurements /samples (incl. blanks) during campaign	Running period	Responsible Institution
Chemical and biological composition of aerosol particles (PM ₁ , PM ₁₀ , TSP _{h.v.} , TSP _{l.v.}), INP measurements	Digitel sampler. 5-stage Berner impactor. Low Volume (l.v.) sampler	8h to 48 h sampling	PM ₁ : 19 PM ₁₀ : 19 TSP _{l.v.} : 3 Impactor: 7x5 stages	21 th Sept – 09 th Oct	TROPOS. Leipzig
Physical characterization of aerosol size distribution	MPSS, APS, PNSD, CCNC	15 min sampling		13 th Sep – 13 th October	TROPOS. Leipzig
Meteorology	Automatic weather station (Davis)	5 min sampling		Continuous	TROPOS. Leipzig
Cloud water for chemical, biological and INP measurements	Cloud water sampler (six samplers run in parallel)	2.5 – 13 h sampling	155	20 th Sept – 09 th Oct	TROPOS. Leipzig

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332 Table S3: Details on bulk water and SML sampling at Bahia das Gatas at N16°53. W24°54.

333

Sampling date	Local sampling time (UTC-1)	Exact coordinates	Sampling device	Sampling conditions	Bulk water			SML			Sample ID
					salinity [ppt equals psU]	pH - value	temperature [°C]	salinity [ppt equals psU]	pH - value	temperature [°C]	
18.09.2017	11:35 -12:00	No data	GP*	very windy, big waves	34.1	8.1	25.0	-	-	-	Seawater 1
19.09.2017	8:38 -9:05	No data	GP	not reported	36.1	8.2	25.2	-	-	-	Seawater 2
20.09.2017	8:32 -9:54	N16°53.341 W24°54.360	GP	Slick and foam at the end of the sampling	36.3	8.1	26.7	36.2	8.1	26.7	Seawater 3
22.09.2017	8:56 -9:20	N16°53'48.002 W24°54'13.858	GP	very windy with big waves	-	-	-	-	-	-	Seawater 4
25.09.2017	9:45 -10:48	N16°53.753 W24°54.117	GP	calm sea	36.4	8.2	26.0	36.4	8.1	25.5	Seawater 5
26.09.2017	10:05-10:51	N16°53.934 W24.54.556	GP	not reported	36.3	8.2	25.1	36.1	8.1	26.4	Seawater 6
26.09.2017	11:10-11:50	N16°53.742 W24°54.061	CAT**	not reported	36.2	8.0	26.1	36.4	8.0	26.1	Seawater 6 (2)

27.09.2 017	8:50 -10:03	N16°53.748 W24°54.134	GP	not reported	36.4	8.1	24.0	36.3	8.1	23.7	Seawater 7
27.09.2 017	8:50 -10:03	N16°53.623 W24°54.257	CAT	not reported	36.2	7.9	27.0	36.0	7.9	26.7	Seawater 7 (2)
28.09.2 017	9:15 -10:05	N16°53.623 W24°54.257	GP	not reported	36.5	8.1	27.8	36.6	8.1	26.8	Seawater 8
28.09.2 017	9:15 -10:05	N16°53.623 W24°54.257	CAT	not reported	36.3	8.1	27.1	36.0	8.1	27.8	Seawater 8 (2)
02.10.2 017	8:30 - 9:30	No data	GP	very windy, drift during sampling	36.0	8.2	20.9	36.5	8.3	20.9	Seawater 9
02.10.2 017	8:30 -9:30	No data	GP	very windy, drift during sampling	35.9	8.2	22.7	36.1	8.1	23.9	Seawater 9 (2)
03.10.2 017	8:15 - 9:35	N16°53.341 W24°54.360	GP	Not reported	36.1	8.2	22.8	36.6	8.2	21.0	Seawater 10
03.10.2 017	8:15 - 9:35	N16°53.341 W24°54.360	CAT	not reported	36.2	8.2	23.5	36.3	8.2	22.4	Seawater 10 (2)
04.10.2 017	8:15 - 9:00	N16°53'48.002 W24°54'13.858	GP	not reported	36.2	8.23	23.7	-	-	-	Seawater 11
04.10.2 017	8:15 - 9:55	N16°53.934 W24.54.556	CAT	not reported	36.3	8.2	22.8	-	-	-	Seawater 11 (2)

