



Assessing the effect of dissolution on planktonic foraminiferal Mg/Ca ratios: Evidence from Caribbean core tops

Marcus Regenberg and Dirk Nürnberg

*Leibniz Institute of Marine Sciences IFM-GEOMAR, Wischhofstr. 1–3, D-24148 Kiel, Germany
(mregenberg@ifm-geomar.de; dnuernberg@ifm-geomar.de)*

Silke Steph

Leibniz Institute of Marine Sciences IFM-GEOMAR, Wischhofstr. 1–3, D-24148 Kiel, Germany

Now at Alfred Wegener Institute for Polar and Marine Research AWI, Am Alten Hafen 26, D-27568 Bremerhaven, Germany (ssteph@awi-bremerhaven.de)

Jeroen Groeneveld

Leibniz Institute of Marine Sciences IFM-GEOMAR, Wischhofstr. 1–3, D-24148 Kiel, Germany

Now at Research Center Ocean Margins RCOM, University of Bremen, Leobener Str., D-28359 Bremen, Germany (jgroeneveld@uni-bremen.de)

Dieter Garbe-Schönberg

Institute of Geosciences, University of Kiel, Luderwig-Meyn-Str. 10, D-24118 Kiel, Germany (dgs@gpi.uni-kiel.de)

Ralf Tiedemann

Leibniz Institute of Marine Sciences IFM-GEOMAR, Wischhofstr. 1–3, D-24148 Kiel, Germany

Now at Alfred Wegener Institute for Polar and Marine Research AWI, Am Alten Hafen 26, D-27568 Bremerhaven, Germany (rtiedemann@awi-bremerhaven.de)

Wolf-Christian Dullo

*Leibniz Institute of Marine Sciences IFM-GEOMAR, Wischhofstr. 1–3, D-24148 Kiel, Germany
(cdullo@ifm-geomar.de)*

[1] In order to assess the dissolution effect on foraminiferal Mg/Ca ratios, we analyzed Mg/Ca of seven planktonic foraminiferal species and four of their varieties from Caribbean core tops from ~900–4700 m water depth. Depending on the foraminiferal species and variety, Mg/Ca start to decline linearly below $\Delta[\text{CO}_3^{2-}]$ levels of ~18–26 $\mu\text{mol/kg}$ by ~0.04–0.11 mmol/mol per 1 $\mu\text{mol/kg}$ $\Delta[\text{CO}_3^{2-}]$, similar to decreases of ~0.5–0.8 mmol/mol per kilometer below ~2500–3000 m water depth. Above these species-specific critical levels, Mg/Ca remains stable with higher intraspecific Mg/Ca variability than below. We developed routines to correct Mg/Ca from below these critical thresholds for dissolution effects, which

reduce the overall intraspecific variability by ~24–64%, and provide dissolution-corrected Mg/Ca appropriate to calculate Holocene paleotemperatures. When taking into account only dissolution-unaffected Mg/Ca from <2000 m, the systematic succession of foraminiferal species according to their Mg/Ca reflects expected calcification depths.

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Theme: Development of the Foraminiferal Mg/Ca Proxy for Paleoceanography

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1. Introduction

[2] Mg/Ca ratios of planktonic foraminiferal tests have been developed as a powerful tool in paleoceanography to assess past ocean temperatures [e.g., Elderfield and Ganssen, 2000; Nürnberg, 2000; Lea, 2003]. This proxy became widely used mainly due to the fact that both stable oxygen isotopes ($\delta^{18}\text{O}$) and Mg/Ca can be measured within the same biotic carrier, which ensures the recording of the same seasonality and/or habitat effects that may occur when proxy data from different faunal groups are compared. Most advantageous, combined measurements of Mg/Ca and $\delta^{18}\text{O}$ allow to extract the $\delta^{18}\text{O}$ record of past upper ocean water, and accordingly salinity variations [e.g., Nürnberg, 2000; Lea, 2003; Schmidt et al., 2004; Pahnke and Zahn, 2005; Nürnberg and Groeneveld, 2006].

[3] The accurate assessment of paleotemperatures from different depth levels of the ocean is a prerequisite to reconstruct and model past changes in salinity and density gradients, stratification patterns, and thus thermohaline circulation. The exponential dependence of Mg/Ca ratios on temperature within single planktonic foraminiferal species is well-studied, resulting in a variety of species-specific [Nürnberg, 1995; Nürnberg et al., 1996a, 1996b, 2000; Lea et al., 1999, 2000; Mashiotta et al., 1999; Dekens et al., 2002; Rosenthal and Lohmann, 2002; Whitko et al., 2002; McKenna and Prell, 2004; McConnell and Thunell, 2005] and multi-species calibration curves

[Elderfield and Ganssen, 2000; Anand et al., 2003]. The accuracy of the Mg/Ca paleothermometry is specified with ± 0.2 – 1.2°C [Nürnberg, 1995; Nürnberg et al., 1996a, 2000; Hastings et al., 1998; Elderfield and Ganssen, 2000; Lea et al., 2000; Dekens et al., 2002; Anand et al., 2003]. Nevertheless, the influence of factors on Mg/Ca other than temperature, which may affect the amount of magnesium incorporated into foraminiferal tests during calcification, is still debated. Primary effects such as intraspecies and ontogenetic variations [e.g., Nürnberg et al., 1996a], pH, carbonate ion concentration ($[\text{CO}_3^{2-}]$), and salinity [e.g., Nürnberg et al., 1996a, 1996b; Lea et al., 1999; Russell et al., 2004] as well as secondary (diagenetic) effects such as preferential removal of Mg^{2+} during calcite dissolution [e.g., Savin and Douglas, 1973; Cronblad and Malmgren, 1981; Brown and Elderfield, 1996; Rosenthal et al., 2000; Rosenthal and Lohmann, 2002; Dekens et al., 2002; de Villiers, 2003] may influence foraminiferal Mg/Ca.

[4] Several efforts have been made to correct the foraminiferal Mg/Ca ratios for the effect of dissolution. For example, Rosenthal and Lohmann [2002] established dissolution-corrected Mg/Ca versus temperature calibrations for *Globigerinoides ruber* and *Globigerinoides sacculifer*, based on shell mass loss studies along depth transects, where the preexponential constant is a function of size-normalized shell weight. In an alternative core-top study along depth transects, Dekens et al. [2002] introduced water depth-dependent dissolution cor-

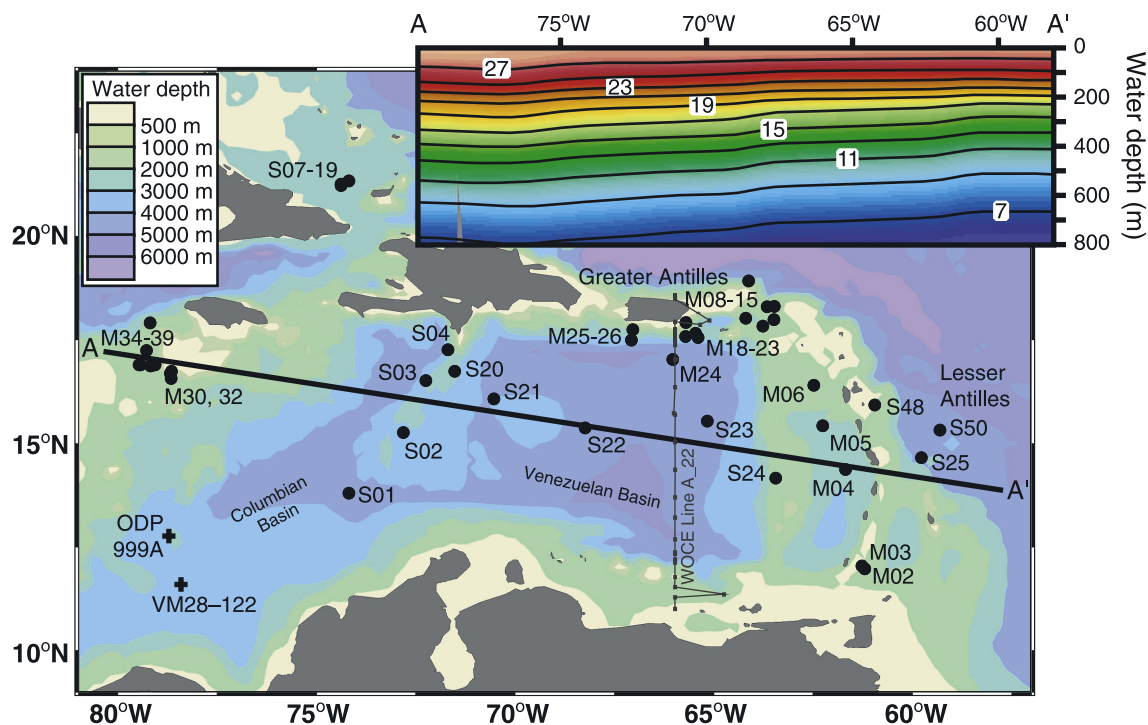


Figure 1. Bathymetric chart of the Caribbean [Schlitzer, 2002] showing the locations of the core-top samples from this study (black dots, station labels abbreviate the cruise stations: S: S0164-...; M: M35/1, M350...) and of two sediment cores (crosses) [Schmidt *et al.*, 2004] discussed here. Inset presents the annual mean temperature [NODC, 2001] versus water depth along the profile AA'. WOCE Line A22 stations are used to compute the carbonate ion concentration considered to be representative for the Caribbean.

rection factors into the exponent of their Mg/Ca versus temperature calibrations for *G. ruber*, *G. sacculifer*, and *Neogloboquadrina dutertrei*, differentiated into Atlantic and Pacific equations. Yet only a few core-top studies covered the upper 2000 m of the water column to a larger extent.

[5] Our study focusses on the impact of water depth-related partial calcite dissolution on the Mg/Ca ratios of seven planktonic foraminiferal species and four of their varieties from water depths of ~900–4700 m. All selected foraminifers were collected from core-top sediments from the Caribbean and the adjacent tropical Atlantic (Figure 1). The thermal structure in the study area minimizes temperature-induced variability in the foraminiferal Mg/Ca, and allows to isolate the impact of calcite dissolution. Our results indicate that Mg/Ca remains stable down to species-specific water depths.

2. Hydrography

[6] Although seasonal variability influences the strength of the trade winds and current patterns in

the Caribbean [Schott *et al.*, 1988; Molinari *et al.*, 1990; Morrison and Smith, 1990; Larsen, 1992; Johns *et al.*, 2002], seasonal temperature changes are relatively small. Within the mixed layer, spatial temperature gradients at similar water depths are in the order of 1°C, increasing to ~4°C within the thermocline [National Oceanographic and Data Center (NODC), 2001] (Figure 1).

[7] The Upper North Atlantic Deep Water (UNADW) with salinity values of 34.7–34.9 [Wüst, 1963; Joyce *et al.*, 1999; Schmuker and Schiebel, 2002] ventilates the Columbian and Venezuelan Basins via the Greater Antilles Passages [Stalcup *et al.*, 1975; Morrison and Nowlin, 1982; Fratantoni *et al.*, 1997]. Above the UNADW, Antarctic Intermediate Water (AAIW) enters the Caribbean between ~550–1000 m [Wüst, 1963; Bulgakov and Lomakin, 1995; Schmuker and Schiebel, 2002]. AAIW mixes with overlying Subtropical Under Water (SUW), forming the permanent thermo- and nutricline in the Caribbean [Kameo, 2002]. The SUW with salinity values of >36.8 [Corredor and Morell, 2001; Schmuker and Schiebel, 2002] and temper-

Table 1. Locations, Depths, and Ages of Core Top Samples of This Study^a

Station	Longitude, W	Latitude, N	Water Depth <i>d</i> , m	Sample, cm	¹⁴ C Age, years BP	Lab. Number	2 Sigma Age Ranges, years BP	Marine Reservoir Correction (ΔR), years
M35006-6	62°27.2'	16°25.3'	888	0–1				
M35013-3	63°27.0'	18°18.9'	899	0–1				
SO164-04-2	71°39.09'	17°16.38'	1,013	0–1				
M35038-1	79°13.8'	17°15.7'	1,066	0–1				
M35012-6	63°37.6'	18°18.3'	1,121	0–1				
M35039-1	79°08.7'	17°55.6'	1,142	0–1				
M35023-4	65°41.0'	17°36.2'	1,183	0–1				
M35037-1	79°23.3'	16°54.8'	1,190	0–1				
M35023-3	65°40.9'	17°36.2'	1,192	0–1				
M35036-3	79°25.3'	16°55.0'	1,196	0–1				
M35034-3	79°03.5'	16°54.2'	1,212	0–1				
M35015-1	63°27.1'	17°59.6'	1,230	0–1				
M35035-1	79°07.9'	16°53.6'	1,252	0–1				
SO164-48-2	60°55.0'	15°57.02'	1,286	0–1				
M35030-1	78°36.6'	16°45.3'	1,298	0–1				
M35003-6	61°14.7'	12°05.1'	1,299	0–1				
M35032-1	78°36.9'	16°35.3'	1,363	0–1				
M35002-1	61°10.6'	12°01.9'	1,506	0–1				
SO164-24-3	63°25.43'	14°11.89'	1,545	0–1				
M35014-1	63°44.2'	17°50.5'	1,604	0–1				
SO164-18-1	74°21.0'	21°13.61'	1,629	0–1	1,350 ± 25	KIA23386	774–924	37 ± 14
SO164-19-3	74°20.98'	21°14.7'	1,706	0–1	> AD 1954	KIA23383		
M35018-1	65°22.2'	17°34.5'	1,728	0–1				
M35025-1	67°00.5'	17°45.2'	1,778	0–1	699 ± 30	^b	280–445	–16 ± 23
M35019-1	65°26.1'	17°40.3'	1,815	0–1				
M35020-2	65°40.2'	17°55.8'	2,005	0–1				
M35005-3	62°13.9'	15°27.2'	2,289	0–1				
M35010-2	64°05.4'	18°56.0'	2,696	0–1				
SO164-25-3	59°44.48'	14°41.25'	2,720	0–1	1,915 ± 30	KIA23385	1,287–1,577	33 ± 60
SO164-07-3	74°08.76'	21°19.46'	2,722	0–1	720 ± 35	KIA23384	262–423	37 ± 14
SO164-03-3	72°12.31'	16°32.40'	2,744	0–1				
M35008-1	64°09.8'	18°01.9'	2,820	0–1				
M35004-1	61°39.7'	14°24.6'	2,885	0–1				
SO164-02-3	72°47.06'	15°18.29'	2,977	0–1	2,205 ± 25	KIA23382	1,717–1,914	–16 ± 23
SO164-20-2	71°29.22'	16°45.49'	3,357	0–1				
M35026-2	67°02.6'	17°30.5'	3,815	0–1	1,175 ± 70	^b	628–900	–16 ± 23
SO164-21-3	70°30.0'	16°06.0'	3,995	0–1				
SO164-50-3	59°16.94'	15°21.25'	4,002	0–1				
SO164-01-3	74°09.028'	13°50.195'	4,026	0–1				
SO164-23-3	65°08.09'	15°34.01'	4,328	0–1				
SO164-22-2	68°12.0'	15°24.0'	4,506	0–1				
M35024-6	66°00.1'	17°02.6'	4,710	0–1	2,510 ± 45	^b	2,059–2,318	–16 ± 23

^aThe ¹⁴C ages are calibrated after *Stuiver et al.* [2005]. Marine reservoir corrections after *Reimer et al.* [2004].

^bFrom *Schmuker* [2000].

atures of 18–23°C [*Morrison and Nowlin*, 1982] ranges between ~80–180 m [*Wüst*, 1964] or even down to ~300 m [*Schmuker and Schiebel*, 2002] (Figure 3). It emanates from the North Atlantic subtropical gyre and enters the Caribbean via the Greater Antilles Passages [*Wüst*, 1964; *Johns et al.*, 2002]. The mixed layer is composed of Caribbean Water (CW) in the upper ~80 m of the water column with salinity values of ~35.5 and temper-

atures of ~28°C [*Wüst*, 1964; *Corredor and Morell*, 2001] (Figure 3).

3. Materials and Methods

3.1. Core-Top Sediments

[8] Mg/Ca analyses were performed on tests of planktonic foraminiferal species or their varieties

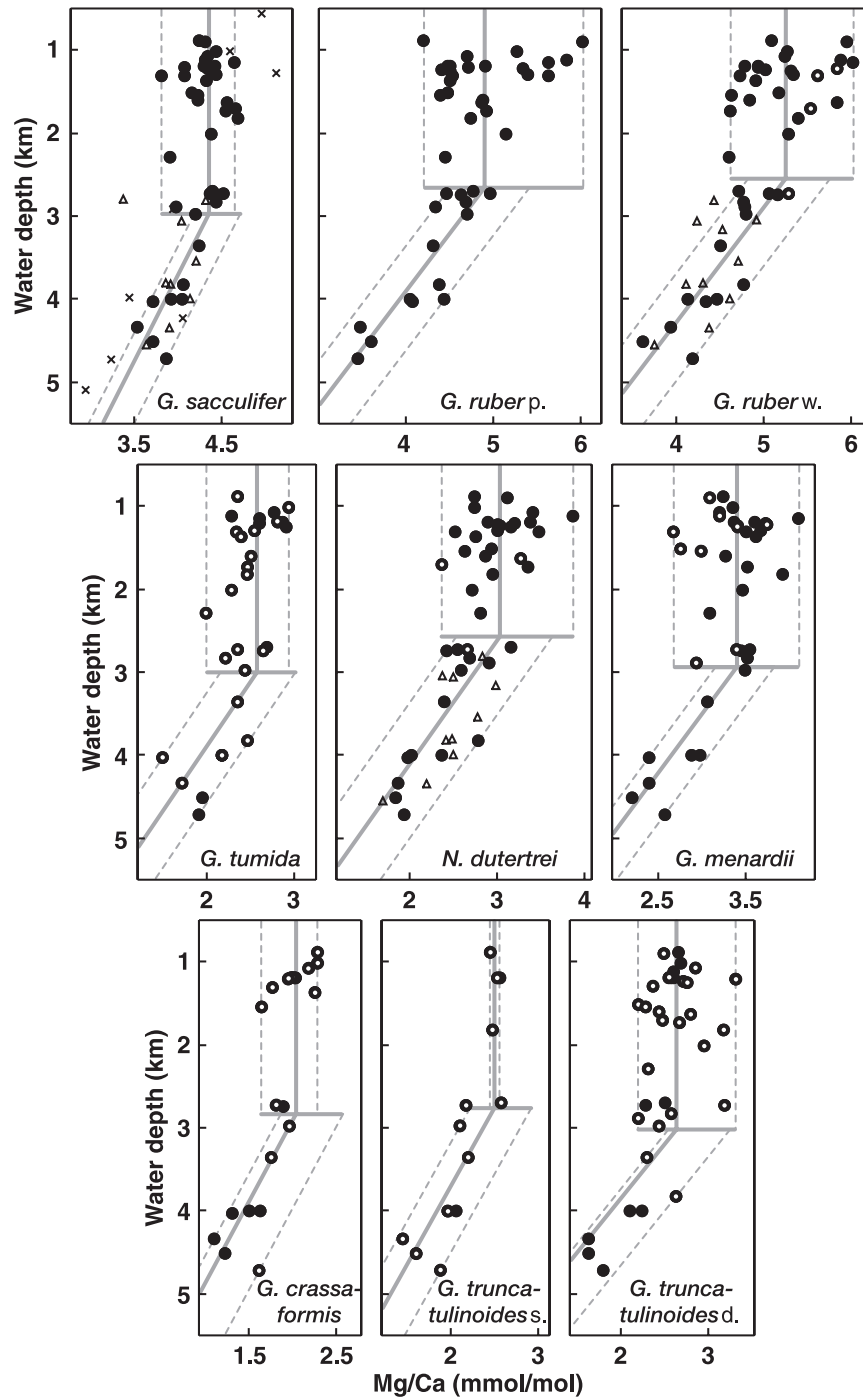


Figure 2. Mg/Ca ratios (black dots from 355–400 μm size fraction; open circles from enlarged size fractions; see Table 2) versus water depth for each planktonic foraminiferal species or variety. The data show significant decreases below species-specific water depth levels ($d_{critical}$ = horizontal line). $d_{critical}$ is defined as the intercept between the vertical (species-specific mean Mg/Ca calculated from samples <2000 m water depth (Table 2)) and diagonal lines (regression lines from Mg/Ca from >3000 m water depth). Below $d_{critical}$, the intraspecific scatter in Mg/Ca is significantly reduced. The dashed lines mark envelopes of measured Mg/Ca. For comparison, Mg/Ca of Rosenthal *et al.* [2000] (crosses) and Dekens *et al.* [2002] (triangles) are included. As the data of Dekens *et al.* [2002] are based on a foraminiferal cleaning protocol involving the reductive hydrazine step, we added 15% to their original Mg/Ca data according to the instructions of Rosenthal *et al.* [2004].

from 42 sediment surface samples (0–1 cm; Figure 1; Table 1), collected during R/V SONNE SO164 [Nürnberg *et al.*, 2003] and R/V METEOR M35/1 cruises [Zahn *et al.*, 1996]. On the basis of 7 AMS¹⁴C-dated core tops revealing ages not older than 2–3 kyrs (Table 1), we assume the remaining core-top sediments to be within the same age range.

[9] Twenty to 25 visually (magnification up to 40×) uncontaminated tests of the foraminiferal species or their varieties were selected from the narrow 355–400 μm size fraction (~550–800 μg) for Mg/Ca analyses to minimize size-related intraspecific elemental variations [Elderfield *et al.*, 2002]. In case of insufficient material, we subsequently extended the selection of tests to a larger size range (maximum 250–500 μm; Figure 2; Table 2).

3.2. Foraminiferal Species Selection

[10] Seven tropical to subtropical planktonic foraminiferal species and four of their varieties were selected for Mg/Ca analyses. Generic assignments of selected specimens in this study follow Kennett and Srinivasan [1983] and Hemleben *et al.* [1989]. The specific adaptations to and demands of light, chlorophyll concentrations, salinity, temperature, and food availability [e.g., Fairbanks *et al.*, 1980, 1982; Fairbanks and Wiebe, 1980; Curry *et al.*, 1983; Deuser, 1987; Deuser and Ross, 1989; Sautter and Thunell, 1991a; Ortiz *et al.*, 1995, 1996, 1997; Watkins and Mix, 1998] determine the vertical distribution patterns and abundances of planktonic foraminiferal species in the water column. Their different preferred depth habitats thus allow the reconstruction of the upper ocean structure in detail, presupposing that the various foraminiferal habitats remain fixed through time and environmental change.

[11] *G. ruber* and *G. sacculifer* are spinose and symbiont-bearing species. They commonly reach highest abundances in the upper 50 m of the mixed layer [Bé, 1977; Fairbanks *et al.*, 1982; Deuser, 1987; Bijma *et al.*, 1994; Kroon and Darling, 1995; Kemle-von Mücke and Oberhänsli, 1999; Schmuker and Schiebel, 2002]. An almost uniform occurrence throughout the year makes *G. ruber* suitable to reflect annual hydrographic conditions [Hemleben *et al.*, 1989; Lin *et al.*, 1997; Tedesco and Thunell, 2003]. Savin and Douglas [1973] showed that *G. ruber* calcifies at shallower water depths than *G. sacculifer*.

[12] *Globorotalia menardii* and *N. dutertrei* live within the tropical to subtropical thermocline [Ravelo and Fairbanks, 1992; Chaisson and Ravelo, 1997]. *N. dutertrei* is known to occur in a well-stratified photic zone near the deep chlorophyll maximum [Fairbanks *et al.*, 1980, 1982; Fairbanks and Wiebe, 1980], often associated with the lower thermocline [Sautter and Thunell, 1991a; Ravelo and Fairbanks, 1992].