05.10.2 017	9:24 - 9:41	N16°53'44.824 W24°54'7.021	GP	smaller and higher waves in change	36.5	8.2	22.9	-	-	-	Seawater 12
06.10.2 017	08:04- 09:47	N16°53.753 W24°54.117	GP	windy	36.3	8.2	23.7	36.6	8.2	20.7	Seawater 13
07.10.2 017	09:22 - 10:35	N16°53.742 W24°54.061	GP	very windy with long and short waves in change	36.4	8.2	21.8	36.7	8.2	21.2	Seawater 14
07.10.2 017	9:17 - 10:46	N16°53.742 W24°54.061	GP CAT	very windy with long and short waves in change	36.5	8.2	22.4	36.7	8.2	24.5	Seawater 14 (2)
09.10.2 017	8:30 - 9:17	N16°53.623 W24°54.257	GP	windy	36.4	8.1	23.6	36.6	8.2	21.5	Seawater 15
10.10.2 017	8:30 - 9:30	N16°53.623 W24°54.257	GP	windy up to very windy with long waves	36.3	8.2	22.4	36.4	8.2	21.7	Seawater 16

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336 *GP = glass plate, **cat = catamaran

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338 Table S4: Concentrations of pigments, DOC and microbial parameters in the SML and bulk water samples.

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Sampling date	18.09. 2017	19.09. 2017	20.09. 2017	22.09. 2017	25.09. 2017	26.09. 2017	27.09. 2017	28.09. 2017	02.10. 2017	03.10. 2017	04.10. 2017	05.10. 2017	06.10. 2017	07.10. 2017	09.10. 2017	10.10. 2017	
Sample ID	Sea-water 1	Sea-water 2	Sea- water 3	Sea- water 4	Sea- water 5	Sea- water 6	Sea- -water 7	Sea- water 8	Sea- water 9	Sea- water 10	Sea- water 11	Sea- water 12	Sea- water 13	Sea- water 14	Sea- water 15	Sea- water 16	
Parameter	unit																
<i>pigments</i>																	
chlorophyll c ₂	µg L ⁻¹	0.016	-	0.026	-	0.031	-	0.041	0.023	0.029	0.028	-	0.025	-	0.039	0.039	0.062
19 butanoyl oxyfucoxanthin	µg L ⁻¹	0.000	-	0.017	-	0.002	-	0.018	0.004	0.019	0.012	-	0.013	-	0.026	0.031	0.058
hexanoyloxyfuco xanthin	µg L ⁻¹	0.015	-	0.044	-	0.018	-	0.045	0.027	0.045	0.033	-	0.031	-	0.062	0.069	0.110
chlorophyll b	µg L ⁻¹	0.021	-	0.042	-	0.055	-	0.038	0.032	0.050	0.061	-	0.035	-	0.063	0.065	0.109
chlorophyll a	µg L ⁻¹	0.112	-	0.216	-	0.323	-	0.335	0.184	0.264	0.298	-	0.192	-	0.346	0.437	0.604
fucoxanthin	µg L ⁻¹	0.014	-	0.044	-	0.093	-	0.171	0.040	0.045	0.060	-	0.037	-	0.105	0.153	0.223
phaeophorbide a	µg L ⁻¹	0.031	-	0.032	-	0.047	-	0.041	0.037	0.034	0.036	-	0.033	-	0.037	0.039	0.038
phaeophytin a	µg L ⁻¹	0.017	-	0.018	-	0.027	-	0.023	0.023	0.026	0.029	-	0.020	-	0.025	0.026	0.027
chlorophyllide a	µg L ⁻¹	0.000	-	0.010	-	0.010	-	0.010	0.000	0.000	0.010	-	0.000	-	0.010	0.010	0.000
violaxanthin	µg L ⁻¹	0.000	-	0.003	-	0.005	-	0.003	0.003	0.004	0.006	-	0.003	-	0.005	0.006	0.007
diadinoxanthin	µg L ⁻¹	0.009	-	0.018	-	0.016	-	0.021	0.014	0.017	0.017	-	0.013	-	0.023	0.030	0.034
lutein	µg L ⁻¹	0.001	-	0.002	-	0.005	-	0.003	0.002	0.003	0.003	-	0.002	-	0.003	0.003	
chlorophyll c ₃	µg L ⁻¹	0.014	-	0.022	-	0.027	-	0.038	0.017	0.025	0.025	-	0.022	-	0.035	0.034	0.059
peridinin	µg L ⁻¹	0.003	-	0.005	-	0.007	-	0.006	0.003	0.007	0.005	-	0.006	-	0.005	0.005	0.008
zeaxanthin	µg L ⁻¹	0.108	-	0.106	-	0.134	-	0.089	0.136	0.141	0.206	-	0.165	-	0.185	0.148	0.129
β-carotene	µg L ⁻¹	0.006	-	0.009	-	0.017	-	0.011	0.010	0.013	0.018	-	0.010	-	0.014	0.015	0.015
<i>DOC</i>																	
DOC SML (GP*)	µg L ⁻¹	3260	2240	2780	2580	1680	2390	2020	2270	2020	2190	2050	-	3330	1610	1940	1820
DOC SML (cat**)	µg L ⁻¹	-	-	-	-	-	2040	1450	1940	1520	1520	-	-	-	1850	-	-
DOC bulk water (GP)	µg L ⁻¹	2800	1480	1090	1419	2370	1640	1070	1370	1340	1370	1090	1140	1290	947	1160	1290