[13] *Globorotalia tumida* calcifies near the bottom of the photic zone [Ravelo and Fairbanks, 1992; Chaisson and Ravelo, 1997]. It shows distinct preferences for low water densities during summer [Hilbrecht, 1996]. *Globorotalia truncatulinoides* and *Globorotalia crassaformis* are deep-dwelling species that reach maximum abundances below the photic zone [Ganssen and Kroon, 2000]. According to Mulitza *et al.* [1997], *G. truncatulinoides* reflects mean ocean conditions at about 200 m, while McKenna and Prell [2004] assign the habitat to the permanent thermocline. In the Caribbean, *G. truncatulinoides* is linked to the SUW [Schmuker, 2000]. Changing coiling directions do not indicate substantially different physical preferences [Hilbrecht, 1996]. *G. crassaformis* calcifies below the photic zone and thermocline [Ravelo and Fairbanks, 1992]. In general, *G. crassaformis* is supposed to behave like *G. truncatulinoides* [Hemleben *et al.*, 1989].

[14] Impending gametogenesis is morphologically signaled by additional calcification of a diminutive final chamber for *G. ruber* [Bijma *et al.*, 1990], a sac-like final chamber for *G. sacculifer* [Bé *et al.*, 1983; Hemleben *et al.*, 1989], and by discarding of spines [Bé, 1980; Duplessy *et al.*, 1981]. Secondary calcite crust formation, associated with reproduction, mainly occurs at greater depths [Hemleben *et al.*, 1989; Lohmann, 1995], and hence may affect the geochemical signature of the foraminiferal tests [e.g., Curry and Crowley, 1987; Spero and Williams, 1989; Lohmann, 1995; Nürnberg *et al.*, 1996a; Rosenthal *et al.*, 2000; Eggins *et al.*, 2003; Mulitza *et al.*, 2003; McKenna and Prell, 2004]. Addition of secondary calcite accounts for about one third of the test's mass of *G. sacculifer* [Bé, 1980; Erez and Honjo, 1981; Hemleben *et al.*, 1989; Schweitzer and Lohmann, 1991; Bijma *et al.*, 1994], and doubles the test's mass of *G. truncatulinoides* [Bé and Lott, 1964]. Large vertical migrations during their ontogenetic cycles were described for *G. menardii*, *G. tumida*, and *G. truncatulinoides* [Bé and Ericson, 1963; Bé,



Table 2. Measured Mg/Ca Ratios and Size Fractions of Selected Specimens^a

Station	d, m	G. ruber p.	G. ruber w.	G. sacculifer	N. dutertrei	G. menardii	G. tumida	G. truncatulinoides d.	G. truncatulinoides s.	G. crassaformis
M35006-6	888	a 4.20	a 5.09	a 4.23	a 2.73	a 3.24	d 2.35	a 2.65	d 2.45	b 2.28
M35013-3	899	a 6.02	a 5.95	a 4.31	a 3.11	d 3.09		d 2.48		b 2.28
SO164-04-2	1,013	a 5.26	a 5.27	a 4.43	a 2.74	a 3.35	i 2.93	c 2.68		g 2.18
M35038-1	1,066	a 4.70	a 5.23	a 4.34	a 3.41	a 3.20	a 2.76	g 2.85		
M35012-6	1,121	a 5.84	a 5.87	a 4.31	a 3.87	i 3.20	a 2.27	a 2.59		
M35039-1	1,142	a 5.62	a 6.02	a 4.64	a 5.59 ^b	a 4.10 ^c	a 2.59	a 5.80 ^b		
M35023-4	1,183	a 4.50						a 2.58		f 2.04
M35037-1	1,190	a 4.91	a 4.94	a 4.41	a 2.89	a 3.36	e 2.80	a 2.55		e 2.01
M35023-3	1,192	a 4.48	a 4.77	a 4.29	a 3.37	a 3.60	a 2.87	a 2.60		a 1.97
M35036-3	1,196	a 4.714	a 4.929	a 4.066	a 3.192	a 3.728	a 2.594	g 3.304	e 2.55	e 1.94
M35034-3	1,212	a 5.34	b 5.84	a 4.34	a 3.00	b 3.74				
M35015-1	1,230	a 4.41	a 5.01	a 4.39	a 3.03	c 3.41		c 2.71		
M35035-1	1,252	a 4.50	a 5.30	a 4.41	a 3.16	a 3.40	a 2.90	h 2.75		
SO164-48-2	1,286	a 5.39	a 5.33	a 4.43	a 3.00	a 3.66	c 2.55	c 2.36		
M35030-1	1,298	a 5.62	b 5.61	a 4.07	a 3.47	a 3.51	d 2.33			
M35003-6	1,299	a 4.53	a 4.72	a 3.80	a 2.51	e 2.67				e 1.77
M35032-1	1,363	a 4.51	a 4.91	a 4.33	a 2.75	a 3.61	c 2.40	d 2.20		d 2.26
M35002-1	1,506	a 4.47	a 5.17	a 4.15	a 2.93	c 2.75		c 2.28		
SO164-24-3	1,545	a 4.39	a 4.63	a 4.23	a 2.63	j 2.98		d 2.44		c 1.64
M35014-1	1,604	a 4.88	a 4.83	a 4.23	a 2.86	a 3.26		b 2.79		
SO164-18-1	1,629	a 4.86	a 5.83	a 4.55	b 3.27			c 2.48		
SO164-19-3	1,706		e 5.53	a 4.65	e 2.36			c 2.67		
M35018-1	1,728	a 4.91	a 4.62	a 4.54	a 3.35	a 3.51	c 2.46	c 2.67		
M35019-1	1,815	a 4.73	a 5.39	a 4.68	a 2.95	a 3.92	b 2.46	b 3.17		
M35020-2	2,005	a 5.14	a 5.28	a 4.38	a 2.71	a 3.45	c 2.28	b 2.95	d 2.47	
M35005-3	2,289	a 4.45	a 4.59	a 3.90	a 2.81	a 3.08	d 1.99	d 2.30		
M35010-2	2,696	a 4.77	a 4.70	a 4.38	a 3.15	a 4.66 ^c	a 2.69	a 2.50	b 2.57	c 1.81
SO164-25-3	2,720	a 4.46	a 5.05	a 4.36	a 2.55	a 3.54	b 2.35	a 2.28	c 2.17	
SO164-07-3	2,722	a 4.96	b 5.28	a 4.51	b 2.66	c 3.39		c 3.19		
SO164-03-3	2,744	a 4.62	a 5.15	a 4.44	a 2.42	a 3.44	c 2.64	b 2.57		a 1.89
M35008-1	2,820	a 4.68	a 4.77	a 4.43	a 2.68	a 3.51	d 2.20	b 2.57		
M35004-1	2,885	a 4.34	a 4.78	a 3.98	a 2.91	c 2.94	b 2.20	b 2.20		
SO164-02-3	2,977	a 4.70	a 4.79	a 4.20	a 2.58	a 3.49	c 2.43	c 2.43	c 2.10	b 1.96
SO164-20-2	3,357	a 4.30	a 4.51	a 4.23	a 2.39	a 3.05	b 2.35	b 2.30	c 2.20	b 1.76
M35026-2	3,815	a 4.37	a 4.76	a 4.06	a 2.78	a 3.99 ^c	d 2.46	g 2.62		
SO164-21-3	3,995	a 4.44	a 4.45	a 4.04	a 2.36	a 2.88	b 2.17	a 2.24	b 1.96	a 1.63
SO164-50-3	4,002	a 4.04	a 4.13	a 3.92	a 2.02	a 2.98	i 2.17	a 2.10	a 2.05	a 1.50
SO164-01-3	4,026	a 4.07	a 4.34	a 3.71	a 1.98	a 2.40	h 1.48			a 1.31
SO164-23-3	4,328	a 3.47	a 3.93	a 3.53	a 1.87	a 2.38	i 1.72		c 1.45	a 1.09
SO164-22-2	4,506	a 3.60	a 3.62	a 3.71	a 1.84	a 2.20	a 1.95	a 1.62	c 1.60	a 1.23
M35024-6	4,710	a 3.44	a 4.19	a 3.86	a 1.93	a 2.56	a 1.91	a 1.79	e 1.88	e 1.61
Mg/Ca _{mean}		4.9	5.25	4.34	3.03	3.36	2.58	2.64	2.5	2.04

Table 2. (continued)

Station	<i>d</i> , m	<i>G. ruber</i> p.	<i>G. ruber</i> w.	<i>G. sacculifer</i>	<i>N. dutertrei</i>	<i>G. menardii</i>	<i>G. tumida</i>	<i>G. truncatulinoides</i> d.	<i>G. truncatulinoides</i> s.	<i>G. crassaformis</i>
sd		0.51	0.44	0.2	0.35	0.32	0.22	0.27	0.05	0.22
Temperature		28.41	29.18	27.06	23.07	24.22	21.28	21.54	20.93	18.67

^a Each single ratio is the average of three measurements, except the samples listed in Table 3. Mean Mg/Ca (Mg/Ca_{mean}) \pm standard deviations (sd) from 2000 m are converted into temperatures after Anand *et al.* [2003] ($Mg/Ca = 0.38 \cdot \exp(0.09T)$). Specimens size fractions: a, 355–400 μ m; b, 315–400 μ m; c, 315–450 μ m; d, 315–500 μ m; e, 250–400 μ m; f, 250–500 μ m; g, 250–500 μ m; h, 355–500 μ m; i, 355–450 μ m; j, 400–500 μ m.

^b Not used for calculation of the mean $<2,000$ m.

^c Not considered for the calculation of the linear regression.

1977; Fairbanks *et al.*, 1980, 1982; Fairbanks and Wiebe, 1980; Erez and Honjo, 1981; Hemleben *et al.*, 1989; Schweitzer and Lohmann, 1991; Brown and Azmy, 2005].

[15] To prevent our analyses from biases due to different amounts of secondary calcite, we took care upon specimen selection. For *G. ruber* (pink and white varieties) and *G. sacculifer*, tests with spines were preferentially selected. For the latter, specimens showing a sac-like final chamber were excluded. For *G. menardii* and for *G. tumida*, we chose thin-walled and thick-encrusted specimens, respectively. Due to the low numbers of tests of *G. truncatulinoides* (dextral and sinistral varieties) and *G. crassaformis*, we did not differentiate between encrusted and nonencrusted specimens, and different morphotypes. In general, specimens with kummerform chambers were rejected.

3.3. Mg/Ca Analyses

[16] The foraminiferal samples were cleaned according to the cleaning protocol of Barker *et al.* [2003]. Prior to cleaning, the tests were gently crushed between two glass plates in order to open all chambers. The foraminiferal fragments were rinsed 5 times with ultrapure water and twice with methanol (suprapure), including ultrasonic treatment after each rinse. Subsequently, samples were treated twice with 250 μ L of a hot (97°C) oxidizing 1% NaOH/H₂O₂ reagent (10 mL 0.1 N NaOH (analytical grade); 100 μ L 30% H₂O₂ (suprapure)) for 10 minutes. Every 2.5 minutes, the solution was cautiously agitated in order to release any gaseous build-up. After 5 minutes, the samples were placed in an ultrasonic bath for a few seconds in order to maintain the chemical reaction. Remaining oxidizing solution was removed by three rinsing steps with ultrapure water. After transferring the samples into clean vials, a weak acid leach with 250 μ L 0.001 M nitric acid (HNO₃, subboiled distilled) was applied with 30 seconds ultrasonic treatment, followed by two rinses with ultrapure water. After removal of any remaining solution, the samples were dissolved in 500 μ L 0.075 M HNO₃ (sub-boiling distilled), and diluted with ultrapure water to achieve Ca concentrations of 30–70 ppm.

[17] Analyses were performed on two ICP OES devices showing no significant offset as revealed by replicate measurements of 21 samples (Table 3). Each single Mg/Ca ratio (Tables 2 and 3) is the average of three measurements, from which the analytical errors are deduced. One set of samples was measured on an ICP OES (ISA Jobin Yvon,

Table 3. Replicate Mg/Ca Measurements on Two ICP OES Devices^a

Station	Species	Mg/Ca, mmol/mol ISA Jobin Yvon		Mg/Ca, mmol/mol Spectro Ciros			Mg/Ca, mmol/mol	
		1	2	1	2	3	Mean	sd
SO164-22-2	<i>G. crassaformis</i>			* 1.22	* 1.23		1.23	0.00
SO164-04-2	<i>G. crassaformis</i>			* 2.29	* 2.28		2.28	0.00
SO164-02-3	<i>G. crassaformis</i>			* 2.07	** 1.85		1.96	0.08
SO164-01-3	<i>G. crassaformis</i>			* 1.32	** 1.30		1.31	0.01
M35018-1	<i>G. menardii</i>			* 3.31	** 3.72		3.51	0.29
M35037-1	<i>G. menardii</i>			* 3.53	** 3.20		3.36	0.23
M35036-3	<i>G. menardii</i>			* 3.81	** 3.28	*** 4.09	3.73	0.41
SO164-20-2	<i>G. ruber</i> p.			* 4.30	* 4.30		4.30	0.00
SO164-02-3	<i>G. ruber</i> p.			* 4.81	** 4.59		4.70	0.08
SO164-18-1	<i>G. ruber</i> p.			* 4.87	* 4.85		4.86	0.00
SO164-25-3	<i>G. ruber</i> w.	* 4.75		* 5.35			5.05	0.21
SO164-23-3	<i>G. ruber</i> w.	* 3.58		** 4.33	*** 3.87		3.93	0.25
SO164-04-2	<i>G. ruber</i> w.	* 5.34		** 5.20			5.27	0.05
SO164-01-3	<i>G. ruber</i> w.	* 4.38		* 4.41	** 4.22		4.34	0.10
SO164-19-3	<i>G. ruber</i> w.	* 5.75		* 5.32			5.53	0.15
SO164-48-2	<i>G. sacculifer</i>			* 4.48	* 4.38		4.43	0.03
SO164-24-3	<i>G. sacculifer</i>	* 4.10		** 4.36			4.23	0.09
M35014-1	<i>G. sacculifer</i>			* 4.36	** 4.09		4.23	0.19
SO164-23-3	<i>G. sacculifer</i>	* 3.41		** 3.66	*** 3.52		3.53	0.08
SO164-22-2	<i>G. sacculifer</i>	* 3.90		** 3.53			3.71	0.13
M35037-1	<i>G. sacculifer</i>			* 4.39	** 4.44		4.41	0.04
SO164-50-3	<i>G. truncatulinooides</i> d.			* 2.14	** 2.07		2.10	0.02
SO164-48-2	<i>G. truncatulinooides</i> d.			* 2.35	** 2.27		2.31	0.03
M35023-4	<i>G. truncatulinooides</i> d.			* 2.59	** 2.56		2.58	0.02
M35023-4	<i>G. truncatulinooides</i> s.			* 2.41	** 2.65		2.53	0.17
SO164-50-3	<i>G. tumida</i>			* 2.17	* 2.16		2.17	0.01
SO164-25-3	<i>G. tumida</i>			* 2.36	* 2.35		2.35	0.00
SO164-48-2	<i>G. tumida</i>			* 2.55	* 2.54		2.55	0.00
SO164-23-3	<i>G. tumida</i>			* 1.64	* 1.64	** 1.87	1.72	0.12
SO164-22-2	<i>G. tumida</i>			* 1.96	* 1.94		1.95	0.01
SO164-20-2	<i>G. tumida</i>			* 2.35	* 2.35		2.35	0.00
SO164-03-3	<i>G. tumida</i>			* 2.66	* 2.63		2.64	0.01
SO164-02-3	<i>G. tumida</i>			* 2.44	* 2.43		2.43	0.00
SO164-01-3	<i>G. tumida</i>			* 1.49	* 1.48		1.48	0.00
SO164-50-3	<i>N. dutertrei</i>	* 1.99		* 2.05			2.02	0.02
SO164-25-3	<i>N. dutertrei</i>	* 2.49		* 2.60			2.55	0.04
SO164-48-2	<i>N. dutertrei</i>	* 3.30	** 2.85	* 3.17	** 2.84	** 2.87	3.00	0.19
SO164-23-3	<i>N. dutertrei</i>	* 1.84		* 1.89			1.87	0.02
SO164-22-2	<i>N. dutertrei</i>	* 1.82	** 1.87	* 1.82	** 1.85		1.84	0.02
SO164-21-3	<i>N. dutertrei</i>	* 2.38	* 2.33				2.36	0.02
SO164-20-2	<i>N. dutertrei</i>	* 2.55		* 2.35	* 2.36		2.42	0.04
SO164-04-2	<i>N. dutertrei</i>	* 2.74		* 2.73			2.74	0.00
SO164-03-3	<i>N. dutertrei</i>	* 2.38		* 2.47			2.42	0.03
SO164-02-3	<i>N. dutertrei</i>	* 2.57		* 2.59			2.58	0.01
SO164-01-3	<i>N. dutertrei</i>	* 1.91	** 1.99	* 1.95	** 2.07		1.98	0.05
SO164-07-3	<i>N. dutertrei</i>	* 2.62		* 2.69			2.66	0.02
SO164-19-3	<i>N. dutertrei</i>	* 2.32		* 2.40			2.36	0.03
SO164-18-1	<i>N. dutertrei</i>	* 3.28		* 3.26			3.27	0.01

^aThe use of same sample solutions is indicated by asterisks.

Spex Instruments S.A. GmbH) with polychromator applying yttrium as an internal standard. Selected element lines for analyses (Ca: 317.93 nm; Mg: 279.55 nm; Y: 371.03 nm) were most intensive and undisturbed. Element detection was performed with photomultipliers, the high-tension of which

was adapted to each element concentration range. The analytical error for Mg is ~0.45% and for Ca ~0.15%. Replicate samples showed an average standard deviation of ~0.1 mmol/mol (Table 3). A second set of samples was measured on a simultaneous, radially viewing ICP OES (Spectro

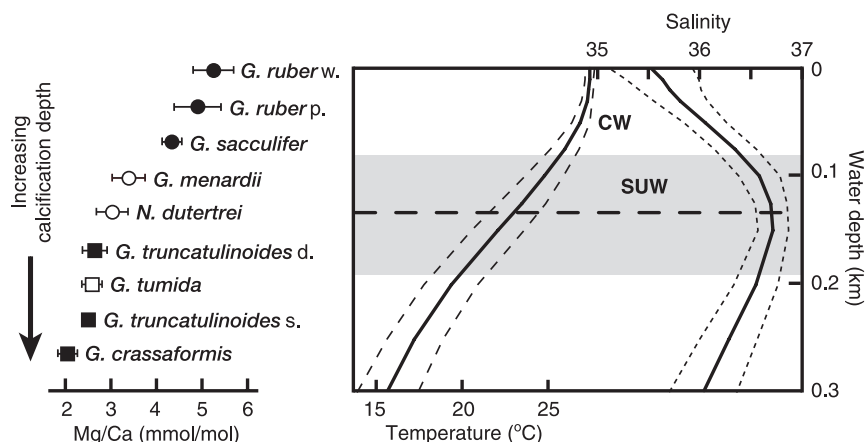


Figure 3. Species-specific mean Mg/Ca ratios from <2000 m water depth (error bars indicate standard deviations) ranked from highest to lowest Mg/Ca versus relative calcification depth. Black dots: shallow-dwelling species; circles: thermocline-dwelling species; open square: bottom-of-photoc-zone-dwelling species; black squares: deep-dwelling species. For comparison, eastern Caribbean annual mean temperatures and salinities (\pm standard deviations) [NODC, 2001] versus water depth are shown. Shaded area indicates the thermocline. Horizontal dashed line marks the base of the photic zone (inferred from the irradiance value of 0.1%). CW, Caribbean Water; SUW, Subtropical Underwater.

CirosCCD SOP). A cooled cyclonic spraychamber in combination with a microconcentric nebulizer (200 μ L/min sample uptake) was optimized for best precision of analytical results and minimized uptake of sample solution. Sample introduction took place via autosampler (Spectro A.I.). For Ca, we used the spectral line with the highest stability (183.801 nm). For Mg, we used the most sensitive line (279.553 nm). Matrix effects caused by varying concentrations of Ca were cautiously checked and found to be insignificant. Drift of the machine during analytical sessions was negligible (<0.5%, as determined by analysis of an internal consistency standard after every 5 samples). The analytical error for the Mg/Ca analyses was \sim 0.1%. Replicate samples showed an average standard deviation of \sim 0.08 mmol/mol (Table 3).

4. Results and Discussion

4.1. Bathymetric Change in Mg/Ca Ratios

[18] For all foraminiferal species and varieties studied, we observe a considerable decrease in Mg/Ca ratios with increasing water depth below certain depth levels. Depending on the species or variety, the change from stable to continuously decreasing Mg/Ca happens in the depth range between \sim 2000 m and \sim 3000 m. Statistical analyses of the intraspecific Mg/Ca reveal that means of samples from <2000 m water depth differ

significantly from those from >2000 m. Related low probability values ($p < 0.015$) of these Mg/Ca means indicate that both data sets behave significantly different and that environmental factors other than temperature bias Mg/Ca at deeper sites. According to the bathymetric distribution of the core-top samples in combination with the pattern of the Mg/Ca data (Figure 2), we assign decreasing Mg/Ca to dissolution and differentiate between three water depth intervals: (1) samples <2000 m with no signs of preferential removal of Mg^{2+} (number of core-top samples $n = 23$; Table 3), (2) samples >3000 m with distinctively decreasing Mg/Ca ($n = 8$; Table 3), and (3) samples between 2000–3000 m ($n = 9$; Table 3). The incomplete distribution pattern of the latter sample set prevents to directly infer the transition depth separating Mg/Ca unaffected and affected by dissolution, respectively.

4.2. Dissolution Unaffected Mg/Ca Ratios From Shallow Water Depths

[19] Evidence for excellent carbonate preservation at shallow water depths is provided by the presence of aragonitic pteropod shells in our Caribbean core-top sediments down to \sim 2700 m. Likewise, species-specific Mg/Ca ratios do not systematically decrease with increasing water depth in the depth interval <2000 m (correlation coefficients $r^2 = <0.1$). We assume these shallow Mg/Ca to be unaffected by dissolution.

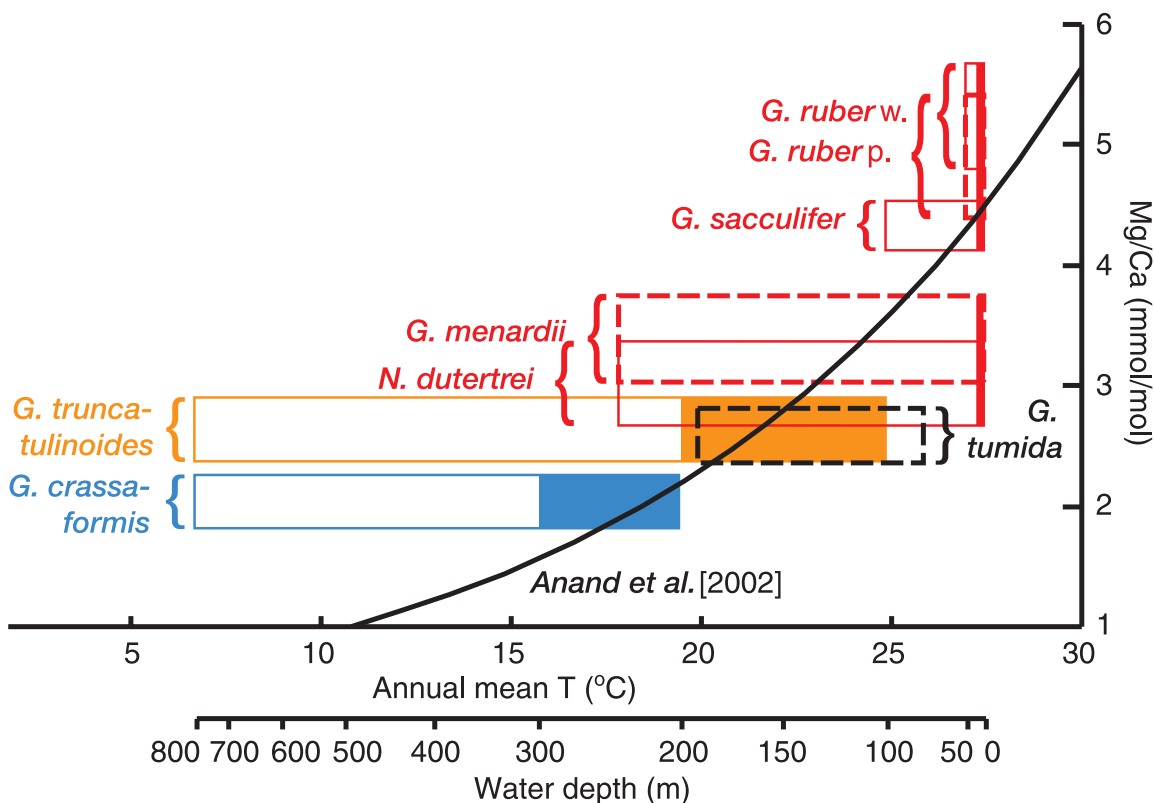


Figure 4. Species-specific mean Mg/Ca ratios from <2000 m water depth (box heights indicate Mg/Ca mean \pm standard deviations) versus eastern Caribbean annual temperatures [NODC, 2001]. The eastern Caribbean annual temperatures were converted from the according habitat depths compiled for the different species and their varieties (widths of open and filled boxes; see section 4.3). Considering only those habitat specifications (filled boxes) derived from eastern Caribbean plankton tows [Schmuker and Schiebel, 2002] (0–20 m (red bars); 100–200 m (orange bar); 200–300 m (blue bar)), the Mg/Ca versus temperature relationship comes close to the multispecies calibration curve of Anand et al. [2003] (black curve).