DOC bulk water (cat)	$\mu\text{g L}^{-1}$	-	-	-	-	-	1810	1160	1280	1260	1260	1840	-	-	1480	-	-
microbial parameters																	
LNA (SML)	cells mL^{-1}	-	-	1.06E+06	-	1.48E+06	9.34E+05	1.05E+06	1.20E+06	9.90E+05	1.30E+06	1.15E+06	-	1.06E+06	-	-	-
HNA (SML)	cells mL^{-1}	-	-	8.91E+04	-	1.62E+05	5.85E+04	1.03E+05	1.23E+05	1.40E+05	7.84E+04	7.04E+04	-	9.58E+04	-	-	-
TCN (SML)	cells mL^{-1}	-	-	1.15E+06	-	1.64E+06	9.92E+05	1.15E+06	1.33E+06	1.13E+06	1.38E+06	1.22E+06	-	1.16E+06	-	-	-
synechococcus (SML)	cells mL^{-1}	-	-	4.42E+04	-	6.70E+04	2.71E+04	2.63E+04	2.61E+04	3.41E+04	7.40E+04	4.53E+04	-	2.42E+04	-	-	-
nanoecaryotes (SML)	cells mL^{-1}	-	-	6.76E+02	-	1.64E+02	5.16E+01	5.16E+01	5.16E+01	1.85E+02	1.02E+01	2.05E+01	-	-	-	-	-
LNA (bulk water)	cells mL^{-1}	-	-	9.70E+05	-	1.24E+06	1.06E+06	1.07E+06	1.05E+06	1.72E+06	1.32E+06	1.21E+06	1.24E+06	1.18E+06	-	-	-
HNA (bulk water)	cells mL^{-1}	-	-	7.19E+04	-	1.17E+05	7.49E+04	9.15E+04	8.66E+04	6.07E+05	1.44E+05	7.91E+04	9.31E+04	9.88E+04	-	-	-
TCN (bulk water)	cells mL^{-1}	-	-	1.04E+06	-	1.36E+06	1.13E+06	1.16E+06	1.14E+06	2.32E+06	1.46E+06	1.28E+06	1.34E+06	1.28E+06	-	-	-
synechococcus (bulk water)	cells mL^{-1}	-	-	2.45E+04	-	4.27E+04	2.05E+04	2.63E+04	1.99E+04	6.82E+04	5.31E+04	3.79E+04	2.96E+04	2.42E+04	-	-	-
nanoecaryotes (bulk water)	cells mL^{-1}	-	-	5.16E+01	-	1.54E+02	6.18E+01	1.23E+02	1.03E+02	1.46E+03	1.74E+02	6.14E+01	6.18E+01	1.06E+01	-	-	-

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*GP: glass plate; *cat: catamaran (MarParCat)

TCN: total bacterial cell numbers

HNA: high nucleic acid containing cells

LNA : low nucleic acid containing cells

- : no measurements available

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*LOD: limit of detection

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352 Table S5: Concentrations and standard deviations (ng m⁻³) of size resolved aerosol particle constituents measured in parallel during parallel measurements between
 353 2.10 and 9.10 obtained from the CVAO (mean value of 5 blocks) and the Mt.Verde (mean value of 6 blocks) station. Aerosol particles were sampled in five different
 354 size fractions with aerodynamic particle diameter D_P (50% cut-off) from: 0.05-0.14 μm (stage 1), 0.14-0.42 μm (stage 2), 0.42-1.2μm (stage 3), 1.2-3.5 μm (stage
 355 4) and 3.5-10 μm (stage 5).
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CVAO	Sodium	Chloride	MSA	Sulfate	Nitrate	Oxalate	Ammonium	Potassium	Magnesium	WSOM	EC
stage 1	<LOD*	<LOD*	0.5±0.4	28.0±12	<LOD*	<LOD*	12.5±4	<LOD*	<LOD*	9.4±21	6.3±0.9
stage 2	<LOD*	<LOD*	12.7±2	741.9±111	<LOD*	3.2±4	232±48	3.7±5	0.3±0.6	38.6±16	27.5±14
stage 3	148±27	50.1±47	6.6±07	472.2±26	13.7±15	0.9±1	23.3±10	11.4±6	16.0±5.3	49.1±3	14.0±10
stage 4	1430±375	2300±680	4.5±0.9	572±159	698±76	22.1±8	16.6±4	54.4±11	156±33	108±6	12.5±1
stage 5	1520±306	2820±697	0.9±0.3	503±123	536±59	2.8±4	17.6±5	56.9±14	181±42	91.5±13	3.1±4
Mt. Verde	Sodium	Chloride	MSA	Sulfate	Nitrate	Oxalate	Ammonium	Potassium	Magnesium	WSOM	EC
stage 1	<LOD*	<LOD*	0.2±0.4	7.4±8	<LOD*	<LOD*	1.4±4	<LOD*	<LOD*	15.8±18	1.8±2
stage 2	<LOD*	<LOD*	2.1±1	111±54	<LOD*	0.3±0.4	36.5±22	0.3±0.7	<LOD*	26.9±21	15.0±12
stage 3	24.9±22	<LOD*	3.7±1	254±86	<LOD*	0.4±0.7	51.8±20	1.3±2.1	1.5±2.6	44.6±29	21.4±11
stage 4	163±110	277±171	2.3±0.7	156±62	96.5±68	0.3±0.8	12.6±10	2.9±5.1	13.4±8.5	80.2±29	3.6±5
stage 5	143±85	256±143	2.1±1	109±35	79.3±46	<LOD*	6.9±7	2.6±4.4	9.2±11	137±56	21.0±16

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359 *LOD: limit of detection

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366 Table S6: Average concentrations and standard deviations of DOM fractions in the SML, bulk water and cloud water (ng L⁻¹) and in aerosol particles (PM₁₀)
 367 sampled at the CVAO and at the Mt. Verde (ng m⁻³) sampled in parallel within the periods: 26. – 27.09., 01. – 02.10., and 08. – 09.10.2017.
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	N	Biopolymers	Humic Substances	Building Blocks	LMW Neutrals	LMW Acids
SML	3	77±27	205±11	163±21	359±84	38±31
Bulk water	3	67±19	234±44	137±40	362±127	2±3
aerosol particles at CVAO	3	<LOD*	<LOD*	114±25	84±38	2±3
aerosol particles at Mt. Verde	2	<LOD*	<LOD*	55±23	58±22	6±4
Cloud water	4	27±21	354±222	242±224	301±227	58±98

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 372 *LOD: limit of detection
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