[20] There are significant differences in Mg/Ca ratios between the species and varieties from <2000 m (Figure 2; Table 2). Mixed layer-dwelling *G. ruber* exhibits highest Mg/Ca of 4.20–6.02 mmol/mol. In accordance with increasing calcification depth and decreasing ambient seawater temperature, Mg/Ca successively decrease to 1.64–2.28 mmol/mol for the deepest-living species *G. crassaformis* (Figure 2). With respect to *G. ruber*, the other species show lower mean Mg/Ca in samples <2000 m water depth (Figure 3): *G. sacculifer* is lower in Mg/Ca by \sim 15%, *G. menardii* by \sim 34%, *N. dutertrei* by \sim 40%, *G. tumida* by \sim 49%, *G. truncatulinoides* by \sim 49%, and *G. crassaformis* by \sim 60%. *G. ruber* pink shows consistently lower mean Mg/Ca from <2000 m by \sim 7% with respect to the white variety. The left-coiling variety of *G. truncatulinoides* (sinistral) is lower by \sim 5% with respect to the right-coiling (dextral) variety (Figures 2 and 3). These differences between species are in general agreement to previous multispe-

cies core-top studies of Hecht et al. [1975], Lorens et al. [1977], Rosenthal and Boyle [1993], Russell et al. [1994], Brown and Elderfield [1996], Hastings et al. [1996], Dekens et al. [2002], and Anand et al. [2003] (Figure 2).

4.3. Temperature Versus Mg/Ca Ratios Unaffected by Dissolution

[21] The accuracy of Mg/Ca versus temperature calibration curves depends on both the accuracy of Mg/Ca analyses and, more importantly, on the accuracy of the temperature assignment. The temperature assignment, however, is dependent on the foraminiferal habitat (e.g., water depth, season, life cycle, nutrient supply), which often comprises a broad range of water depths wherein calcification takes place. In Figure 4, we plotted the foraminiferal Mg/Ca ratios from <2000 m versus Caribbean annual mean temperatures [NODC, 2001]. These temperatures were converted from the specific

foraminiferal habitat assignments found in the literature [Fairbanks *et al.*, 1980, 1982; Fairbanks and Wiebe, 1980; Erez and Honjo, 1981; Hemleben *et al.*, 1989; Sautter and Thunell, 1991a; Schweitzer and Lohmann, 1991; Ravelo and Fairbanks, 1992; Mulitza *et al.*, 1997; Kemle-von Mücke and Oberhänsli, 1999; Schmuker and Schiebel, 2002; Anand *et al.*, 2003; McKenna and Prell, 2004]. Considering only the habitat specifications of Schmuker and Schiebel [2002] inferred from abundance maxima of eastern Caribbean plankton tows (April–May 1996), an exponential multispecies Mg/Ca versus temperature relationship suggests itself, which approaches the multispecies calibration curve of Anand *et al.* [2003] established for the Sargasso Sea (Figure 4). The slope of such a calibration curve, however, is highly dependent on the determination of the foraminiferal habitat depths. In fact, while estimates on where foraminifers live and where calcification occurs are manifold or speculative, $\delta^{18}\text{O}$ data of primarily shallow-dwelling species indicate that planktonic foraminifera calcify in depth zones that are significantly narrower than the overall vertical distribution of these species implies [Fairbanks *et al.*, 1980]. From Mg/Ca of the thermocline-dwelling species *G. menardii* and *N. dutertrei*, which are lower than those of the shallow-dwelling species *G. ruber* and *G. sacculifer* (Figures 3 and 4), we hypothesize that the potential calcification depths of these species, where the Mg/Ca signal is generated, are clearly deeper than the shallow abundance maxima given by Schmuker and Schiebel [2002].

[22] Despite the overall correspondence between foraminiferal Mg/Ca ratios and temperature shown in Figure 4, which heavily relies on assumptions concerning the foraminiferal habitats, we note a broad intraspecific range of Mg/Ca in core-top samples from <2000 m water depth, in accordance to early observations of Nürnberg [1995] on the high variability of Mg/Ca in foraminiferal tests. The scatter appears to be largest in shallow- and thermocline-dwelling species with a range of ~ 1.4 mmol/mol in *G. ruber* white, ~ 1.8 mmol/mol in *G. ruber* pink, ~ 0.9 mmol/mol in *G. sacculifer*, ~ 1.3 mmol/mol in *G. menardii*, and ~ 1.5 mmol/mol in *N. dutertrei*, and a commonly smaller range in deep-dwelling species of ~ 0.7 mmol/mol in *G. tumida*, ~ 1.1 mmol/mol in *G. truncatulinoides* dextral, ~ 0.1 mmol/mol in *G. truncatulinoides* sinistral, and ~ 0.7 mmol/mol in *G. crassaformis* (Figure 2). The observed range of core-top Mg/Ca appears high, but is close to other shallow core-top

and sediment-trap data from other ocean areas at similar temperatures showing a wide range of Mg/Ca of ~ 1 mmol/mol [e.g., Elderfield and Ganssen, 2000; Whitko *et al.*, 2002; Anand *et al.*, 2003] (Figure 5).

[23] The observed scatter in the core-top Mg/Ca ratios may be due to several reasons: (1) An annual mixed layer temperature range of ~ 2 – 3°C at the respective core sites within the mixed layer (i.e., ~ 30 – 50 m water depth, assumed habitat of shallow-dwelling species [Schmuker and Schiebel, 2002]) and a smaller annual temperature range of only ~ 1 – 2°C at intermediate water depths (down to ~ 230 m water depth, assumed habitat of *G. crassaformis* [Schmuker and Schiebel, 2002]) (Figure 5) introduces a potential variability in Mg/Ca of ~ 0.1 – 0.4 mmol/mol when applying the Mg/Ca versus temperature equation of Anand *et al.* [2003]. (2) The assumed age range of ~ 2 – 3 kyrs for the core-top sediments (see section 3.1) might have introduced a certain variance in Mg/Ca due to short-term climate variations. Indeed, a cooling of Caribbean surface water temperatures of about 2°C during the Little Ice Age [Watanabe *et al.*, 2001] could account for a change in Mg/Ca of ~ 0.7 mmol/mol. (3) A species-specific size-effect on Mg/Ca generally was avoided by selecting foraminiferal tests from narrow size ranges (see section 3.1; Table 2). Especially for the deep-dwelling species, however, we were forced to widen the size range due to insufficient numbers of specimens for analysis (Table 2). Taking the published deviations from mean Mg/Ca of 12–36% by enlargement of the foraminiferal test sizes [Elderfield *et al.*, 2002; Anand and Elderfield, 2005], we estimate the size effect on our Mg/Ca to a maximum of ~ 0.2 – 0.7 mmol/mol. (4) The presence of varying amounts of secondary, gametogenic calcite (see section 3.2), which is added to the foraminiferal tests at greater water depths and hence records lower temperatures, biases Mg/Ca toward lower ratios. The quantitative effect on Mg/Ca, however, remains uncertain as we have no control on the amounts of gametogenic calcite. (5) Unintended mixing of morphotypes may have led to a widened scatter in Mg/Ca. Foraminiferal species are known to form different morphotypes showing different geochemical signatures [e.g., Williams *et al.*, 1981; Deuser and Ross, 1989]. Steinke *et al.* [2005] presented a statistically significant difference in Mg/Ca for two morphotypes of *G. ruber* white of 0.38 ± 0.30 mmol/mol. In spite of all these factors potentially affecting Mg/Ca, the succession of foraminiferal species and varieties

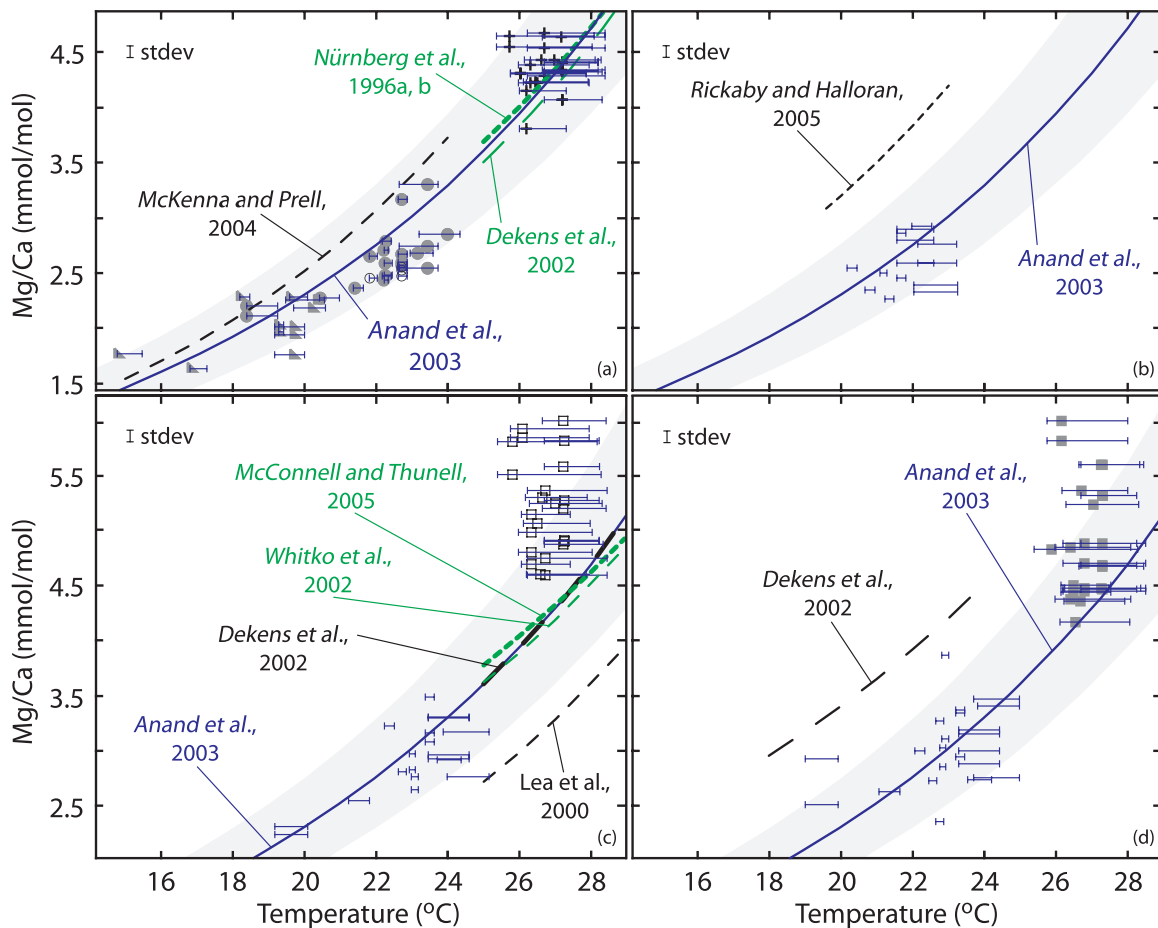


Figure 5. Caribbean core-top Mg/Ca ratios versus temperature data of 9 planktonic foraminiferal species and varieties in comparison with published Mg/Ca-paleotemperature equations. Horizontal bars indicate entire annual temperature ranges [NODC, 2001] at the respective core locations from preferred living depths of the foraminiferal species and varieties. Symbols are placed at April-May temperatures when foraminiferal abundance maxima were observed [Schmuker and Schiebel, 2002]. Shaded areas indicate the Mg/Ca data variability around the Anand et al. [2003] multispecies calibration curve (solid lines: $Mg/Ca = 0.38 \cdot \exp(0.09T)$). Average standard deviations of replicate Mg/Ca analyses (stdev) is indicated by vertical bars. (a) Deep-dwelling *G. crassaformis* (triangles), *G. truncatulinoides* dextral (dots) and sinistral (open circles), and shallow-dwelling *G. sacculifer* (crosses): Habitat depths of ~ 230 m, ~ 160 m, and ~ 40 m, respectively (see section 4.3); species-specific paleotemperature calibration curves for *G. truncatulinoides* dextral from McKenna and Prell [2004] ($Mg/Ca = 0.355 \cdot \exp(0.098T)$), and for *G. sacculifer* from Dekens et al. [2002] ($Mg/Ca = 0.37 \cdot \exp(0.09T)$) and Nürnberg et al. [1996a, 1996b] ($Mg/Ca = 0.472 \cdot 10^{(0.036T)}$). (b) Deep-dwelling *G. tumida*: Habitat depth defined to ~ 180 m; species-specific paleotemperature equation for *G. tumida* from Rickaby and Halloran [2005] ($Mg/Ca = 0.53 \cdot \exp(0.09T)$). (c) Thermocline-dwelling *G. menardii* and shallow-dwelling *G. ruber* white (open squares): Habitat depths defined to ~ 150 m and ~ 34 m, respectively; species-specific paleotemperature equations for *G. ruber* white from Lea et al. [2000] ($Mg/Ca = 0.3 \cdot \exp(0.09T)$), Dekens et al. [2002] ($Mg/Ca = 0.38 \cdot \exp(0.09T)$), Whitko et al. [2002] ($Mg/Ca = 0.57 \cdot \exp(0.074T)$), and McConnell and Thunell [2005] ($Mg/Ca = 0.69 \cdot \exp(0.068T)$). (d) Thermocline-dwelling *N. dutertrei* and shallow-dwelling *G. ruber* pink (squares): Habitat depths defined to ~ 150 m and ~ 37 m, respectively; species-specific paleotemperature equation for *N. dutertrei* from Dekens et al. [2002] ($Mg/Ca = 0.6 \cdot \exp(0.08T)$).

according to their Mg/Ca clearly reflects the expected depth habitats (Figure 3).

[24] As we are not able to establish core-top Mg/Ca versus temperature calibrations due to the restricted

temperature range covered by our Caribbean samples, we compared our (dissolution-unaffected) core-top Mg/Ca data from <2000 m water depths to published Mg/Ca versus temperature relationships. Mg/Ca ratios of the deep-dwelling species

G. truncatulinoides dextral and sinistral, and *G. crassaformis* exhibit a clear relationship to temperature. Each species covers a regional temperature range of $\sim 5\text{--}6^\circ\text{C}$ when applying the habitat specification of *Schmuker and Schiebel* [2002] of ~ 160 m and ~ 230 m water depth, respectively (Figure 5a). The *G. truncatulinoides* dextral and sinistral, and *G. crassaformis* data fall onto published Mg/Ca versus temperature curves of *Anand et al.* [2003] proposed for multispecies planktonic foraminifers, and of *McKenna and Prell* [2004] for *G. truncatulinoides* dextral (Figure 5a). Also, the Mg/Ca data of Caribbean *G. sacculifer* agree with published calibration curves of *Nürnberg et al.* [1996a, 1996b], *Dekens et al.* [2002], and *Anand et al.* [2003] when relating water temperatures from ~ 40 m water depth [*Schmuker and Schiebel*, 2002] to the Mg/Ca ratios (Figure 5a).

[25] The Mg/Ca ratios of *G. tumida* are very close to those of *G. truncatulinoides* (Figure 4), suggesting akin living depths. If we assume a living depth of ~ 180 m, which is in accordance to *Ravelo and Fairbanks* [1992] and *Chaisson and Ravelo* [1997], then the according Mg/Ca versus temperature data fall onto the *Anand et al.* [2003] calibration curve (Figure 5b). Thermocline-dwelling *G. menardii* and *N. dutertrei* have less well-defined habitat specifications. In view of the low Mg/Ca ratios, the extremely shallow abundance maximum of $\sim 40\text{--}50$ m observed by *Schmuker and Schiebel* [2002] is highly unlikely to truly reflect the foraminiferal calcification depth (Figure 4). Assuming an average depth habitat of ~ 150 m water depth inferred from various sources [*Fairbanks et al.*, 1980, 1982; *Fairbanks and Wiebe*, 1980; *Sautter and Thunell*, 1991b; *Ravelo and Fairbanks*, 1992; *Chaisson and Ravelo*, 1997] results in Mg/Ca versus temperature data matching the *Anand et al.* [2003] calibration curve reasonably well (Figures 5c and 5d).

[26] In contrast to the above mentioned foraminiferal species, *G. ruber* is different. First, the spread of Mg/Ca ratios in dissolution-unaaffected core-top samples is largest among our studied species. Second, Mg/Ca of *G. ruber* white and pink deviate from existing Mg/Ca versus temperature calibrations when considering depth habitats of ~ 34 m and ~ 37 m, respectively, which are the April-May abundance maxima in Caribbean plankton nets [*Schmuker and Schiebel*, 2002] (Figures 5c and 5d). Such depth estimates for *G. ruber* seem plausible as MOCNESS plankton tows from the western Gulf of Mexico show the depth preference

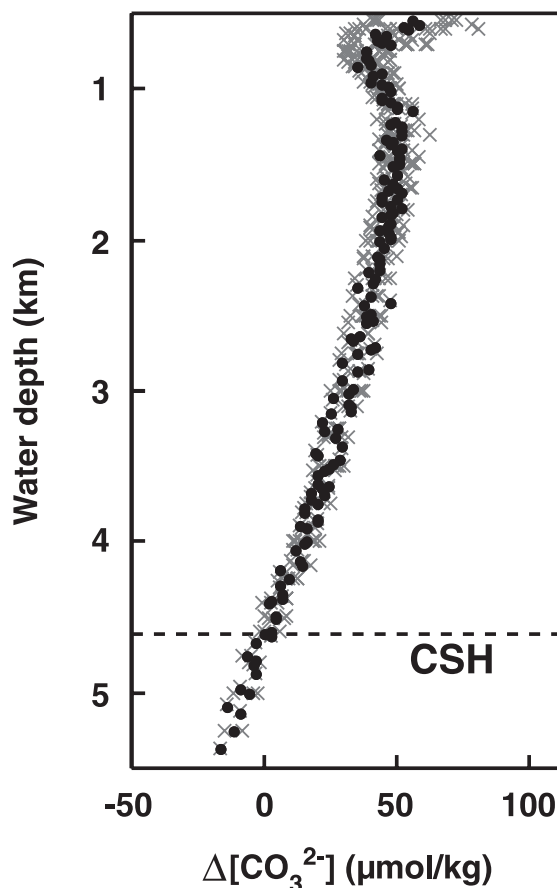


Figure 6. $\Delta[\text{CO}_3^{2-}]$, defined as the difference between the in situ carbonate ion concentration ($[\text{CO}_3^{2-}]$) and $[\text{CO}_3^{2-}]$ at saturation, versus water depth. Total alkalinity (TA) and TCO_2 data necessary to calculate the in situ $[\text{CO}_3^{2-}]$ were obtained from the World Ocean Circulation Experiment (WOCE) Line numbers A22 and A22_2003A (Caribbean stations 1–38 and 2–37, respectively; available at <http://whpo.ucsd.edu/index.htm>; Figure 1). The in situ $[\text{CO}_3^{2-}]$ was calculated using the program of *Lewis and Wallace* [1998] developed for CO_2 System Calculations. $[\text{CO}_3^{2-}]$ at saturation was calculated after *Jansen et al.* [2002]. Gray crosses mark computed $\Delta[\text{CO}_3^{2-}]$ data; black dots indicate $\Delta[\text{CO}_3^{2-}]$ data averaged for every 20 m water depth. CSH is calcite saturation horizon (dashed line) as the approximate top of the lysocline.

for *G. ruber* white to be 0–45 m, for *G. ruber* pink to be slightly deeper (25–65 m) [*Tedesco and Thunell*, 2003]. Even if one would assume a very shallow habitat of <10 m being $\sim 0.2^\circ\text{C}$ warmer than at ~ 30 m, the Mg/Ca data would still deviate from available calibration curves. The reason for that is not yet clear. It is evident, however, that *G. ruber* definitely occupies a shallower habitat than *G. sacculifer* as suggested by higher Mg/Ca (Figure 4) and generally lighter $\delta^{18}\text{O}$ values

Table 4. Species-Specific Parameters for Equations (2)–(7)^a

Species	d Correction, Equation (6)				Δ Correction, Equation (7)			
	a	b	r ²	d _{critical}	a	b	r ²	Δ _{critical}
<i>G. crassaformis</i>	6,833	1,962	0.61	2,838.4	−20.00	22.70	0.86	26.31
<i>G. menardii</i>	7,734	1,413	0.8	2,938.8	−27.61	15.03	0.78	22.88
<i>G. ruber p.</i>	9,403	1,381	0.79	2,631.1	−44.08	14.03	0.84	24.67
<i>G. ruber w.</i>	9,793	1,382	0.76	2,535.3	−40.93	12.61	0.71	25.28
<i>G. sacculifer</i>	12,030	2,088	0.75	2,967.8	−68.50	20.87	0.78	22.10
<i>G. truncatulinoides d.</i>	6,456	1,303	0.56	3,018.8	−16.23	13.79	0.74	20.17
<i>G. truncatulinoides s.</i>	7,471	1,888	0.65	2,746.4	−27.43	21.28	0.94	25.77
<i>G. tumida</i>	6,985	1,547	0.55	2,988.5	−6.62	9.43	0.32	17.70
<i>N. dutertrei</i>	7,057	1,484	0.67	2,567.9	−12.73	11.65	0.48	22.58

^a Here, *a* are y axis intercepts; *b* is slope of the regression lines; *r*² are correlation coefficients; *d*_{critical} (m) are species-specific water depths where Mg²⁺ removal starts; Δ_{critical} (μmol/kg) are species-specific Δ[CO₃^{2−}] levels where Mg²⁺ removal starts.

[Anand *et al.*, 2003]. Their oxygen isotope signal commonly points to summer sea-surface conditions [Deuser, 1987], and both Flower *et al.* [2004] and Anand *et al.* [2003] pointed out that Mg/Ca of *G. ruber* white commonly represents warmer than average temperatures in the Orca Basin (Gulf of Mexico) and in the Sargasso Sea, respectively.

4.4. Correcting Deep Water Mg/Ca Ratios for the Effect of Dissolution

[27] Dependent on the foraminiferal species and varieties, a major and systematic decrease in Mg/Ca ratios starts below species-specific critical water depths, which are situated in the depth interval between 2000–3000 m. This is far above the calcite saturation horizon (CSH) at ~4600 m (Figure 6), which approximates the present-day top of the lysocline at a Δ[CO₃^{2−}] (difference between the in situ carbonate ion concentration ([CO₃^{2−}]) and the [CO₃^{2−}] at saturation) of 0 μmol/kg, and is equal to a calcite solubility ratio (Ω) of 1 [Lewis and Wallace, 1998; Tyrrell and Zeebe, 2004]. In fact, below >3000 m, Mg/Ca decreases linearly with increasing water depth (*r*² = 0.55–0.80; Figure 2; Table 4). This close relationship is also reflected in the linear decrease of Mg/Ca with decreasing Δ[CO₃^{2−}] (*r*² = 0.32–0.94; Figure 5; Table 4) below 20 μmol/kg (Figure 7; Table 4).

[28] As only few Mg/Ca ratios cover the interval between 2000–3000 m (40–20 μmol/kg Δ[CO₃^{2−}]), the onset of dissolution was determined by calculating the intersection between the unaffected Mg/Ca means from <2000 m (>40 μmol/kg) and the regressions through the dissolution-affected ratios from >3000 m (<20 μmol/kg). The decrease of Mg/Ca can be described as

$$\Delta \text{Mg/Ca} = \text{Mg/Ca}_{\text{initial}} - \text{Mg/Ca}_{\text{measured}} \quad (1)$$

where ΔMg/Ca is the difference between the (unknown) initial, dissolution-unaffected Mg/Ca (Mg/Ca_{initial}) and the measured Mg/Ca (Mg/Ca_{measured}). A reasonable approximation for Mg/Ca_{initial} is the mean Mg/Ca (Mg/Ca_{mean}) from <2000 m (Table 2). The intersection of the Mg/Ca_{mean} with the regression line calculated for Mg/Ca from >3000 m water depth and <20 μmol/kg Δ[CO₃^{2−}], respectively, provides the critical water depth (*d*_{critical} in m) and the critical calcite saturation state (Δ_{critical} in μmol/kg), where Mg²⁺-removal due to dissolution starts:

$$d_{\text{critical}} = a - b \cdot \text{Mg/Ca}_{\text{mean}} \quad (2)$$

$$\Delta_{\text{critical}} = a - b \cdot \text{Mg/Ca}_{\text{mean}} \quad (3)$$

where *a* is the y axis intercept, and *b* is the slope of the regression lines (Figures 2 and 7; Table 4). The levels of beginning preferential dissolution of Mg²⁺ appear to be species-specific. Δ_{critical} spans values of ~18–26 μmol/kg (Table 4) which correspond to Ω values of ~1.5, accompanied by *d*_{critical} of ~2500–3000 m (Table 4).

[29] Different susceptibility of calcitic tests from different planktonic foraminiferal species to dissolution is a well-known feature [e.g., Berger, 1968, 1970, 1971], likewise is the heterogeneous distribution of Mg²⁺ in foraminiferal tests [Bender *et al.*, 1975; Duckworth, 1977; Nürnberg, 1995; Brown and Elderfield, 1996; Nürnberg *et al.*, 1996a; Rosenthal *et al.*, 2000; Eggins *et al.*, 2003; McKenna and Prell, 2004; Anand and Elderfield, 2005; Bentov and Erez, 2005, 2006]. Sadekov *et al.* [2005] recently illustrated by electron microprobe mapping that low-Mg/Ca (~1–5 mmol/mol) and high-Mg/Ca bands (~8–11 mmol/mol) alternate within tests of planktonic foraminifera. As Mg²⁺

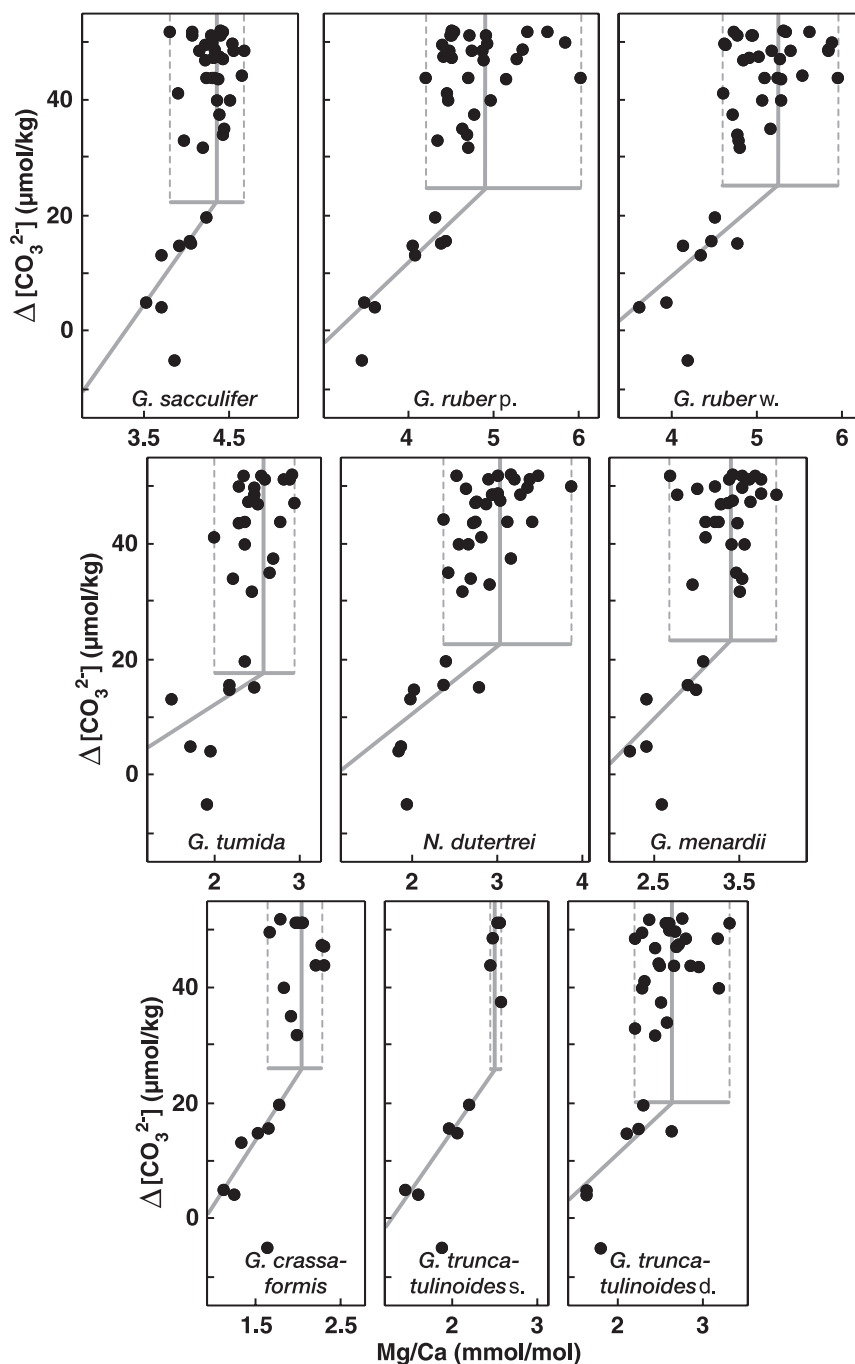


Figure 7. Mg/Ca ratios versus $\Delta[\text{CO}_3^{2-}]$ for each planktonic foraminiferal species and variety indicating significant Mg^{2+} loss below species-specific $\Delta[\text{CO}_3^{2-}]$ levels (Δ_{critical} = horizontal lines). Δ_{critical} is defined by the intercept between the vertical lines (species-specific mean Mg/Ca calculated from samples <2000 m water depth, similar to $\Delta[\text{CO}_3^{2-}] > 40 \mu\text{mol/kg}$ (Table 2)) and diagonal lines (regression lines from Mg/Ca at $\Delta[\text{CO}_3^{2-}] < 20 \mu\text{mol/kg}$). Deepest core-top sample M35024–6 from below the CSH was defined as an outlier and excluded from the calculation of the regressions.

has a pronounced effect on the stability of biogenic calcite [Walter and Morse, 1985], and as ion activity products increase with increasing Mg^{2+} , Brown and Elderfield [1996] deduced that Mg-enriched phases are less dissolution resistant. The

selective removal of Mg^{2+} -enriched foraminiferal parts with increasing dissolution should thus cause the reduction of the intraspecific Mg/Ca scatter, which is observed in samples from >3000 m water depth in comparison to shallow samples from

<2000 m. This reduction amounts to 36% for *G. ruber* white, 50% for *G. ruber* pink, 44% for *G. sacculifer*, 46% for *G. menardii*, 27% for *N. dutertrei*, 14% for *G. tumida*, 36% for *G. truncatulinoides* dextral, while *G. crassaformis* shows no discernable change.

[30] Nonetheless, the slopes b of the regression lines indicate a relatively uniform $\sim 0.5\text{--}0.8$ mmol/mol change in Mg/Ca ratios per kilometer water depth and $\sim 0.04\text{--}0.11$ mmol/mol per $1\ \mu\text{mol/kg}$ decrease in $\Delta[\text{CO}_3^{2-}]$ with respect to $\text{Mg/Ca}_{\text{mean}}$ for all studied foraminiferal species and varieties (Table 4). In contrast to observations of Savin and Douglas [1973], Lorens et al. [1977], Brown and Elderfield [1996], and Rosenthal et al. [2000], the relatively similar decline of Mg/Ca in combination with broadly the same d_{critical} and Δ_{critical} (Figures 2 and 7; Table 4) for all foraminifers studied does not support the notion that calcite of shallow-dwelling foraminifers with higher Mg/Ca dissolves preferentially with respect to deep-dwelling species bearing lower Mg/Ca.

[31] The absolute change in Mg/Ca ratios of $\sim 0.5\text{--}0.8$ mmol/mol per kilometer water depth is equivalent to a relative decrease of $\sim 11\text{--}29\%$ per kilometer. Our results support previous core-top studies showing a decrease of Mg/Ca with increasing water depth [Savin and Douglas, 1973; Bender et al., 1975; Rosenthal and Boyle, 1993]. For *G. sacculifer*, Lorens et al. [1977] showed a decrease in Mg/Ca of $\sim 40\%$ in tropical East Pacific Rise and central Pacific samples over a depth range from $\sim 600\text{--}4000$ m. Both Lorens et al. [1977] and Brown and Elderfield [1996] detected a decrease of 40% in Mg/Ca of *G. tumida* over a depth range from $\sim 1900\text{--}4700$ m, while the decline in Mg/Ca of *N. dutertrei* amounts to $\sim 59\%$ from $\sim 600\text{--}3800$ m, and $\sim 25\%$ for *G. ruber* from $\sim 1900\text{--}4700$ m [Lorens et al., 1977], all from tropical Pacific samples. Assuming a linear Mg/Ca descent with depth, these numbers would convert to Mg/Ca reductions of $\sim 12\%$ for *G. sacculifer*, $\sim 14\%$ for *G. tumida*, $\sim 18\%$ for *N. dutertrei*, and $\sim 9\%$ for *G. ruber* per kilometer water depth. At Ontong Java Plateau, Lea et al. [2000] showed a decrease in Mg/Ca in tests of *G. ruber* by $\sim 12\%$ per kilometer water depth, while Dekens et al. [2002] revealed a decrease of $\sim 14\%$ per kilometer. At Sierra Leone Rise and Ceara Rise, a decrease in Mg/Ca of $\sim 7\%$ and $\sim 5\%$ per kilometer, respectively, was observed for *G. ruber* [Dekens et al., 2002]. For *N. dutertrei*, Dekens et al. [2002] found a Mg/Ca decline of up to $\sim 23\%$ per kilometer.

[32] Our results on decreasing Mg/Ca ratios below species-specific $\Delta[\text{CO}_3^{2-}]$ levels of $\sim 18\text{--}26\ \mu\text{mol/kg}$ are in general accordance with Dekens et al. [2002], who also noted the onset of Mg^{2+} loss below $\sim 20\ \mu\text{mol/kg}$ $\Delta[\text{CO}_3^{2-}]$ at Ceara Rise (Atlantic) and Ontong Java Plateau (Pacific). In the South China Sea, Whitko et al. [2002] found declining Mg/Ca in core-top *G. ruber* below ~ 2000 m water depth, implying a strong dissolution effect below ~ 2000 m. This is considerably shallower than both, the present-day lysocline and the carbonate compensation depth in this ocean area at ~ 3000 m and ~ 3800 m, respectively [Miao et al., 1994; Feely et al., 2002, 2004]. Although we do not have $\Delta[\text{CO}_3^{2-}]$ values for this ocean basin at hand, we speculate that the onset of Mg/Ca change observed by Whitko et al. [2002] might coincide with the approximate threshold in $\Delta[\text{CO}_3^{2-}]$ of $\sim 20\ \mu\text{mol/kg}$. It needs to be proven, though, whether this threshold is globally valid.

[33] In order to correct the measured Mg/Ca ratios for the selective loss of Mg^{2+} due to dissolution, $\Delta\text{Mg/Ca}$ was calculated for every measured Mg/Ca from below d_{critical} and Δ_{critical} :

$$\Delta\text{Mg/Ca} = (d - d_{\text{critical}})/b \quad (4)$$

$$\Delta\text{Mg/Ca} = (\Delta_{\text{critical}} - \Delta)/b \quad (5)$$

where d is the water depth of the core-top sample in meter (with $d > d_{\text{critical}}$) (Tables 1 and 2), and Δ is the averaged $\Delta[\text{CO}_3^{2-}]$ in $\mu\text{mol/kg}$ at the seafloor (with $\Delta < \Delta_{\text{critical}}$) (Figure 6). We then combined equations (1) and (4), as well as (1) and (5) to recalculate the dissolution-unaffected $\text{Mg/Ca}_{\text{initial}}$ (water depth correction (d -corrected) according to equation (4) (Figure 8); and Δ_{critical} correction (Δ -corrected) according to equation (5)):

$$\text{Mg/Ca}_{\text{initial}}(d - \text{corrected}) = \text{Mg/Ca}_{\text{measured}} + (d - d_{\text{critical}})/b \quad (6)$$

$$\text{Mg/Ca}_{\text{initial}}(\Delta - \text{corrected}) = \text{Mg/Ca}_{\text{measured}} + (\Delta_{\text{critical}} - \Delta)/b \quad (7)$$

[34] Both independently developed routines for the correction of dissolution-induced Mg/Ca decline produce comparable results indicating that selective Mg^{2+} removal is mainly driven by the calcite saturation state (Figure 9). Application of the correction routines reduces the overall intraspecific variability of Mg/Ca by $\sim 24\text{--}64\%$ (Figure 8).

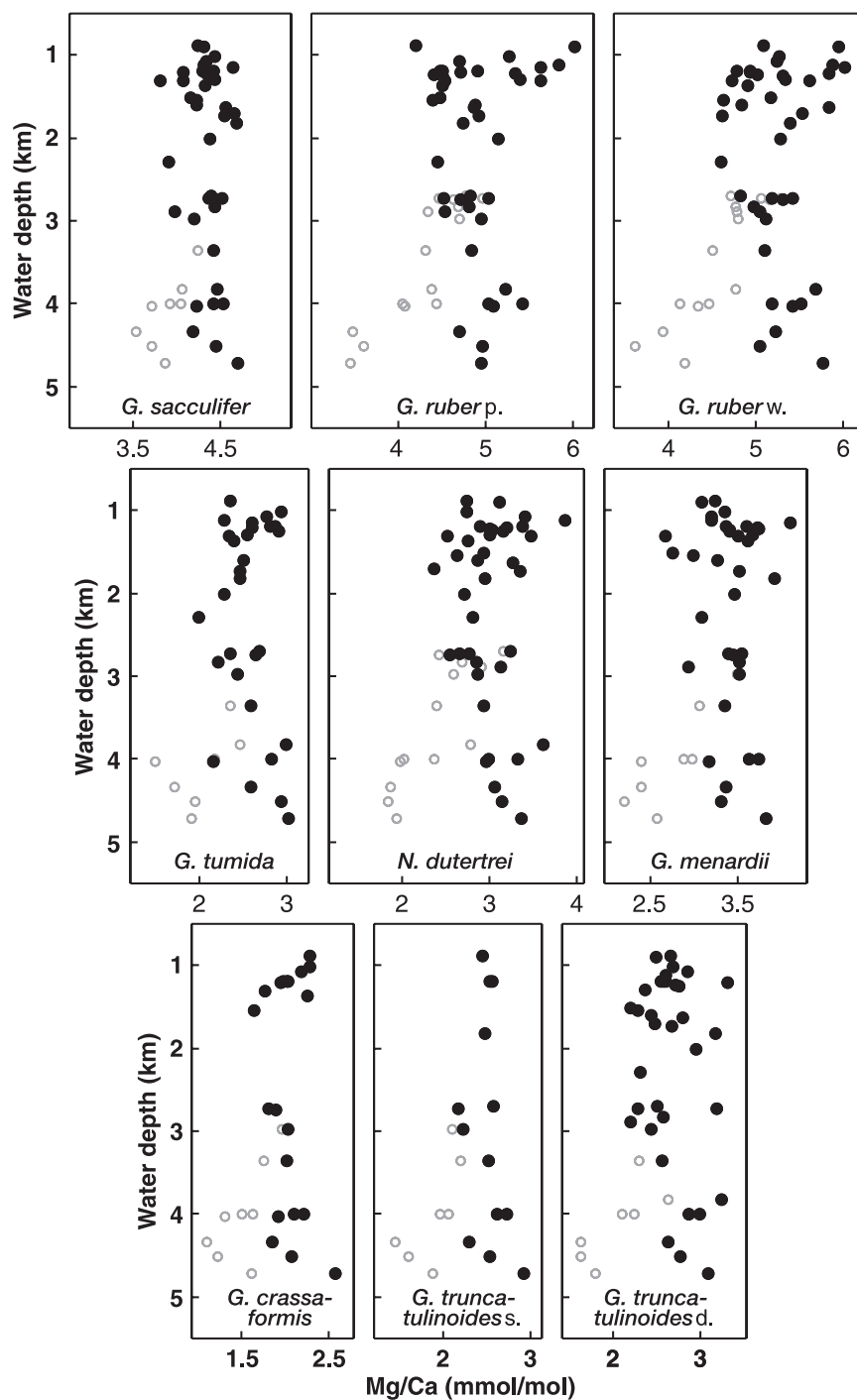


Figure 8. Foraminiferal Mg/Ca ratios versus water depth (black dots): Mg/Ca from below $d_{critical}$ were corrected for water-depth induced calcite dissolution according to equation (6). For comparison, the measured, noncorrected Mg/Ca data are included (circles). The application of the correction routine leads to a reduction of the intraspecific Mg/Ca variability by 24–64%.

4.5. Implications for Paleorecords

[35] We chose the Mg/Ca records of *G. ruber* white of two Caribbean cores [Schmidt *et al.*, 2004] to test the downcore applicability of the proposed

correction routine. Today, both records chart similar temperature conditions within the Columbian Basin (Figure 1) and hence are expected to show similar Mg/Ca ratios. The impact of dissolution on the Holocene (3–6 kyrs) mean Mg/Ca from shal-

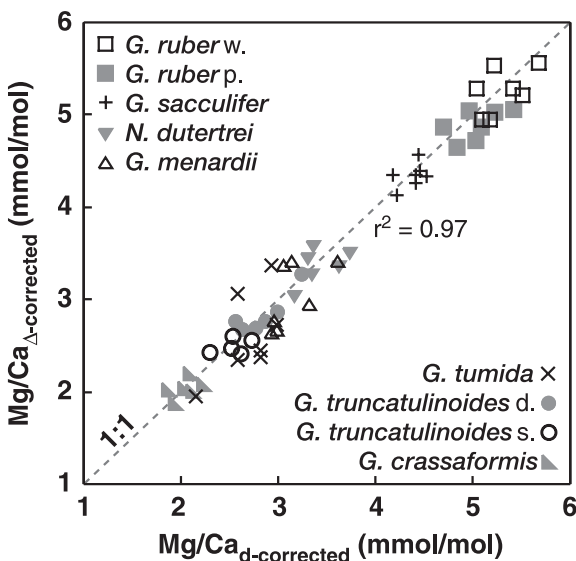


Figure 9. Depth-corrected (d-corrected, equation (6)) versus $\Delta[\text{CO}_3^{2-}]$ -corrected (Δ -corrected, equation (7)) Mg/Ca ratios: The independently corrected ratios from below d_{critical} and Δ_{critical} highly correlate ($r^2 = 0.97$). Related high probability values ($p = 0.77$) imply that the corrected Mg/Ca means of all species and varieties do not statistically deviate from each other, implying that the selective removal of Mg^{2+} from foraminiferal calcite is mainly a function of the calcite saturation state $\Delta[\text{CO}_3^{2-}]$.

lower ODP Site 999A (2827 m, 4.13 mmol/mol) appears to be less than on Mg/Ca from deeper site VM22–128 (3623 m, 3.75 mmol/mol), differing by 0.38 mmol/mol. Depth correction (equation (6); parameters see Table 4) leads to higher Holocene Mg/Ca of 4.34 mmol/mol at ODP Site 999A and 4.53 mmol/mol at site VM28–122, reducing the offset to 0.19 mmol/mol. Such inter-site offset converts to a $\sim 0.5^\circ\text{C}$ difference (calculated after *Anand et al.* [2003]; see Table 2), which is in accordance with the modern temperature pattern in the Columbian Basin.

[36] For glacial times, instead, reduced $p\text{CO}_2$ and increased $[\text{CO}_3^{2-}]$ in Caribbean intermediate and deep waters [*Barker and Elderfield, 2002; Broecker and Clark, 2002*] suggest a better carbonate preservation [*Haddad and Droxler, 1996; Anderson and Archer, 2002*]. Hence Mg/Ca ratios should have been unaffected by dissolution effects even at greater water depths. This is reflected in the similar last glacial (19–21 kyrs) mean Mg/Ca of Caribbean cores ODP 999A and

VM28–122 [*Schmidt et al., 2004*], both revealing 3.36 mmol/mol.

5. Conclusions

[37] We analyzed Mg/Ca ratios in tests of seven planktonic foraminiferal species and four of their varieties from 42 Caribbean and tropical Atlantic core-top samples covering water depths of ~ 900 –4700 m in order to quantify the effect of dissolution on Mg/Ca. As lateral temperature gradients at similar water depths in the study area are minor, the well-known temperature effect on intraspecific Mg/Ca variations is considered to be minimal. Shallow core-top samples from above 2000 m water depth being unaffected by dissolution processes, clearly reveal interspecific differences in Mg/Ca with high ratios in shallow-dwelling, and low ratios in deep-dwelling species. This pattern reflects the expected habitat depths, and clearly points to different calcification depths at different temperature regimes.

[38] The core-top samples exhibit a linear decline of foraminiferal Mg/Ca ratios below water depths of ~ 2500 –3000 m (d_{critical}) depending on the foraminiferal species or variety. Hence the onset of selective Mg^{2+} removal (d_{critical}) takes place far above the present-day lysocline and concurs with calcite saturation state $\Delta[\text{CO}_3^{2-}]$ levels of ~ 18 –26 $\mu\text{mol/kg}$ (Δ_{critical}). Mg/Ca from above these species-specific critical levels appear to remain stable and hence are considered to be unaffected by dissolution. Nonetheless, the intraspecific variability at shallow depths is larger than at greater water depths.

[39] Below the species-specific levels (Δ_{critical} and d_{critical}) of apparent Mg^{2+} removal, Mg/Ca ratios decline linearly by ~ 0.04 – 0.11 mmol/mol per 1 mol/kg decrease in $\Delta[\text{CO}_3^{2-}]$ and ~ 0.5 – 0.8 mmol/mol per kilometer water depth. The relatively similar rates of Mg/Ca change, and the broadly similar d_{critical} and Δ_{critical} for all species and varieties studied, imply that low magnesium calcite of shallow-dwelling foraminifers showing higher Mg/Ca does not dissolve preferentially with respect to calcite of deep-dwelling species bearing lower Mg/Ca.

[40] We developed independent routines to correct core-top Mg/Ca ratios from below Δ_{critical} and d_{critical} for the effect of dissolution. Both routines produce comparable results, implying that the selective Mg^{2+} removal is mainly driven by the calcite saturation state at the seafloor. They may be used as robust approaches for the assessment and

correction of the dissolution effect on planktonic Mg/Ca. The water-depth correction of Mg/Ca (equation (6)), however, should only be applied to samples from ocean areas with resembling calcite saturation states. Instead, the $\Delta[\text{CO}_3^{2-}]$ correction (equation (7)) is applicable to any ocean area as long as the $\Delta[\text{CO}_3^{2-}]$ levels are known. The critical $\Delta[\text{CO}_3^{2-}]$ level of $\sim 20 \mu\text{mol/kg}$ as an effective threshold for the onset of Mg^{2+} removal from low magnesium foraminiferal calcite may be globally valid in this respect, even through geological time spans. The correction routines proposed here may help to improve core-top Mg/Ca versus temperature calibrations, and to recalculate initial Holocene Mg/Ca, from which paleotemperature estimates can be derived.

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